1 Fully Solution Processed n-i-p-*like* Perovskite Solar Cells with Planar Junction: How the

2 Charge Extracting Layer Determines the Open Circuit Voltage

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- 47 Keywords: perovskite solar cells, interface engineering, planar junctions, solution process,
- 48 open-circuit voltage
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50 This is the peer reviewed version of the following article: Fully Solution Processed n-i-p-like 51 Perovskite Solar Cells with Planar Junction: How the Charge Extracting Layer Determines 52 Open Circuit Voltage, which has been published in final form the at 53 10.1002/adma.201604493. This article may be used for non-commercial purposes in 54 accordance with Wiley Terms and Conditions for Use of Self-Archived Versions

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Solution-processable hybrid perovskite semiconductors have risen to the forefront of 58 59 photovoltaics research, offering the potential to combine low-cost fabrication with high power conversion efficiency.^[1] The first architectures made use of nanocrystalline TiO₂ deposited on 60 61 the transparent electrode, both in the form of mesoporous and compact films, as selective contact for electrons alike the so-called Grätzel solar cells.^[2-4] Such a structure is generally 62 63 indicated in the field as *direct architecture* as electrons are collected at the front transparent 64 electrode and holes at the metal contact (often gold). Further development has been driven by 65 empirical optimization strategies and testing of a variety of architectures. One of these 66 promising architectures uses fully organic selective contacts. This choice brings, first of all, the advantage of low temperature processing; additionally, a beneficial effect in terms of 67 68 electrical stabilization of the device with organic electron extraction layers in a flat junction configuration, with respect to the use of TiO₂, has been repeatedly observed.^[5-7] So far, 69 architectures with fully organic selective contacts are either limited to the so-called inverted 70 71 architecture, as a consequence of difficulties in the solution processing of multi-layers, or they have to rely on vacuum deposition methods.^[6-8] Solution-processed, flat junctions, multi-72 73 layered direct architectures, which would open the path towards large-area, low-temperature processing of high-efficiency direct cells, have been poorly explored, and their power 74 conversion efficiencies are still limited.^[9-11] Intuitively, the development and optimization of 75 76 different architectures need a full engineering of different interfaces. This is especially critical 77 for metal halide perovskites that, unlike more widely studied inorganic semiconductors, have 78 fluctuating ionic structures where tilting, distortions and polarizability of the lattice strongly 79 affect their optoelectronic properties. This makes the reliability of perovskite-based devices

strongly dependent on the control of the microstructure of the active material and of its
response to external stimuli, such as chemical interactions upon interface formation,^[12]
electric field,^[13,14] light^[15] and environmental agents.^[16,17]

It is also worth noting that, while perovskite solar cells have made spectacular progress in terms of solar-to-electrical current conversion efficiency in very short time, such advances are mainly based on an empirical approach, which has led to a myriad of architectures, whose advantages and limitations cannot be attributed yet to a fundamental property of the active material or to the design and processing of the device.

88 Herein we present the fabrication of a fully solution processed direct perovskite solar cell with 89 planar junction, reaching power conversion efficiencies (PCEs) close to 19% and an open 90 circuit voltage (V_{oc}) larger than 1.1 V. Such a result is achieved by using as a selective contact 91 for electrons a fullerene derivative, the [6,6]-phenyl-C61-butyric styryl dendron ester 92 (PCBSD), functionalized with a dendron containing two styryl groups as thermal cross-linkers. 93 In situ crosslinking of PCBSD by thermal treatment forms a robust, adhesive, and solvent-94 resistant thin electron accepting layer, on top of which a high quality perovskite film 95 processed from solution can be grown. We show that the choice of the electron extracting layer (EEL) is of utmost importance to maximize the solar cell V_{oc} as interface engineering 96 97 accounts not only for the energy level alignment between the EELs and the perovskite, but 98 also for the quality of the microstructure of the perovskite bulk film that is driven by the 99 substrate surface. We further demonstrate that in this device architecture, at V_{oc} conditions the 100 high carrier density is sustained not only by the charge extracting contacts but also by the 101 semiconducting active layer, thus rationalizing the optimal opto-electronic performance of the 102 proposed devices as a beneficial effect of the engineering of the solution processed 103 EEL/perovskite interface in terms of micro-structure and reduced carrier recombination/ 104 charge extraction improvement.

