

1 DOI: 10.1002/ ((please add manuscript number))

2 Article type: Full Paper

3

4 **High-performance Flexible Nanoporous Si-Carbon Nanotube Paper Anodes**  
5 **for Micro-battery applications**

6

7 Erika Biserni<sup>1,2</sup>, Alice Scarpellini<sup>3</sup>, **Andrea Li Bassi<sup>2,3,1,2</sup>**, Paola Bruno<sup>3,1</sup>,

8 Yun Zhou<sup>4</sup>, Ming Xie<sup>5,\*</sup>

9

- 10 1. Center for Nano Science and Technology @PoliMi, Istituto Italiano di Tecnologia, Via  
11 Pascoli 70/3, 20133 Milano, Italy
- 12 2. Department of Energy, Politecnico di Milano, Via Ponzio 34, 20133 Milano, Italy
- 13 3. Department of Nanochemistry, Istituto Italiano di Tecnologia, Via Morego 30, 16163  
14 Genova, Italy
- 15 4. Chongqing Normal University, College of Chemistry, Chongqing, 401311, China
- 16 5. Wuhan ATMK Super EnerG Technologies Inc, Wuhan, P.R.China

17

18 Corresponding author: Ming Xie [xmdxx1182@hotmail.com](mailto:xmdxx1182@hotmail.com)

19

20 Keywords: Nanoporous, **flexible**, Si-CNT paper anodes

21

22

23 **Abstract**

24 Nanoporous Si has been grown by Pulsed Laser Deposition on a free-  
25 standing carbon nanotube (CNT) paper sheet for micro-battery anodes. The Si  
26 deposition shows the conformal coverage on the CNT paper, and the Si-CNT paper  
27 anodes demonstrate high areal capacity of  $\sim 1000 \mu\text{Ah}/\text{cm}^2$  at a current density of  
28  $54 \mu\text{A}/\text{cm}^2$ , while, 69% of its initial capacity is preserved when the current density  
29 is increased by a factor 10. Excellent stability without capacity decay up to 1000  
30 cycles at a current density of  $1080 \mu\text{A}/\text{cm}^2$  is also demonstrated. After bending along  
31 the diameter of the circular paper disc many times, the Si-CNT paper anodes  
32 preserve the same morphology ~~and electrochemical performances~~, indicating that  
33 nanoporous Si-CNT paper anodes can find its application for flexible micro-  
34 batteries.

35

## 36 **1. Introduction**

37 With the rapid growth in the fields of microelectronics and wearable devices, the  
38 development of integrated flexible power sources that enable the continued device  
39 operation is of great importance. To meet these demanding applications, batteries with high  
40 energy and power density per unit area are urgently required [1-2]. In addition, wearable  
41 devices require that batteries should be highly flexible and have long cycling life. It would  
42 be also beneficial that the micro-battery fabrication process is compatible with state-of-the-  
43 art integrated circuit (IC) techniques in order to lower the costs.

44 Silicon alloys with lithium up to  $\text{Li}_{15}\text{Si}_4$  [3-4] at room temperature, this resulting in  
45 a theoretical capacity almost 10 times larger ( $3579 \text{ mAh/g}$  for  $\text{Li}_{15}\text{Si}_4$ ) than that of graphite  
46 ( $372 \text{ mAh/g}$ ) [5], which is the standard commercial material for negative electrodes.

47 However, Si-based anodes need to overcome some major barriers for large-scale  
48 implementation in micro-batteries. First, they suffer from poor cycle life due to detrimental  
49 volume changes (i.e., theoretically up to 280% volume expansion[6]) of the host lattice  
50 upon alloying and de-alloying with Li. Extended fractures lead to a complete loss of  
51 electrical contact between active material and current collector and to pulverization.  
52 Second, the solid-electrolyte ~~interphase~~ ~~interface~~(SEI) layer in contact with common  
53 electrolytes is generally unstable and this detrimentally impacts on the capacity retention  
54 [7-8].

