

1 **High Detectivity Perovskites Light Detectors Printed in Air from** 2 **Benign Solvents**

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15 **Efficient light detection by solution processed devices can impact a broad range of**
16 **scientific and technological applications, from security to communication, automation**
17 **and healthcare. The excellent photoconversion properties of hybrid perovskites**
18 **semiconductors have been exploited in this direction, but harsh processing solvents and**
19 **highly controlled growth conditions limit their applicability. Here we propose a simple**
20 **and low temperature approach for the synthesis of methylammonium lead halide**
21 **perovskite inks based on sub-micrometer sized particles with tunable band-gap. The**
22 **particles allow to formulate inks in benign solvents, such as isopropanol, and to print**

23 **under ambient conditions the photoactive layers of planar photoconductors with scalable,**
24 **large-area coating techniques. When surface traps are passivated with [6,6]-phenyl-C61-**
25 **butyric acid methyl ester fullerene (PCBM), a high photoconductive gain, exceeding 200**
26 **in the blue, and reduced noise, produce record high Specific Detectivities exceeding $7 \times$**
27 **10^{13} cmHz^{0.5}/W, and Gain-Bandwidth product values of 7.5×10^6 Hz. Given the extreme**
28 **simplicity of the presented device architecture and the straightforward processing,**
29 **yielding printable light detectors rivalling established technologies, the present work**
30 **promises a short time deployment of printed perovskite detectors in a multitude of opto-**
31 **electronic applications.**

32

33 Halide perovskite semiconductors have the potential to merge the highly efficient operational
34 principles of conventional inorganic semiconductors with the low-temperature solution
35 processability of emerging organic and hybrid materials. Since the first two reports on high
36 efficiency, solution processed, solid-state hybrid perovskite based solar cells in 2012,^{1, 2} there
37 has been a world-wide explosion of activities on these materials, with laboratory scale solar
38 cells power conversion efficiencies recently exceeding 22%.³ The growing interest in 3D
39 hybrid perovskite semiconductors has soon expanded their field of application, with proof of
40 concept electroluminescent and lasing devices,⁴ thus potentially impacting optoelectronics
41 under many aspects. Light detection is one of the fields where there is large potential for
42 achieving real applications in a short to medium time-frame. In fact light detectors require some
43 of the key properties that make halide perovskite a good material for solar cells, while moving
44 the competition with existing technologies from a difficult “cost per Watt peak” scenario to a
45 cost per functionality one. Notably, large absorption coefficients of 10^5 cm⁻¹ in the visible and
46 UV range, and long carrier lifetimes allow to closely match, or even to exceed, Si photodiodes
47 performances.^{5, 6, 7, 8, 9, 10} On top, solution processability would result in a drastic simplification

48 and affordability of the manufacturing processes, leaving ample room for advanced pixel and
49 array design for imaging applications, for example. Nevertheless, there are a few aspects that
50 currently limit the potential of perovskites for light detection and in general their use for
51 consumer electronics. The toxicity of lead-based compounds is probably the most discussed
52 topic, thus intense research is already tackling the development of lead-free materials.
53 Moreover recent studies are also raising the argument of the possibility of full re-cycling the
54 lead present in exhaust car batteries¹¹ or in active layer of the opto-electronic devices,¹² thus
55 getting away from the practical and expensive issue of disposal of hazardous materials. Another
56 important aspect regards the organic solvents most often used to prepare perovskite precursor
57 solutions, which present health and environmental hazards.¹³ Furthermore, typical solvents
58 employed in the processing of lead halide-based perovskites possess high boiling points. These
59 aspects complicate the deployment of energy efficient fabrication processes and increases the
60 environmental impact and the cost of waste handling. On top of that, the precursors-based
61 approach restricts the available process window.^{14, 15, 16, 17, 18} In fact the final microstructure,
62 which dominates the optoelectronics properties of the light absorbing layers, depends on how
63 the constituent ions self-assemble during crystallization on the selected substrate, a complex
64 function of precursor ratio, solvents, processing additives, substrate roughness and surface
65 energy, atmospheric/environmental conditions, annealing temperature and treatment time.

66 Both the use of hazardous solvents and the difficult control of the polycrystalline structure may
67 limit the potential advantages presented by this emerging technology. Particle-based ink
68 formulations in safer and more environmentally friendly solvents can represent a successful
69 strategy to fully deploy the technology. Various synthetic procedures for fully inorganic¹⁹ or
70 hybrid organic/inorganic^{20,21} lead halide-based perovskite particles have been reported. Among
71 these procedures, colloidal synthesis has the advantage of allowing a high control and tailoring
72 of the morphological and optoelectronic properties of the semiconductor crystals before

73 deposition, thus helping in the standardization and reproducibility of the subsequently
74 produced films. However, so far, the nanocrystals suspensions are predominantly stabilized by
75 the use of long ligands that are, together with the remnants of high boiling point solvents,
76 difficult to remove. Their complete removal thus requires annealing at elevated temperatures
77 that already overlap with the temperature range where some sensitive perovskite compounds
78 start to deteriorate. This, of course, hampers the fabrication of thin films of good optoelectronic
79 quality. Recently, successful approaches involving the use of shorter ligands have been tested,¹⁹
80 though a full tunability of the spectral range is not yet achievable due to a lack of stability of
81 colloidal suspensions. Moreover, owing to the low concentration of the inks used to hinder
82 aggregation, achieving the required thickness of the semiconductor layer often requires
83 multiple deposition steps. Alternative syntheses of methylammonium lead halide perovskite
84 nanoparticles were also proposed, based on laser ablation,²² grinding of the solid perovskite
85 components,²³ ultrasonication of components^{24, 25, 26} or a combination of grinding and
86 sonication.²⁷ However, all these methodologies still utilize growth controlling capping ligands
87 and/or are hardly scalable and/or involve the use of expensive, high boiling point and often
88 hazardous solvents.