105 The PCBSD is prepared by optimizing the synthesis route described by Hsieh et al. in order to achieve a final yield of 81% (see Supporting Information (SI) for details).^[18] The crosslinking 106 107 of PCBSD is a thermal-induced radical polymerization between the terminal vinyl bonds of PCBSD (see Figure 1a).^[19,20] In order to optimize the timing of the crosslinking process at 108 109 160 °C (see Figure S16 of SI for the temperature optimization), its progress is monitored with 110 in-situ infrared (IR) spectroscopy for 45 min. Figure 1b presents the initial and the final spectra. The bands at 987 cm⁻¹ and 908 cm⁻¹, which are assigned to the CH and CH₂ wagging 111 112 of the vinyl groups in *trans* conformation (see Figure S11 in SI for assignment through DFT 113 calculations), show an absolute reduction in intensity in time during the thermal crosslinking. Their lowering is due to the conversion of the sp^2 carbon atoms of the vinyl groups into sp^3 114 carbon atoms, which occurs during the PCBSD crosslinking.^[21] The band at 908 cm⁻¹ is 115 116 selected for quantitative analysis and monitoring of the crosslinking time. The peak area of the band at 908 cm⁻¹ is determined every 5 min up to a final reaction time of 45 min. The 117 118 values, normalized over the peak area at t = 0, show that the reaction follows a 1st order 119 exponential decay and 35% of the carbon atoms of the vinyl groups are converted after a 45-120 min thermal anneal (Figure 1c). To prove that the achieved crosslinking degree is sufficient 121 enough for the crosslinked PCBSD (C-PCBSD) layer to be solvent resistant, UV-visible 122 spectra of the substrates are measured before and after being rinsed with dimethyl 123 formamidine (DMF). The as-cast PCBSD films are almost completely washed-off (Figure 1d), 124 whereas only a weak decrease in the intensity of the bands is shown after crosslinking induced 125 with a 30-min thermal anneal at 160 °C (Figure 1e).

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Before incorporating such EELs in a device structure, we tested its electronic properties. One of the main challenges to face upon crosslinking procedure of an organic semiconductor is a possible degradation of its transport properties.^[22] C-PCBSD has been tested in a field-effect transistor (FET) geometry, which is quite demanding device in terms of electronic properties,

131 as it requires in-plane charge, in this case electrons, transport along tens of microns. Electron percolation pathways can be evidently formed and a neat field-effect modulation of the 132 133 current can be observed, already in a non-optimized device, exhibiting an electron mobility in the order of 10^{-5} cm² V⁻¹ s⁻¹ for surface charge densities in the range 8 x 10^{15} to 3 x 10^{16} m⁻² 134 $(1.3 \times 10^{-7} \text{ to } 5.1 \times 10^{-7} \text{ C cm}^{-2})$ (Figure S1 in SI). This result shows that our C-PCBSD has 135 136 almost an order of magnitude higher mobility than what previously reported in undoped C-PCBSD based FET devices.^[9] Thus we can safely confirm that the C-PCBSD film can sustain 137 138 in-plane electron transport over several microns, and it is therefore likely to show good 139 transversal electron transport properties over tens of nanometres of its thickness thus being a 140 potential candidate as a thin EEL.

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142 To start the solar cell fabrication, a very thin layer of C-PCBSD (less than 10-20nm, see 143 Figure S14 in SI) is deposited onto the FTO/TiO₂ substrates as thicker films eventually lead 144 to an increase of the device series resistance (see Figure S15 in SI). Its surface work function 145 is measured in air by Kelvin probe and it results more electronegative with respect to that of 146 TiO₂ and comparable to the [6,6]-phenyl- C_{61} -butyric acid methyl ester (60-PCBM) (see Table 147 S1 in SI). Subsequently a methylammonium lead iodide (CH₃NH₃PbI₃) thin film is directly 148 synthesised from solution on top of FTO/TiO₂/C-PCBSD via a two-step procedure (see 149 methods in SI for details). The final thickness of CH₃NH₃PbI₃ depends on the thickness of the 150 PbI₂ layer (see SI for the layer optimization). Figure 2a shows X-ray diffraction spectra 151 measured before and after the CH₃NH₂I spin cast on top of PbI₂ followed by thermal annealing on top of TiO₂ and TiO₂/C-PCBSD substrates. In both cases we do not observe any 152 153 trace of the crystal precursors and the sharp peaks suggest the formation of large perovskite 154 grains. Figure 2b, 2c, 2d, and 2e show the top-view images of the perovskite films grown on 155 TiO₂ and TiO₂/C-PCBSD, respectively, at different magnifications. The perovskite films 156 show a good coverage on both substrates; however, the film grown on TiO₂/C-PCBSD