55 So far, silicon planar thin-film anodes for micro-batteries can only be realized  
56 within sub- $\mu\text{m}$  thickness[9-13], leading to low areal capacities ( $\text{mAh}/\text{cm}^2$ ). Therefore, it is  
57 important to develop Si thin film anodes with enhanced areal capacity, meanwhile offering  
58 a long and stable lifetime by managing mechanical instability due to volume changes.  
59 Secondly, the Cu current collector usually used for the anodes is very heavy and considered  
60 as “dead weight”. It would be advantageous to eliminate the “dead weight” or replace it  
61 with some other lighter current collector, such as a carbon nanotube (CNT) paper which  
62 itself can also contribute to capacity. Fabrication of Si flexible micro-anodes by slurry  
63 coating with highly volatile organic solvents is incompatible with the state-of-the-art  
64 fabrication of microbatteries that usually require vacuum deposition.

65 Previously, we addressed these drawbacks by a two-layer architecture [14]. We  
66 fabricated novel nanocomposite Si-C anodes by depositing nanostructured porous  
67 amorphous Si films onto a Cu current collector by pulsed laser deposition (PLD) at room  
68 temperature, followed by chemical vapor deposition (CVD) of a thin carbon coating. The  
69 mesoporosity of the nanostructured Si films and its lack of crystallinity are expected to

70 reduce the detrimental effects of volume variations and to avoid mechanical stressing due  
71 to amorphization in the first cycle. The thin CVD-grown carbon layer is expected to  
72 promote the formation of a stable solid electrolyte interphase (SEI) layer and protect Si  
73 from direct contact with the electrolyte. In this work, we replace the Cu current collector  
74 with flexible CNT papers, and deposit Si film as thick as 3.5  $\mu\text{m}$  by PLD but without the  
75 need for further CVD carbon coating. Eliminating the CVD step prevents high temperature  
76 processes and simplifies the whole process. These flexible free-standing Si-CNT paper  
77 anodes show a much higher capacity than that obtained in our previous work, and stable  
78 electrochemical performance up to 1000 cycles.

79

## 80 **2. Experimental**

### 81 *2.1 CNT Paper Fabrication*

82 Large-area CNT paper can be fabricated using a filtration method using a  
83 proprietary solvent developed by Wuhan ATMK Super EnerG Technologies Inc. Briefly,  
84 MWCNTs purchased from NanoTech Labs Inc. with a large aspect ratio of  $\sim 10^4$ , are  
85 dispersed by ultrasonic probe in the solvent without surfactants for 30 minutes. No organic  
86 binder is needed. The dispersion is then poured into a home-made vacuum filtration  
87 machine. The whole filtration process takes less than 15 seconds. Then the CNT paper with  
88 the filtration paper together is transferred into an oven at 100°C for 4 hours until dried.  
89 Afterwards, CNT paper can be easily peeled off from the filtration paper. The size of CNT  
90 paper can be as large as 30cm by 30cm and its surface resistivity is as low as  $\sim 0.1\Omega/\square$ .

91 CNT paper prepared as here described has an areal weight of 21.7 g/m<sup>2</sup>, which is  
92 much lower compared to the areal weight 806.4 g/m<sup>2</sup> of a Cu foil of the same thickness.

### 93 2.2 Si Film Deposition

94 Si films with the thickness of 3.5 μm were deposited on CNT substrates kept at  
95 room temperature by PLD in a mixture of Ar and H<sub>2</sub> as a background gas, with analogous  
96 procedure as described in our previous work [14]. The background gas pressure was set to  
97 60Pa to reach a meaningful balance between introducing a considerable amount of pores  
98 in the film and preserving its mechanical stability. According to previous investigations,  
99 these deposition parameters allow to grow nanostructured films with hierarchical  
100 mesostructure where the cluster size is in the order of ~10nm.

101 Flexible CNT discs (1.3mm diameter, average thickness 90μm) were cut out from  
102 CNT paper and used as substrates. The thickness of Si film was determined from cross-  
103 sectional Scanning Electron Microscope images taken from a film deposited on a planar  
104 substrate under the same conditions. Samples for HRSEM, TEM and Raman analyses were  
105 prepared on purpose on a CNT substrate by depositing a 1μm-thick Si film under the same  
106 conditions.