89 Here we report a straightforward and scalable low temperature synthesis of methylammonium
90 lead tri-iodide (MAPbI₃) perovskite particles with tunable dimensions, down to the
91 submicrometer range, based on a cyclic size reduction process of PbI₂, a solid precursor for
92 perovskite particle synthesis. In contrast to previously reported synthetic procedures, our
93 approach employs only water and isopropanol (IPA), which are widely utilized across
94 industrial, medical and household applications and present much lower health hazard and
95 environmental toxicity concerns in comparison to other commonly used solvents.^{13, 28}
96 Synthesized particles are formulated into printable inks with IPA, without additives.
97 Subsequently they are deposited by a large-area bar coating technique²⁹, and coated with a thin

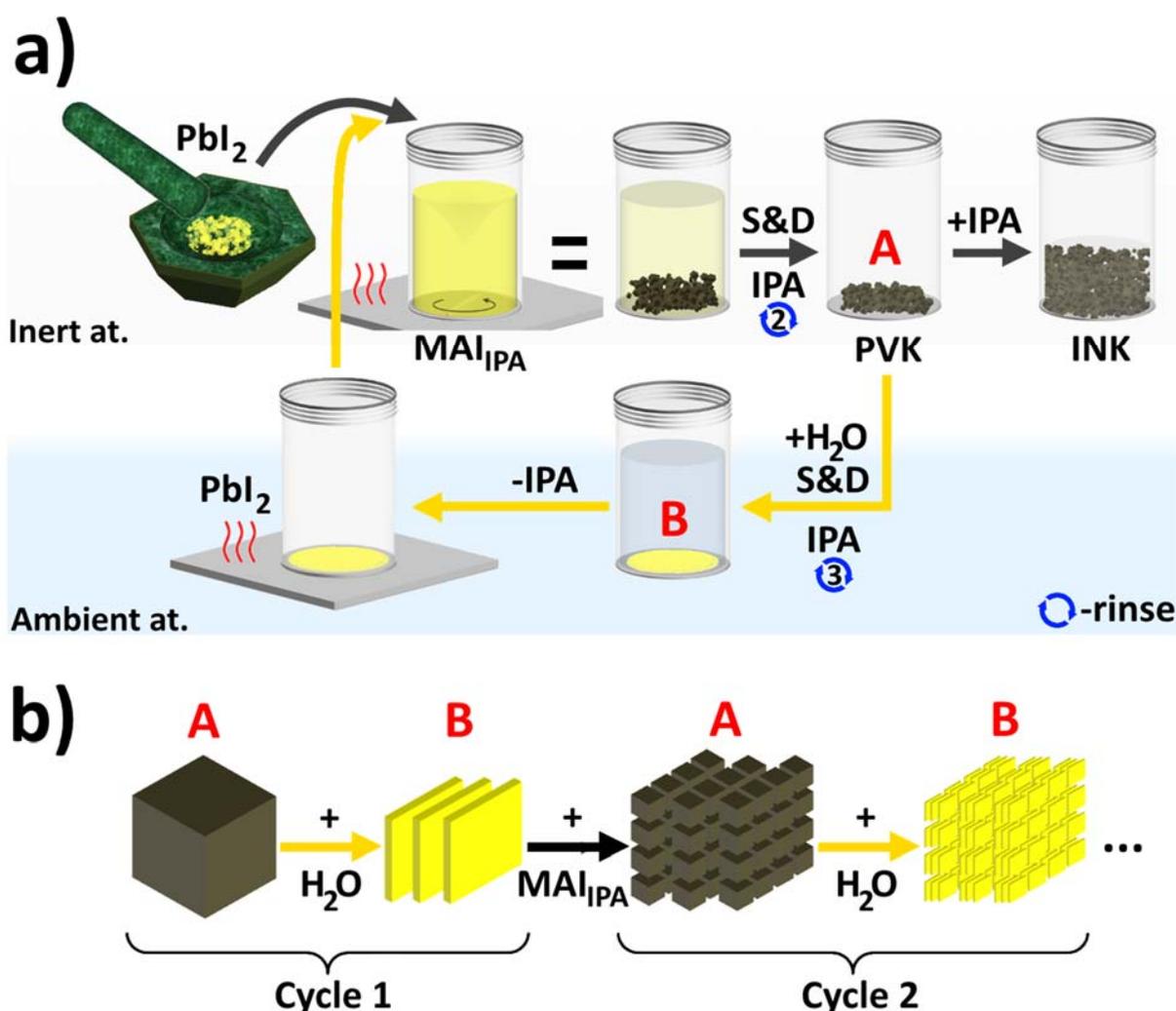
98 [6,6]-phenyl-C61-butyric acid methyl ester fullerene (PCBM) layer, which passivates surface
99 traps, accepts electrons and ensures selective charge circulation of holes. This strategy permits
100 to demonstrate the highest measured specific detectivity in printed light detectors till date,^{14,}
101 ^{15, 16, 17, 18, 30, 31, 32, 33} reaching a maximum value of 7.23×10^{13} cm Hz^{0.5}/W in the visible
102 spectrum, even higher than in commercially available Si photodiodes. Moreover, the gain and
103 response time are at least one order of magnitude higher and lower, respectively, than in any
104 other printed perovskite photodetector, allowing to achieve a Gain-Bandwidth product of $7.5 \times$
105 10^6 Hz, thereby demonstrating a good balance between signal amplification and device
106 response speed.³⁴

107 **Methylammonium lead halide particles synthesis and inks formulation**

108 For the synthesis of **MAPbI₃** perovskite particles we devised a cyclic approach based on the
109 successive reduction of the size of precursor PbI₂ particles. A schematic illustration of the
110 process is shown in **Fig. 1**. In the first cycle, the starting PbI₂ precursor is obtained by grinding
111 as-bought PbI₂ powder in a ceramic mortar, under nitrogen atmosphere, to increase its surface
112 area. 200 mg of PbI₂ is slowly added into 10 ml of highly concentrated (100 g/l) methyl
113 ammonium iodide (MAI) solution in isopropanol under stirring at 80 °C. Upon PbI₂ addition
114 into the MAI solution, immediate insertion of MA⁺ and I⁻ ions into the PbI₂ lattice follows,
115 inducing the growth of MAPbI₃ cubic-like crystals and turning the MAI solution into a
116 sedimenting dispersion of black perovskite particles. Such formed particles are subsequently
117 isolated through sedimentation and decantation of the residual MAI solution and washed with
118 isopropanol for two times. After this first cycle, we obtain MAPbI₃ cubic-like particles with
119 characteristic size of a few (1 to 5) micrometers (see SEM image in Fig. 2a).