157 presents much larger grains (> 1 μ m) and a lower roughness. The static contact angle of water 158 on C-PCBSD is more than twice larger with respect to the case of TiO₂ (shown in Figure S2), 159 indicating a more hydrophobic surface. Thus we conclude that the different morphology is the 160 result of a lower wettability of the C-PCBSD surface to dimethyl formamidine/PbI₂ solution 161 with respect to the TiO₂ surface, which drives the formation of crystallite with larger aspect ratio.^[6] To complete the devices, we deposited spiro-MeOTAD by spin coating as hole-162 163 extracting layer (HEL) followed by gold via thermal evaporation. In Figure 3a and 3b we 164 show a cross section of the optimized planar devices employing TiO₂ and TiO₂/C-PCBSD as 165 EELs. Please note that such layers are extremely thin, of the order a few tens of nanometres, 166 nearly conformal to the rough FTO substrate, while the perovskite layers have a different 167 optimized thicknesses, approximately 400 nm and 320 nm respectively. Figure 3c and 3d 168 show the current density versus voltage (J-V) characteristics measured in air under air mass 169 1.5 global (AM 1.5G) conditions. To check for the possible hysteresis phenomena, which are 170 known to strongly influence the device testing, especially for the planar junctions reported so far in the literatures^[12,23,24] we show the J-V curves as a function of scan rate, in forward and 171 172 reverse scan directions. Table 1 summarizes the main figures of merit of the tested samples, 173 i.e., short-circuit current density (J_{sc}) , V_{oc} , fill factor (FF) and PCE. As expected, the device 174 based on bare TiO₂ as EEL has a characteristic that depends on the polarization record of the 175 device (Figure 3c), and the photocurrent, upon polarization, has a transient time of a few 176 seconds (Figure 3e). The figures of merit of such a device correspondingly depend on the 177 testing conditions and are reported in Table 1. On the other hand, the TiO₂/C-PCBSD 178 interface electrically stabilizes the device. Not only the J-V characteristics have very limited dependence on scan directions and rates (Figure 3d), but also the photocurrent has no 179 180 transient time upon polarization (Figure 3f). Such a stable device delivers a PCE as high as 18.7%, with a J_{sc} of approximately 21 mA cm⁻², V_{oc} over 1.1 V and FF close to 80%. The 181

reliability of such numbers is confirmed by a statistical study made over 61 devices, whichcan be found in Figure S3 and S4 in SI.

184 It is worthy to notice that flat junction solar cells, especially those with direct architecture, hardly achieve a V_{oc} of 1 V.^[3,25,26] However here we show a V_{oc} that is consistently higher 185 186 than 1.1 V (see statistics in Figure S3). In this regard, these devices even outperform those 187 made with 60-PCBM (Figure S5 in SI), despite the two EELs presenting very similar energy 188 levels, which will likely align in the same way once interfaced with the perovskite 189 semiconductor given the same nature of the material. Recently, it has been shown that the structural order of the 60-PCBM layer has a significant impact on the enhancement in V_{oc} due 190 191 to a concomitant reduction of the energy disorder, as deduced from the significant decrease in the DOS of the organic semiconductor.^[27] However, here, while the 60-PCBM clearly shows 192 193 a sharp X-ray diffraction peak at 20.5° as a signature of order and crystallinity, there is hardly 194 any diffraction peak in C-PCBSD (see Figure S6 in SI), thus excluding any direct role of the 195 electron extracting layer morphology. Note that, for a fair comparison, we grew the perovskite 196 film by thermally evaporating PbI₂ followed by MAI deposition by spin coating from 2-197 propanol, a procedure that some of us have recently developed^[7] to preserve the soluble 60-198 PCBM layer and we noticed, also in this case, a tendency of forming larger crystallites over 199 the C-PCBSD (see Figure S7 in SI). Overall, beyond the interfacial energetics, the 200 recombination dynamics will contribute to the definition of V_{oc} of solar cells. Since the 201 perovskite film is grown on top of the EEL, the latter's choice will determine its micro-202 structure, which may influence the degree of defect states and the carrier recombination losses within the perovskite film.^[28] In the following we will try to disentangle all these effects by 203 204 combining excitation density dependent steady state photocurrent measurements and photo-205 induced transient techniques.