### 107 2.3 SEM and TEM Characterizations

108 HRSEM images were taken with a JEOL JSM-7500F instrument, equipped with a  
109 cold field emission gun source and operating at 5 kV. TEM imaging and electron  
110 diffraction analyses were carried out with a Jeol JEM 1011 instrument, operated at 100 kV,  
111 equipped with a thermionic tungsten source. The sample, due to the weak and soft nature

112 of the CNTs disc, has been prepared by picking a small quantity of film with the tweezers  
113 and dispersing it in toluene, successively dropped it onto a holey carbon coated grid.

#### 114 *2.4 Raman Characterization*

115 Raman spectra were acquired upon excitation by the second harmonic (532nm) of  
116 an air-cooled Nd:YAG laser. Laser power was kept below 0.4 mW (sample surface) while  
117 sampling Si films, in order to avoid laser-induced annealing effects. Spectra were recorded  
118 in the range 100-1800 $\text{cm}^{-1}$  in the Stokes region and were calibrated against the 520.5 $\text{cm}^{-1}$   
119 line of an internal silicon wafer reference. The signal-to-noise ratio was enhanced by  
120 repeated acquisitions.

#### 121 *2.5 Electrochemical Characterization*

122 Half-cells were assembled with Si-CNT papers with a diameter of 3/8" as working  
123 electrodes and Li metal foil as reference electrode. Si-CNT papers were bent 5 times along  
124 the diameter before cell assembling. The electrolyte was 1 M LiPF<sub>6</sub> dissolved in a 1:1  
125 (volume ratio) mixture of ethylene carbonate (EC) and diethyl carbonate (DEC); the  
126 separator was a glass microfibre disc (Whatman GF/F), and the shell was a stainless steel  
127 CR2032 coin cell (VWR Inter.). Cells were then tested with an Arbin 2000 battery test  
128 station under constant current conditions, within the voltage range 1.5V - 5mV.

#### 129 *2.6 Bending test*

130 In order to assess the bendability of Si-CNT anodes, a dedicated Si-CNT paper disc  
131 was bent repeatedly for 30 times along the diameter, so as to alternate traction-compression  
132 stresses on the film. To perform this bending test, the Si-CNT disc was bind to a paper  
133 sheet by taping two opposite points of one diameter; the paper sheet was then hold in

134 proximity of these two points and bent repeatedly by hands above and below the starting  
135 plane; bending amplitude was  $>45^\circ$  above starting plane and  $>45^\circ$  below. Bending  
136 frequency allowed to complete a whole bending cycle (above and below starting plane) in  
137 1 second. With this procedure, the diameter perpendicular to the binding points sees the  
138 highest strain conditions.

139

### 140 3. Results and Discussion

141 Figure 1(a) and (b) (same scale) show the difference between the silicon-coated  
142 CNT paper and the uncoated sample. No silicon micrometer-size particles or  
143 agglomeration can be observed. The silicon coverage appears conformal along CNTs. The  
144 cross-section image in figure 1(c) indicates that, due to the high presence of pores from the  
145 CNT paper, the silicon film has grown preferentially on each CNT as a nucleation site  
146 instead of forming a uniform layer like a film deposition onto a planar substrate. The Si  
147 film fabricated by PLD exhibits a hierarchically nanostructured morphology with  
148 mesoporosity, as shown in figure 1(d) and 2(a). Figure 2(a) evidences that the Si film  
149 covering the carbon nanotube has a hierarchically organized algae-like mesostructure.  
150 Bright field (BF) TEM images in figure 2(b) and (c) taken on the same sample confirm the  
151 hierarchical structure of the silicon film, with no trace of crystalline Si phases. Si film  
152 thickness assessed by SEM images on calibration samples returns the value of  $3,5\mu\text{m}$ , thus  
153 accounting for a Si/C weight ratio estimate of 0,15.