120 The successive cycles allow to reduce the average size of MAPbI₃ particles. To start the
121 iteration, the latter are re-dissolved in water, in ambient atmosphere without specific

122 environmental control. The addition of water promotes rapid extraction of MA^+ and I^- ions
 123 from the perovskite lattice that immediately collapses back into water poorly soluble, PbI_2
 124 particles, which were afterwards additionally rinsed with IPA to remove residual MA^+ and I^-
 125 ions. The recrystallized PbI_2 particles appear in the form of thin sheets, indicating layered
 126 structure of PbI_2 ,³⁵ and are smaller in size than the starting PbI_2 precursor. For example, after
 127 the first rinse of perovskite particles with water, we obtained PbI_2 sheets with lateral sizes of a
 128 few micrometers (Fig. 2b) and thickness of several hundreds of nanometers.

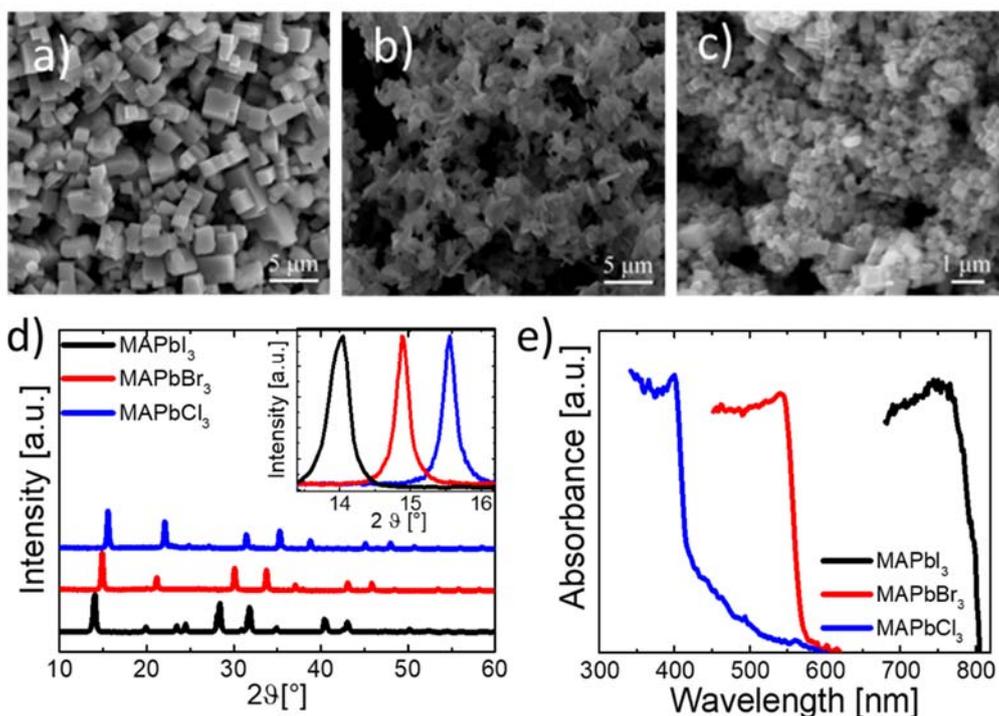


129
 130 **Fig. 1| Synthesis of MAPbI_3 particles.** a) Schematic representation of cyclic MAPbI_3
 131 synthesis, based on three main steps: *i*) addition of PbI_2 (ground, as-bought powder in the first
 132 cycle, extracted PbI_2 sheets with smaller dimensions in the following cycles) into a hot MAI

133 solution under stirring; *ii*) isolation of perovskite particles through sedimentation and
134 decantation (S&D) followed by rinsing with IPA for two times; *iii*) dissolution of perovskite
135 particles in water and precipitation of thin PbI₂ sheets which are isolated through sedimentation
136 and decantation of water solution followed by rinsing with IPA for three times. Such extracted
137 PbI₂ sheets are used as starting precursors for the following steps. b) Scheme of the effect of
138 synthesis cycles on the characteristic size of perovskite particles (A) and lead iodide sheets (B).

139

140 The obtained PbI₂ sheets are then adopted as precursors for a new cycle, following the exact
141 same procedure described before, and obtaining sheets with decreasing lateral dimensions and
142 thickness at each step. Starting from smaller PbI₂ sheets, correspondingly smaller cubic
143 perovskites particles can be isolated. Previously, Ha et al. determined a linear relation between
144 the thickness of PbI₂ sheets grown in the [001] direction and the resulting perovskite platelet
145 thickness grown out of it when gas-induced transformation is applied.³⁶ Assuming that there is
146 no appreciable dissolution neither of PbI₂ or perovskite phase into the MAI solution within the
147 short time frame of particle sedimentation (few minutes), in analogy with the gas-induced
148 transformation where no recrystallization occurs, we infer that the characteristic size of cubic
149 perovskite particles is primarily determined by the PbI₂ sheet thickness.



150

151 **Fig. 2| Characterization of Synthesized Particles.** Top view SEM images of (a) first cycle
 152 synthesis MAPbI₃ particles, (b) recrystallized PbI₂ sheets and (c) final MAPbI₃ particles. (d)
 153 X-Ray diffraction spectra and (e) absorption band edges of synthesized MAPbI₃, MAPbBr₃
 154 and MAPbCl₃ particles.