207 To understand the effect of the fabrication route and the substrate on the quality of the thin 208 perovskite film, we performed photo-current measurements within a photo-conductor device configuration.^[29] Briefly, planar symmetric gold contacts separated by 1 mm are evaporated 209 210 on top of the film and continuous wavelength (cw)-laser induced carriers are extracted by 211 applying a bias voltage of 10 V across them (Figure 4a). Note that the electric fields within 212 this configuration are smaller by a factor of 100 in comparison to the usual electric fields 213 within the solar cell architecture. In Figure 4b we show the photocurrent plotted as a function 214 of the photo-excitation density. All measured films are approximately 300 nm thick and 215 grown on a thin layer of TiO₂, TiO₂/60-PCBM and TiO₂/C-PCBSD on glass; this guarantees 216 to have a thin film micro-structure comparable to the one present in the device while 217 observing mainly bulk processes upon top layer illumination. Since perovskite cannot be 218 solution processed on top of TiO₂/60-PCBM we also compare two thin films grown on C-219 PCBSD, one solution processed and the other grown through thermal evaporation.

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While all the films show a monotonic growth of photocurrent with the excitation density, they 221 222 exhibit distinct intensity trends. As a guide for the eyes, we draw hypothetical linear ($\sim I$) and sublinear ($\sim \sqrt{I}$) intensity dependent photocurrent in dotted lines. The photocurrent trends in 223 224 all the films follow sublinear behaviour at low excitation densities and reach a linear trend at 225 higher densities. Such a behaviour can be explained within a trap-limited Shockley-Read-Hall 226 (SRH) like recombination scenario, where at low-excitation densities, electrons (holes) are 227 most likely trapped within defects and thus do not contribute to the photocurrent. In such case 228 the photocurrent has a sole contribution from hole (electrons) population, which can be shown to scale as \sqrt{I} under steady state conditions,^[35] subsequently giving rise to the sublinear 229 230 behaviour of the photocurrent. As the excitation density is increased, there is a substantial 231 contribution from the untrapped electron (hole) current. Nevertheless, given that the effective defect density is around 10¹⁶ cm⁻³, the carrier dynamics at the investigated densities still lie 232

within the trap-limited regime,^[36,37] where the electron (hole) population can be shown to
scale linearly with intensity within the SRH formalism.

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236 Since the photocurrent is proportional to the number of free charge carriers and therefore composed of electron and hole current density, $j = j_e + j_p$, we expect that the general light 237 238 intensity dependence in the simplest approximation can be described with a linear and square 239 root contribution. The ratio between this sublinear and linear component is therefore an 240 indication for the degree of trap-limited behaviour of the perovskite films, with greater linear 241 component suggesting lower degree of carrier trapping. The photocurrent extracted from films 242 grown on C-PCBSD in Figure 4b follows a behaviour closer to the linear limit at excitation densities of 10^{13} - 10^{15} cm⁻³ relevant for the PV operation, in comparison to the films grown 243 244 directly on TiO₂ and TiO₂/60-PCBM, indicating a lower density of traps. We have already 245 highlighted above that the wettability of PbI₂ dissolved in DMF on C-PCBSD is quite poor 246 with respect to TiO₂, which leads to the formation of larger and flatter grains and hence a 247 lower density of defects. Interestingly, the intensity dependence of the photocurrent in 248 evaporated MAPbI₃ on TiO₂ and TiO₂/60-PCBM follows exactly the same behaviour, proving 249 similar film growth conditions on both bottom layers. Thus, the choice of the EEL on the 250 bottom strongly influences the opto-electronic property of the perovskite films and thus must 251 be take into account in the device optimization.