154 From SEM and TEM images, it is clear that the Si film, while uniformly covering  
155 the CNT surface, keeps its nanogranular morphology and this promotes the formation of a

156 nanoscale/mesoscale porosity that consists of inter-cluster voids. In a previous work by  
157 some of the authors, the porosity of Si films grown with the same procedure was  
158 investigated by the Brunauer-Emmet-Teller method. The method analyses the  
159 adsorption/desorption isotherms of gases at low temperature to determine specific surface  
160 area of porous samples. Exploiting the BET theory, the calculated surface area of the film  
161 is 109 m<sup>2</sup>/g. For more details on porosity characterization, the reader is addressed to  
162 [\*please add DOI 10.1149/2.0531509jes].

163 In order to assess the bendability of Si-CNT anodes, a dedicated Si-CNT paper disc  
164 was bent repeatedly for 30 times along one diameter, so as to alternate traction-  
165 compression stresses on the film. From HRSEM images (figure 3(a-d)) taken from on the an  
166 area close to its diameter under subjected to the highest strain conditions, the film looks  
167 almost intact with respect to the unstressed state and Si remains attached to CNT (figure  
168 3(a, b)). Only in some spots, Si detaches from CNTs, as shown in figure 3(d). All of the  
169 above images are obtained prior to cell assembling.

170 Raman spectra were acquired on dedicated samples made by depositing a 1µm-  
171 thick silicon film on a disc cut off a CNT paper (Figure 4). In the spectral range 50-600  
172 cm<sup>-1</sup> the features of amorphous Si are clearly visible, with the four characteristic Gaussian  
173 bands centred at 145, 330, 430 and 490 cm<sup>-1</sup>, that are generally attributed to the transverse  
174 acoustic (TA), longitudinal acoustic (LA), longitudinal optic (LO) and transverse optic  
175 (TO) modes of amorphous silicon, respectively[15]. In the spectral range 1200-1800 cm<sup>-1</sup>  
176 instead, the two D and G bands coming from the CNT foil are clearly defined (see reference  
177 spectrum taken on CNT foil), with D band (disordered carbon) being centred around 1350  
178 cm<sup>-1</sup> and G band (graphitic carbon) centred around 1590 cm<sup>-1</sup>. As confirmed by Raman



179 spectroscopic analysis, Si film is fully amorphous and this is well consistent with BF-TEM  
180 observations and with previous analysis on Si film grown by PLD [\*please add ref. DOI:  
181 10.1007/s11051-014-2461-8].

182 In figure 5(a) the capacity of the Si-CNT anodes is presented under different areal  
183 current densities. The initial capacity of  $\sim 1500 \mu\text{Ah cm}^{-2}$  is due to formation of the SEI  
184 layer, resulting in a 35% losses from the 2<sup>nd</sup> cycle. Si-CNT paper shows the highest  
185 capacity of  $\sim 1000 \mu\text{Ah cm}^{-2}$  at  $54 \mu\text{A cm}^{-2}$ . Some micro-battery anodes reported in  
186 literatures are summarized in Table1 for comparison with this work. At  $540 \mu\text{A cm}^{-2}$ , 69%  
187 of its initial capacity is still preserved, indicating exceptional rate capability of Si-CNT  
188 paper anodes. Si-CNT paper anodes show excellent coulombic efficiency ( $>99\%$ ) when  
189 reaching stable condition. The mesoporous silicon on CNTs appears capable of tolerating  
190 volume change, as confirmed by the high coulombic efficiency during cycling.

191 Coulombic efficiency could be effectively increased by using additives to the electrolyte.  
192 In particular, the addition of Fluoroethylene carbonate and Vinylene carbonate has proven  
193 to be highly beneficial in the formation of a stable SEI and the reduction of irreversible  
194 losses [\*please add DOI 10.1016/j.jpowsour.2012.08.066, DOI 10.1039/c4cp01948b and  
195 DOI 10.1039/c2cc31712e\*]. In addition, initial coulombic efficiency can be further  
196 increased by a layer of alumina deposited by Atomic Layer Deposition [\*please add ref.  
197 DOI 10.1021/ja205119g ].