155

156 By applying the synthetic cycle three times, we were able to obtain MAPbI₃ particles with
 157 characteristic size well below 1 μm (Fig. 2c). Such particles are than dispersed in IPA at a
 158 concentration of 100mg/ml to obtain a printable ink formulation. Further synthetic cycles do
 159 not lead to further reduced particles size. X-ray diffraction measurements show formation of a
 160 pure phase of PbI₂ and MAPbI₃ at each intermediate step (Fig. S1, in the Supplementary
 161 Information, SI) and confirmed that the incorporation of MA⁺ and I⁻ ions into PbI₂ to form
 162 MAPbI₃ is complete after the final step: the corresponding spectrum (Fig. 2d) shows the
 163 characteristic peaks for MAPbI₃ at 14.1°, 28.4° and 31.8°, corresponding to the reflection of

164 the (110), (220) and (310) crystal planes, respectively. No unreacted MAI or PbI₂ (main peaks
165 at 10° and 12.7°, respectively) were detected.

166 Starting from the synthesized MAPbI₃ particles, tuning of the energy bandgap can be achieved
167 through halide exchange. Halide exchange in the MAPbX₃ lattice has been previously reported
168 and it is a facile strategy to tune the semiconductor bandgap. Starting from MAPbI₃, MAPbBr₃
169 or MAPbCl₃, full conversion or mixed halide perovskite (MAPbI_{3-x}Br_x, MAPbI_{3-x}Cl_x,
170 MAPbBr_{3-x}Cl_x), can be obtained depending on the stirring time. This reaction occurs if the
171 solution containing MAPbX₃ crystals is stirred at 65 °C in presence of MAI, MABr or MACl
172 (details reported in [the SI](#)).

173 **Printed light detectors**

174 We have used the formulated inks to print light signal detectors. We adopted a very simple
175 planar architecture, where the MAPbI₃ crystals [ink in IPA](#) is deposited in ambient atmosphere
176 by bar coating, a large-area and roll-to-roll compatible coating technique,^{29, 37} on top of pre-
177 patterned, interdigitated gold electrodes (Fig. 3a, Fig. S2). Such devices have almost symmetric
178 current-voltage (*I-V*) characteristics and show light sensitivity when exposed to a
179 monochromatic radiation at 532 nm of 1 mW/cm² (Fig. 3c). The spectral responsivity $R(\lambda)$ is
180 comprised between 0.1 and 1 A/W (Fig. 3d) from 400 nm to 800 nm (applied bias of 5 V).
181 These performances are drastically improved when we introduce a coating of the printed
182 perovskite layer with the electron acceptor PCBM on top of the perovskite particles. Owing to
183 the voids among particles, the solution permeates the few microns thick photoactive film,
184 leading to a coverage of the surface of the printed perovskites particles with PCBM (Fig. 3b).
185 As an effect, the photoresponse is drastically increased: the perovskite/PCBM detector shows
186 a dark current of 9 nA, an order of magnitude higher than for the pristine perovskite (*vide infra*),
187 and when the device is exposed to the same impinging radiation as before, the photocurrent

188 reaches $4.1 \mu\text{A}$, three orders of magnitude more than in dark (Fig. 3c). As an example, PCBM
 189 deposited on top of a compact MAPbI₃ film produces only a marginal improvement owing to
 190 a much less extended perovskite/PCBM interface (Fig. S3). Such strong photoactivity of our
 191 printed MAPbI₃ crystals with the PCBM coating corresponds to a responsivity that varies
 192 between 10 A/W and 78.7 A/W within the spectral range from 400 nm to 820 nm, the latter
 193 corresponding to the absorption onset of the semiconductor (Fig. 3d). Such performances are
 194 reproducible over different printed photodetectors (Fig. S4). Moreover, the printed perovskite

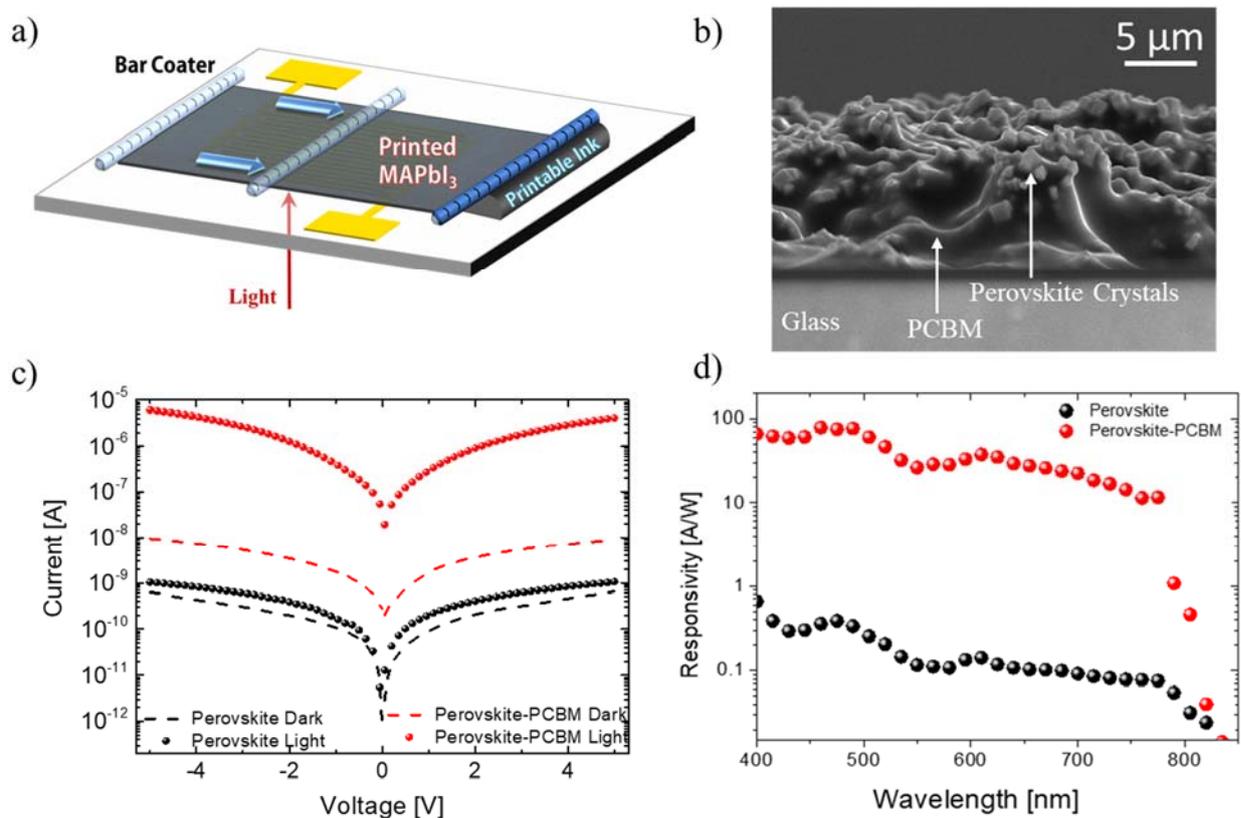
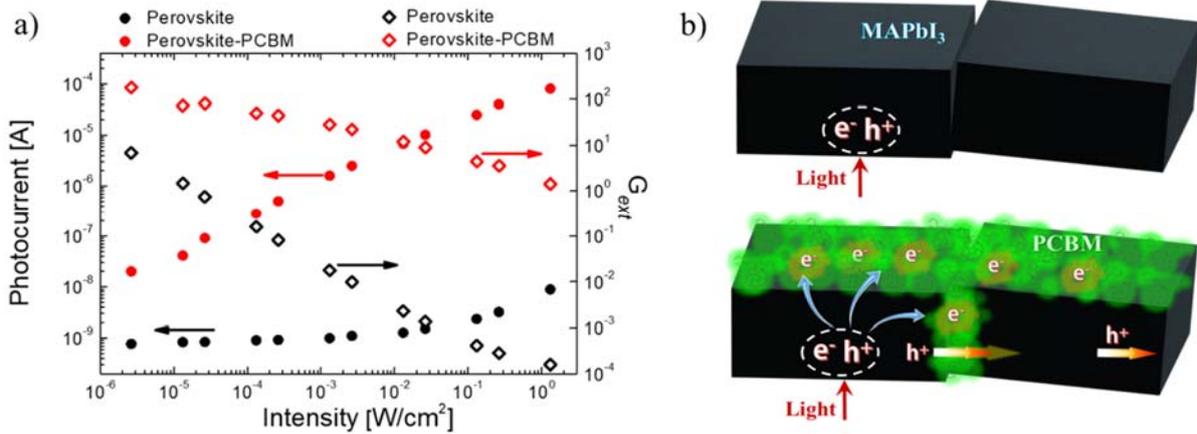