We now turn to the measurement of carrier recombination kinetics in working conditions (i.e. under 1 sun light bias), comparing the different devices architecture presented in this work. First, we carried out the measurement of the device capacitances using photo-induced differential charging (PIDC) after subtraction of the geometrical capacitance (**Figure 5**a) as reported before.^[30] Keeping the layers thickness comparable, these measurements allow us to compare the accumulated charge densities in solar cells at different bias. We report such densities in Figure 5b, showing that the solar cell with the C-PCBSD fullerene selective

259 contact is capable to accumulate higher amount of charges. It is worthy to mention that the 260 capacitance increases super linearly for voltages above 0.8 V and it is independent on the 261 thickness of the C-PCBSD layer (see Figures S12 and S13 in the SI). This profile has been also observed in other thin film solar cells alike polymer^[31] and small molecule organic solar 262 cells.^[32,33] While the linear increase of capacitance with the light bias is correlated to the 263 accumulation of carriers at the device contacts,^[34] the super linear behaviour indicates that the 264 265 contacts are not able to sustain the entire density of charges which accumulate in the bulk of 266 the device as well. This is further supported by the fact that charge density, as well as the charges lifetimes, scale as the thickness of the perovskite active layer (see Figures 267 268 S16,S17,S18 of the SI). Importantly, this leads to a greater split of the quasi-Fermi levels for 269 holes and electrons leading to a larger V_{oc} - as long as strong recombination losses do not kick 270 in which then points to the importance of the electronic quality of bulk semiconductor layer.

271 We now can compare the carrier recombination kinetics for the solar cells at equal carrier densities (n). As mentioned above the V_{oc} of solar cells will be influenced by the materials 272 energetics (band alignment) and the carrier recombination kinetics. Figure 6 illustrates the 273 274 measurements carried out using the same devices as in Figure 5 in identical light irradiation 275 conditions. As can be seen, the solar cells using C-PCBSD as electron selective contact show longer carrier lifetimes and lower recombination losses. For instance at illumination 276 277 intensities equivalent at 1 sun the 60-PCBM based devices display a total charge density of 20 nC cm⁻² alike the TiO₂ based ones and carrier lifetimes of 0.6 µs and 2 µs, respectively 278 279 (shown with a vertical line in Figure 6). Nonetheless, considering the same charge density for 280 C-PCBSD perovskite solar cells the carrier lifetime is hundred times slower ($\tau > 200 \ \mu$ s). In fact, illuminating these solar cells at 1 sun the total charge was of hundreds of nC cm⁻² and 281 282 still the carrier recombination lifetime was the slowest between the three devices with a value 283 close to $\tau = 10 \ \mu s$. For a fair comparison with the decays recorded for the organic based

perovskite solar cells, we have used in Figure 6 the fastest component of the bi-exponentialdecay.

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287 In conclusion, by employing a crosslinking strategy we demonstrate a fully solution processed planar junction solar cell in direct architecture. The device delivers a PCE close to 19% and a 288 289 $V_{\rm oc}$ larger than 1.1 V. It is electrically stable, i.e. its J-V characteristics do not depend on the 290 polarization history of the device, and accordingly the photocurrent has a virtually 291 instantaneous response to polarization and remains stable over time. Such high reliability of 292 the device is fully attributed to the high quality of the interface between the C-PCBSD EEL 293 and the perovskite layer. We demonstrate that in such device architecture, under 1 sun 294 illumination, the high carrier density achieved in open circuit condition is not sustained by the 295 extracting contact only, but accumulates within the active layer as well. Thus to enhance the 296 $V_{\rm oc}$, where greater separation of the quasi-Fermi levels is desired, the carrier recombination 297 losses via carrier trapping in the bulk perovskite films has to be minimized. Here, we show 298 that the choice of the EEL layer not only determines the interface energetic, but also the 299 electronic quality of the active layer and, with c-PCBSD we achieve superior characteristics 300 in both these aspects. These results, beyond the technological validity provided by the 301 demonstration of a high efficiency, fully solution processed, flat junction solar cell, which is 302 electrically stable, define an important direction towards an engineered design of highly 303 performing perovskite solar cells.

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- 306 Supporting Information

307 Supporting Information is available online from the Wiley Online Library or from the author.

309 Acknowledgements

310 Fundings from the EU Horizon 2020 Research and Innovation program under the grant

agreement No. 643238 (SYNCHRONICS) and from Fondazione Cariplo (project GREENS n.

- 312 2013-0656)" are acknowledged.
- 313

314 315 316 317		Received: ((will be filled in by the editorial staff)) Revised: ((will be filled in by the editorial staff)) Published online: ((will be filled in by the editorial staff))					
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- 394
- 395 Figures Captions

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Figure 1. (a) Polymerization of PCBSD in solid state at 160 °C. (b) FT-IR spectra of as-cast
PCBSD film (dash line) and after 45 min at 160 °C (solid line). (c) The 1st order exponential
decay of polymerization reaction determined by normalized peak area of the band at 908 cm⁻¹.
The absorption spectra of as-cast PCBSD (c) and C-PCBSD (d) before (solid) and after (dash)
being rinsed with DMF.