198 A silicon thin film with the same thickness of  $3.5 \mu\text{m}$  was deposited onto a Cu current  
199 collector for comparison. Much lower capacities at various current densities and stability  
200 are observed in Figure 5(b). We believe CNT paper plays a very important role. First, CNT  
201 paper can contribute extra capacity as a current collector; secondly, CNT offers a pre-

202 formed porous substrate, leading to silicon deposition with increased mesoporosity, which  
203 can give enough space to tolerant volume expansion of silicon. Even under “stressful”  
204 conditions (current density of  $1080 \mu\text{A cm}^{-2}$ ), Si-CNT paper anodes behave in a very stable  
205 way, showing almost no capacity loss up to 1000 cycles (3% losses in the last 300 cycles),  
206 as shown in Figure 5(c). Although the area capacity drops dramatically when  $1080 \mu\text{A cm}^{-2}$   
207 is applied, for micro-batteries and wearable batteries power density is less critical for  
208 usage compared to the areal energy density and lifetime.

209 The charge/discharge curves of the first three cycles, measured between 5mV and  
210 1.5V versus Li/Li<sup>+</sup> at a current density of  $54\mu\text{A/cm}^2$ , are shown in Figure 5(d). The first  
211 lithiation curve shows a rapid drop of the potential with two plateaus: the first one, around  
212 0.8V, can be related to the SEI formation, which concurs to irreversible capacity losses;  
213 the second one, around 0.3V is attributed to lithiation of amorphous silicon. In delithiation  
214 profiles, the smooth change in slope around 0.5V can be ascribed to delithiation of  
215 amorphous  $\text{Li}_x\text{Si}$  to give amorphous Si[16]. From the 2nd cycle, no plateau is observed  
216 during lithiation, which is the typical voltage profile of Li insertion/extraction in  
217 amorphous structure materials without phase transformation. These observations are  
218 consistent with other works [17-19]. We believe the ~~synergistic~~-synergetic effect of  
219 nanoporous amorphous structure and flexible CNT paper contribute to the excellent cycling  
220 performance of silicon with the thickness over 1 $\mu\text{m}$ .

221

#### 222 4. Conclusions

223 Nanoporous amorphous Si films have been produced by PLD in hierarchical  
224 mesoporous morphology and on a CNT paper current collector. The mesoporosity and the

225 crucial role of CNT substrate allowing for volume expansion of Si during charge/discharge  
226 have been exploited, resulting in very promising electrochemical performances. Capacity  
227 retention and rate capability are dramatically improved with respect to Si on Cu current  
228 collector, revealing the enhanced capacity of  $\sim 1000\mu\text{Ah}/\text{cm}^2$  at  $54\mu\text{A}/\text{cm}^2$  and a good  
229 stability with no decay for at least 1000 cycles at  $1080\mu\text{A}/\text{cm}^2$ . Thanks to the combination  
230 of PLD and flexible CNT paper, it is possible to realize high-performance, flexible Si-CNT  
231 paper anodes for micro-batteries. The process-flow to obtain Si-CNT anodes, mainly based  
232 on PLD, can successfully minimize the required steps and reduce the involved temperatures.  
233 It allows to obtain free-standing anodes with minimized processing times/cost and to open  
234 the way to an up-scaled production of Si-based anodes for micro-batteries.

### 235 **Acknowledgement**

236 The work at Chongqing Normal University was supported by the Natural Science  
237 Foundation of China (No. 21301199), Chongqing Science and Technology Commission  
238 (cstc2014jcyjA50035), Chongqing Municipal Education Commission (KJ130601) and  
239 the Key Laboratory of Green Synthesis and Applications, College of Chemistry,  
240 Chongqing Normal University. This work at Wuhan ATMK Super EnerG Technologies  
241 Inc. is supported by the 3551 Recruitment Program of Global Experts by Wuhan East Lake  
242 Hi-Tech Development Zone, China. Erika Biserni would like to express her gratitude to  
243 Dr. Fabio Di Fonzo of the Centre for Nanoscience and Technology of IIT, Milan, for  
244 training and useful discussions on the pulsed laser deposition technique. Ming Xie and  
245 Erika Biserni contributed to this work equally.