Fig. 3| *I-Vs* and Responsivity of printed MAPbI₃ detectors. **a**, Schematic image of the perovskite photoconductor printed on a simple planar geometry with Au electrodes (channel width, $W = 20$ mm and channel length, $L = 2.5 \mu\text{m}$). **b**, Cross-sectional SEM image of the printed perovskite active layer with PCBM penetrating the crystals. **c**, Dark and light (532 nm LED light, $1\text{mW}/\text{cm}^2$) *I-V* characteristic curves of the photoconductors with and without PCBM. **d**, Responsivity of the printed perovskite photodetectors with and without PCBM passivation at 5 V applied bias.

195 films stored in ambient conditions at a temperature of 300 K and at a relative humidity of 70
196 % does not show any measurable XRD peak of PbI_2 formation and show signature of hydration
197 only after 4 weeks (Fig. S5). Photodetectors show good stability under continuous illumination
198 for an observation time up to 1 h (Fig. S6).

199 Such high responsivity, achieved for a low applied bias voltage of 5 V over a channel length of
200 2.5 μm , suggests a gain mechanism.^{38, 39} In fact, the external quantum efficiency (EQE)
201 calculated in the blue region of the spectrum exceeds 100 %. Such EQE value, together with
202 the symmetric I - V curves, indicates a photoconductive nature of the light detector. The external
203 gain (G_{ext}), defined as the unitless ratio between the number of collected photocarriers and the
204 number of incident photons,^{38, 39} is estimated to be 213 at a wavelength (λ) equal to 460 nm
205 (Fig. S7). Such gain is one order of magnitude higher than in any printed perovskite
206 photodetector reported so far (Table 1). The photocurrent and G_{ext} as a function of the incident
207 light intensity are reported in Fig. 4a. At the low excitation density of 2 $\mu\text{W}/\text{cm}^2$ ($\lambda = 560$ nm),
208 the gain increases from about 6, in the case of the pristine perovskite device, to 180, when the
209 perovskite is coated with PCBM. With the light intensity increasing of almost 6 orders of
210 magnitude, the gain quickly vanishes below 1 for the pristine material, while such decrease is
211 strongly limited in presence of PCBM, with a gain which is still higher than 1 above 1 W/cm^2
212 ($G_{\text{ext}} = 1.44$ at 1.31 W/cm^2). Moreover, with addition of PCBM, the photocurrent is a power
213 law of light intensity, i.e. $\propto I^\alpha$ with $\alpha \approx 0.7$, on the base of which it is possible to retrieve the
214 input light signal from the recorded photocurrent.^{40, 41, 42}



215

216 **Fig. 4| Intensity Dependent Gain.** **a**, Photocurrent and G_{ext} as a function of a constant incident
 217 laser light intensity at $\lambda = 560$ nm for photoconductors with and without PCBM. **b**, Schematic
 218 representation of the photoconduction mechanism: PCBM serves a dual role of acting as an
 219 acceptor and trap for photogenerated electrons, and as passivation of particles surface traps,
 220 thus facilitating holes transport and boosting both G_{ext} and Bandwidth of the detectors.