402 403

404 **Figure 2.** (a) XRD patterns of PbI₂ on glass/FTO/TiO₂ and glass/FTO/TiO₂/C-PCBSD before 405 and after the transformation. SEM top-view of CH₃NH₃PbI₃ grown on TiO₂ (b, c) and 406 TiO₂/C-PCBSD (d, e).

407 408

409 Figure 3. SEM photovoltaic cross section of complete devices: glass/FTO/TiO₂/CH₃NH₃PbI₃/spiro-OMeTAD/Au glass/FTO/TiO₂/C-PCBSD/ 410 (a) and 411 CH₃NH₃PbI₃/spiro-OMeTAD/Au (b). J-Vcurves for photovoltaic devices FTO/TiO₂/C-PCBSD/ 412 FTO/TiO₂/CH₃NH₃PbI₃/spiro-OMeTAD/Au and (c) CH₃NH₃PbI₃/spiro-OMeTAD/Au (d) measured with different scan rates (V s⁻¹) and different 413 414 scan directions under AM 1.5G white light illumination with an intensity of approximately 415 100 mW cm⁻². Stabilized current density and power output measured close to the maximum 416 power point for TiO_2 (e) and TiO_2/C -PCBSD (f) devices with the corresponding J-V curves 417 shown in (c) and (d), respectively.

418 419

420 **Figure 4.** (a) Sketch of the sample geometry for photocurrent measurement. (b) Photocurrent 421 as a function of excitation density of four perovskite layers deposited via evaporation 422 (eMAPbI₃) or spincoating (sMAPbI₃) on different electron extracting layers. The scattered 423 points represent the measured data, the lines are fits according to $y = c(ax + b\sqrt{x})$. The grey 424 lines show the two extreme cases of the fit model (b = 0 and a = 0, respectively) as a guide to 425 the eyes.

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- 428 429

Figure 5. Capacitance after subtraction of the geometrical capacitance (a) and the charge
density (b) of TiO₂, TiO₂/60-PCBM, and TiO₂/C-PCBSD photovoltaic devices.

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Figure 6. Recombination lifetime versus charge density deduced from TPV measurements of
 TiO₂, TiO₂/60-PCBM, and TiO₂/C-PCBSD photovoltaic devices.

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437

Table 1. Summary of the best device performance for perovskite solar cells with TiO₂ and
 TiO₂/C-PCBSD as EELs with the structure of FTO/EEL/CH₃NH₃PbI₃/spiro-OMeTAD/Au.

					r 1	
Device	Scan direction	Scan rate	V _{oc}	$J_{ m sc}$	FF	PCE
	Scandiection	(V s ⁻¹)	(V)	(mA cm ⁻²)	(%)	(%)

	Povorco ccan	0.2	0.80	19.0	74.5	11.4
TiO	NEVEISE SUdi	0.023	0.85	18.7	73.1	11.6
	Forward coop	0.2	0.73	19.1	64.4	9.0
	FUIWAIU SCAII	0.023	0.73	18.9	64.7	8.9
	Deveree eeen	0.2	1.12	21.1	79.0	18.7
	Reverse scan	0.023	1.11	21.1	77.8	18.3
		0.2	1.10	21.2	69.4	16.2
	Forward scan	0.023	1.11	21.1	76.1	17.8

- **Fully solution processed direct perovskite solar cells with planar junction** are realized by 442 incorporating a crosslinked [6,6]-phenyl-C61-butyric styryl dendron ester layer as electron 443 extracting layer. They deliver power conversion efficiencies close to 19% and an open circuit 444 voltage exceeding 1.1 V with negligible hysteresis. A perovskite film with superb 445 optoelectronic qualities is grown, which reduces carrier recombination losses and hence 446 increases V_{oc} .
- 448 Keyword: perovskite solar cells, interface engineering, planar junctions, solution process,
- 449 open-circuit voltage

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 454 and Annamaria Petrozza*

456 Fully Solution Processed Direct Perovskite Solar Cells with Planar Junction: How the 457 Charge Extracting Layer Determines the Open Circuit Voltage

- 457 Charge Extracting Layer Determines the Open Circuit Voltage