246

### 247 **References**

248

- 249 [1] T. Jimbo, P. Kim and K. Suu, 2012 *2011 2nd International Conference on*  
250 *Advances in Energy Engineering (Icaee)* **14** 1574.
- 251 [2] A. Patil, V. Patil, D. W. Shin, J. W. Choi, D. S. Paik and S. J. Yoon, 2008 *Mater.*  
252 *Res. Bull.* **43** 1913.
- 253 [3] M. N. Obrovac and L. Christensen, 2004 *Electrochem. Solid St.* **7** A93.
- 254 [4] Y. H. Xu, G. P. Yin and P. J. Zuo, 2008 *Electrochim. Acta* **54** 341.
- 255 [5] Y. Kubota, M. C. S. Escano, H. Nakanishi and H. Kasai, 2007 *J. Appl. Phys.* **102**.
- 256 [6] M. N. Obrovac and L. J. Krause, 2007 *J. Electrochem. Soc.* **154** A103.
- 257 [7] D. E. Arreaga-Salas, A. K. Sra, K. Roodenko, Y. J. Chabal and C. L. Hinkle, 2012  
258 *J. Phys. Chem. C* **116** 9072.
- 259 [8] M. Y. Nie, D. P. Abraham, Y. J. Chen, A. Bose and B. L. Lucht, 2013 *J. Phys.*  
260 *Chem. C* **117** 13403.
- 261 [9] M. D. Fleischauer, J. Li and M. J. Brett, 2009 *J. Electrochem. Soc.* **156** A33.
- 262 [10] H. J. Jung, M. Park, Y. G. Yoon, G. B. Kim and S. K. Joo, 2003 *J. Power Sources*  
263 **115** 346.
- 264 [11] S. Ohara, J. Suzuki, K. Sekine and T. Takamura, 2004 *J. Power Sources* **136** 303.
- 265 [12] M. N. He, Q. Sa, G. Liu and Y. Wang, 2013 *Acs Appl. Mater. Inter.* **5** 11152.
- 266 [13] H. J. Jung, M. Park, S. H. Han, H. Lim and S. K. Joo, 2003 *Solid State Commun.*  
267 **125** 387.
- 268 [14] E. Biserni, M. Xie, R. Brescia, A. Scarpellini, M. Hashempour, P. Movahed, S.  
269 M. George, M. Bestetti, A. Li Bassi and P. Bruno, 2015 *J. Power Sources* **274**  
270 252.

- 271 [15] Z. Li, W. Li, Y. D. Jiang, H. H. Cai, Y. G. Gong and J. A. He, 2011 *J. Raman*  
272 *Spectrosc.* **42** 415.
- 273 [16] C. Pereira-Nabais, J. Swiatowska, A. Chagnes, A. Gohier, S. Zanna, A. Seyeux,  
274 P. Tran-Van, C. S. Cojocar, M. Cassir and P. Marcus, 2014 *J. Phys. Chem. C*  
275 **118** 2919.
- 276 [17] E. Biserni, N. Garino, A. Li Bassi, P. Bruno and C. Gerbaldi, 2014 *ECS*  
277 *Transactions* **62** 107.
- 278 [18] S. H. Ng, J. Wang, D. Wexler, S. Y. Chew and H. K. Liu, 2007 *J. Phys. Chem. C*  
279 **111** 11131.
- 280 [19] W. R. Liu, J. H. Wang, H. C. Wu, D. T. Shieh, M. H. Yang and N. L. Wu, 2005 *J.*  
281 *Electrochem. Soc.* **152** A1719.
- 282 [20] W. Wei, G. Oltean, C. W. Tai, K. Edstrom, F. Bjorefors and L. Nyholm, 2013 *J.*  
283 *Mater. Chem. A* **1** 8160.
- 284 [21] J. H. Zhu, J. Jiang, Y. M. Feng, G. X. Meng, H. Ding and X. T. Huang, 2013 *Acs*  
285 *Appl. Mater. Inter.* **5** 2634.
- 286 [22] W. Wang, M. Tian, A. Abdulagatov, S. M. George, Y. C. Lee and R. G. Yang,  
287 2012 *Nano Lett.* **12** 655.
- 288 [23] M. S. Park, G. X. Wang, H. K. Liu and S. X. Dou, 2006 *Electrochim. Acta* **51**  
289 5246.
- 290 [24] J. B. Kim, H. Y. Lee, K. S. Lee, S. H. Lim and S. M. Lee, 2003 *Electrochemistry*  
291 *Commun.* **5** 544.
- 292
- 293