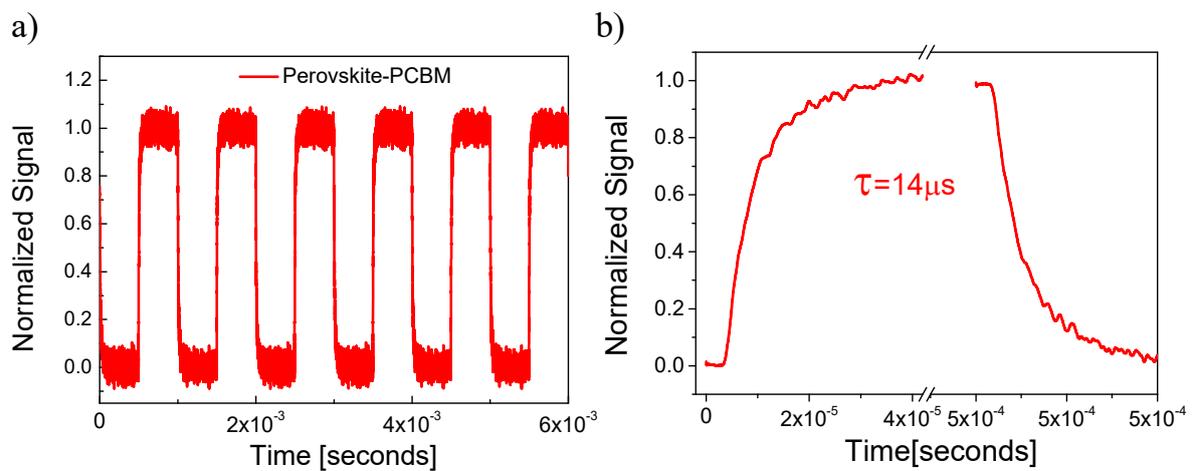
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222 In a photoconductor the gain originates from a strong unbalance in the transport properties of
 223 the photogenerated holes and electrons following the absorption of a photon, where one of the
 224 two carriers is deeply trapped, and the other can drift under the external field.^{38, 39} Once the
 225 mobile carrier is collected at the electrode, another carrier is injected for charge neutrality.
 226 Recirculation of the mobile carrier proceeds until the trapped carrier is finally released,
 227 determining the gain that is proportional to the ratio between the de-trapping time and the
 228 transit time. The presence of unbalanced charge transport in pristine MAPbI₃ upon photo-
 229 excitation has been recently reported;^{43, 44} moreover, the high work function of Au used for the
 230 electrodes introduces a high energy barrier for injection of electrons for photoconduction, while
 231 favoring recirculation of holes. This explains the behavior observed for the detector based on
 232 pristine perovskite. At low light intensity the gain is high since trapping of one of the two
 233 carriers by defect states and selective barrier for electrons at the contact unbalance the transport.

234 As traps are filled with increasing intensity, the transport becomes more balanced and/or the
235 trapping time of the less mobile carrier reduces and the gain is correspondingly smaller. As
236 largely reported in literature,^{45, 46} PCBM has a passivation effect when deposited on the
237 perovskite thin film (Fig. 4b). In fact, as a Lewis acid it is able to oxidize negatively charged
238 PbI_3^- antisites or under-coordinated halide ions which work as deep traps for holes.⁴⁷ Therefore,
239 by penetrating through the porous film of perovskite particles (Fig. 3b), it significantly
240 improves the hole conductivity allowing the carriers to more easily drift through the particles
241 under the externally applied bias: this explains the 10-fold increase in dark current.
242 Simultaneously, it also acts as a sink by extracting photogenerated electrons from the
243 perovskites phase. In such a way, the lifetime of the free holes in the perovskite is further
244 increased, while electrons are confined in the PCBM phase with short diffusion length. All this
245 results in a 30 times enhancement in gain at low excitation density ($\sim \mu\text{W}/\text{cm}^2$), and in a G_{ext}
246 still higher than 1 at high excitation density ($\sim \text{W}/\text{cm}^2$). Since electrons are extracted from the
247 perovskites phase in the PCBM treated device, the device is less sensitive to light intensity,
248 explaining the markedly different intensity dependence of photocurrent and gain between
249 pristine and PCBM coated perovskite.

250 As photoconductors are characterized by a trade-off between gain and response speed, or
251 bandwidth, it is important to assess the transient time response of our printed perovskite/PCBM
252 detectors. We evaluated the time response by irradiating the samples with a light pulse train
253 with 500 μs pulse width and recording the transient photoresponse. The printed detectors
254 follow very well the train of pulses (Fig. 5a, Fig. S8). The rise and fall times of the
255 photoconductor with PCBM, defined as the time taken by the photocurrent to rise from 10 %
256 to 90 % of the regime value, and vice-versa, are 14 μs in both cases (Fig. 5b). This photo-
257 response is more than one order of magnitude faster than in previously reported printed
258 perovskite photodetectors (Table 1). Such response time would allow the detector to follow

259 light signals with frequencies up to $f_{max} \approx 35$ kHz. Notably, despite the higher gain, it is about
 260 4 time faster with respect to a device which uses only pristine perovskite (rise time about 55
 261 μs , $f_{max} \approx 9$ kHz, Fig. S8). A decrease in transient response time with PCBM, and a concomitant
 262 increase of the gain, corroborate our picture of an improvement of the transport mechanism for
 263 holes within the perovskite layer, and an effective trapping of electrons in the acceptor PCBM
 264 phase. We have calculated the Gain–Bandwidth product of our detector, a figure of merit useful
 265 to compare combined performances of detectors with gain.³⁴ We obtain values as high as 7×10^6
 266 Hz at $10 \mu\text{W}/\text{cm}^2$ for $\lambda = 460$ nm, which is a few orders of magnitude higher than previously
 267 reported printed perovskite photodetectors. With respect to all perovskite photodetectors
 268 reported so far, irrespectively of the fabrication process, our device photodetector shows one
 269 of the most **optimized** Gain-Bandwidth tradeoff (Fig. S9).



270

271 **Fig. 5| Detector Transient Response. a**, Response of the perovskite-PCBM detector to a light
 272 pulse train (532 nm LED, 500 μs pulse width, $1 \mu\text{W}/\text{cm}^2$ intensity) with 5 V bias. **b**, Transient
 273 response of the photodetector with equal rise and fall times of 14 μs .

274

275 **Noise characterization and Specific Detectivity**

276 Another key feature of a light detector is related to its sensitivity, namely the possibility of
 277 detecting small signals. The minimum signal for which an electrical response is discernable is
 278 ultimately limited by the intrinsic device noise. One of the most adopted figure of merits to
 279 quantify this aspect is the Specific Detectivity (D^*), which normalizes the Noise Equivalent
 280 Power (NEP), *i.e.* the power of a sinusoidal monochromatic radiation producing the same r.m.s.
 281 output signal of the device current noise (i_n) in an ideal noise-less device over a Δf bandwidth,
 282 to the active area (A):

$$283 \quad NEP = \frac{(\overline{i_n^2})^{0.5}}{R(\Delta f^{0.5})} [\text{W}]/[\text{Hz}^{0.5}]$$

$$284 \quad D^* = \frac{(A)^{0.5}}{NEP} [\text{cmHz}^{0.5}]/[\text{W}]$$

285 First estimations of D^* can be obtained by assuming that the limiting noise is represented by
 286 the white shot noise owing to fluctuations of the dark current. However, this approximation
 287 may lead to large overestimation of D^* ,⁸ especially at low frequencies where $1/f$ noise, so called
 288 “flicker” noise can be significant. Such scenario is even more critical for photoconductors,
 289 since traps, at the base of photoconduction, can significantly contribute to the flicker noise,^{6,9}
 290 pushing the corner frequency (f_c) between $1/f$ and the white shot noise, *i.e.* the frequency above
 291 which shot noise dominates D^* , to high frequency. If the device cannot operate at such high
 292 frequency, the assumption that white noise dominates completely fails. Therefore, D^* has to be
 293 quantitatively evaluated by performing accurate noise measurements. We have measured the
 294 equivalent current noise of both printed perovskite detectors, with and without PCBM, over a
 295 large bandwidth (10 Hz - 150 kHz, details on the measurements are reported in the
 296 Supplementary Information), with the aim to assess the effect of the organic coating on noise
 297 as well. In the investigated range, the noise of the printed detector without PCBM is completely
 298 dominated by flicker noise, thus reaching high level of noise over the whole bandwidth at which
 299 the detector can be operated (Fig. 6a). At the maximum operative frequency of 9 kHz for the

300 pristine perovskite device $(\overline{i_n^2})^{0.5}$ is $\sim 1.1 \times 10^{-13}$ A, almost two orders of magnitude higher
 301 than the shot noise, corresponding to a NEP of $0.16 \text{ pW/Hz}^{0.5}$, by assuming a 1 Hz bandwidth.
 302 Very interestingly, when the PCBM coating is introduced, the flicker noise is reduced by
 303 almost two orders of magnitude, so that f_c is retroceded within the investigated range, falling
 304 at 4 kHz (Fig. 6a). Since the perovskite-PCBM device can be operated up to 35 kHz, there is a
 305 frequency window from f_c up to such maximum frequency where the noise in the device is

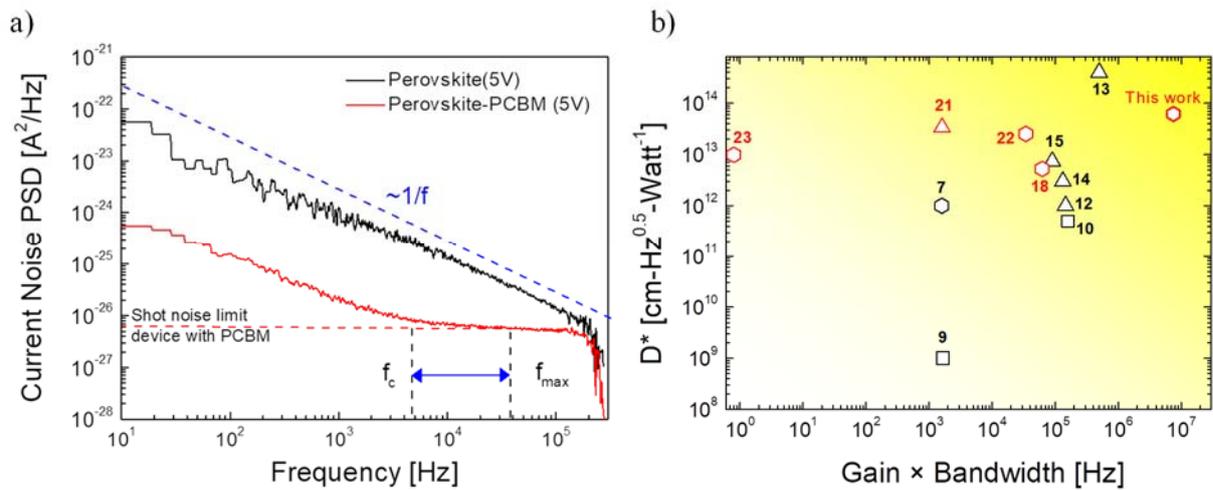


Fig. 6| Noise characteristics and device performance comparison. a, Current Noise Power Spectral Density (PSD) of the photoconductors with and without PCBM. Shot noise limit, corner frequency f_c , maximum operating frequency f_{max} , and frequency range where detectors are operated at shot noise limit is shown for devices with PCBM. **b,** Comparison chart for all types of perovskite photodetectors including the best performing printed photodetectors from all technologies including perovskites. Only devices with measured D^* are included. Red symbols: Best Printed Detectors; Black symbols: Non-Printed perovskite detectors; \circ : Metal-Semiconductor-Metal Photoconductors; \square : Photo-Gated Transistors; Δ : Photodiodes. Reference numbers correspond to the list reported in the Supporting Information, where a complete comparative table and method for comparison is reported.

306 limited purely by the shot noise. This results into a considerably lower $(\overline{i_n^2})^{0.5}$ of $\sim 7.5 \times 10^{-14}$
307 A above f_c , with a NEP of $0.95 \text{ fW/Hz}^{0.5}$, despite of the higher dark current observed in devices
308 with PCBM (Fig. 3c). Considerable reduction of $1/f$ noise upon interfacing perovskite crystals
309 with PCBM is an interesting evidence and confirms that PCBM acts as a trap passivation (Fig.
310 4b).^{45,46} We can suggest that PCBM acts on surface trap states of the crystals, which determines
311 the higher flicker noise in the device without PCBM.⁴⁸ We compared the device noise, where
312 the flicker noise dominates (500 Hz), with the noise expected from the Hooge's empirical
313 formula, used to rationalize the $1/f$ noise in terms of fluctuations in the electrical conductance
314 of homogeneous samples: $(\overline{i_n^2}) = \alpha_H I^\gamma / (fN)$ (Fig. S10),⁴⁹ where α_H is Hooge's parameter, I
315 is the dark current, N is the total number of carriers. In the case of the pristine perovskite device,
316 the Hooge's formula hardly fits the data, and $\gamma > 2$ is extracted on a limited range of currents
317 and frequencies (Fig. S10). A value of γ exceeding 2 is seldom observed in literature, and may
318 suggest a complex source of noise.^{48,49,50} When PCBM is introduced, the low frequency flicker
319 noise correlates well with the Hooge's formula for homogeneous samples ($\gamma = 2$, Fig. S10).
320 Accumulation of ions at the grain boundaries in the presence of electric field have been
321 consistently observed in MAPbI₃ thin films.⁵¹ This effect would be exacerbated in the case of
322 the pristine perovskite film, where the connectivity between individual perovskite crystals is
323 lower than in spin coated films. Large presence of accumulated I⁻ ions at the grain boundaries
324 is expected to give rise to inhomogeneity in energy landscapes, contributing to an increased
325 noise. The passivation of ions by the presence of PCBM strongly reduces this source of noise,
326 and renders γ equal to 2, as often verified in inorganic elemental semiconductors.

327 Thanks to the noise measurement, we can extract the specific detectivity D^* , which for the
328 perovskite-PCBM detector is very high in all visible spectrum, reaching a maximum of $7.23 \times$
329 $10^{13} \text{ cmHz}^{0.5}/\text{W}$ at $\lambda = 460 \text{ nm}$ (Fig. S11). Our printed perovskite particles device shows one
330 of the highest measured D^* for all perovskites detectors reported till date (Table S1) and is at

331 least one order of magnitude higher than any printed perovskite photodetector (Table 1). In
 332 Fig. 6b (Table S1) we plot D^* versus the Gain-Bandwidth product, in our case and for the best
 333 perovskite photodetectors reported so far where such figures of merit have been reported. We
 334 also compare the performance with state of the art printed photodetectors from established
 335 technologies such as quantum dots/nanocrystals and organic photodetectors with highest
 336 measured D^* . The very good overall performance of our detector is a consequence of a unique
 337 combination of trap passivation by PCBM, that reduces both the mobile (holes) carrier transit
 338 time and the $1/f$ noise, while simultaneously acting as an electron acceptor ensuring a selective
 339 hole circulation to achieve high gain.

340 **Table 1:** Main figure of merits of printed perovskite photodetectors with the highest D^*
 341 reported in literature and in this work. Theoretical calculated D^* by assuming shot noise
 342 limitation are marked with “!”.

343

Printing Technique	R (A/W)	G_{ext}	Resp. time (μ s)/ Bandwidth (Hz)	Detectivity $\times 10^{10}$ cmHz ^{0.5} /W	Intensity [μ W/cm ²]	Voltage Applied	Ref. #
Roll-to-Roll Gravure	5.2×10^{-3}	!	!	!	36,000 μ W	10 V	14
Inkjet Printing	1.2	2.3	10,000 μ s	239(!)	100 μ W	10 V	15
Blade Coating	13.5	30	240 μ s	524	500 μ W	5 V	16
Blade Coating and Hot Casting	8.95	20.9	7700 μ s	290(!)	3700 μ W	10 V	17
Blade Coating	10.75	25	9000 μ s	!	!	10 V	18
Bar-coating	78.7	213	14 μ s	7230	10 μ W	5 V	This Work

344

345 **Conclusions**

346 In summary, we have formulated inks based on sub-micrometer organic-inorganic perovskite
347 particles first synthesized in water and isopropanol and then dispersed in non-hazardous alcohol
348 solvents. Simple planar light detectors were fabricated by printing the perovskite active phase
349 by means of bar-coating, a large area and roll-to-roll compatible technique. The obtained
350 devices show excellent photoconducting response when the active phase is functionalized with
351 PCBM. The fullerene derivative on the one hand passivates hole traps, enhancing hole transport
352 and reducing the rise and fall time to 14 μ s, corresponding to a maximum operation frequency
353 of 35 kHz; on the other, it acts as an acceptor phase for electrons, thus increasing the
354 photoconductive gain. As a result, the device achieves a record high gain-bandwidth product
355 of 7.5×10^6 Hz for printed perovskite light detectors. Importantly, the defects passivation also
356 strongly suppresses $1/f$ noise, so that shot noise dominates already above 4 kHz. Thanks to the
357 good dynamic response, the device can be operated in such frequency region, reaching the very
358 high specific detectivity of 7.23×10^{13} cmHz^{0.5}/W which is the highest measured D^* for any
359 printed photodetector till date.

360 Thanks to our perovskite particles inks we therefore demonstrate the fastest printed perovskite
361 detector and the highest gain-bandwidth product so far. Moreover, the specific detectivity
362 values observed are higher than commercially available silicon photodiodes, which is
363 remarkable considering our photoconductor is printed with a very simple geometry. Our work
364 therefore opens up attractive possibilities to use printable perovskites for large area light
365 detection applications.

366 **Acknowledgements**

367 V.V., P.T. and A.P. would like to acknowledge funding from the DESTINY network within
368 the European Union Seventh Framework Programme [FP7/2007-2013] under grant agreement

369 316494. Authors are thankful to Michele De Bastiani, Marina Gandini and Nikhil Tiwale for
370 useful discussions.

371 **Authors Contributions**

372 P.T. and R.S. synthesized and characterized the materials, produced and printed the inks on
373 inter-digitated electrodes prepared by D.N.. V.V. developed the fullerene passivation,
374 performed all photodetector measurements and analyzed photodetector results with M.C. and
375 A.P. . V.V. performed noise measurements and analyzed the data with G.F. and M.C. . V.V.
376 and S.N. performed intensity dependent measurements. M.C. and A.P. guided the project. All
377 the authors contributed to draft and revised the manuscript.

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