294

295

296

297

298

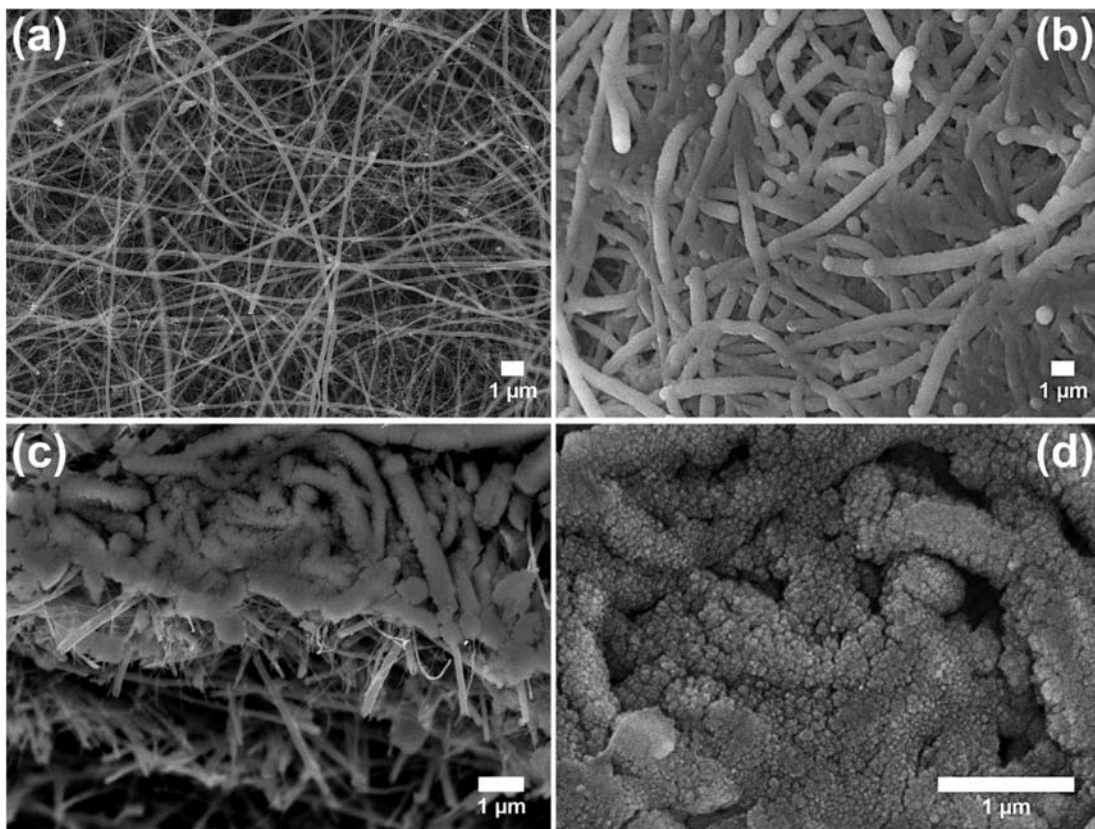
299

300

301

302

303

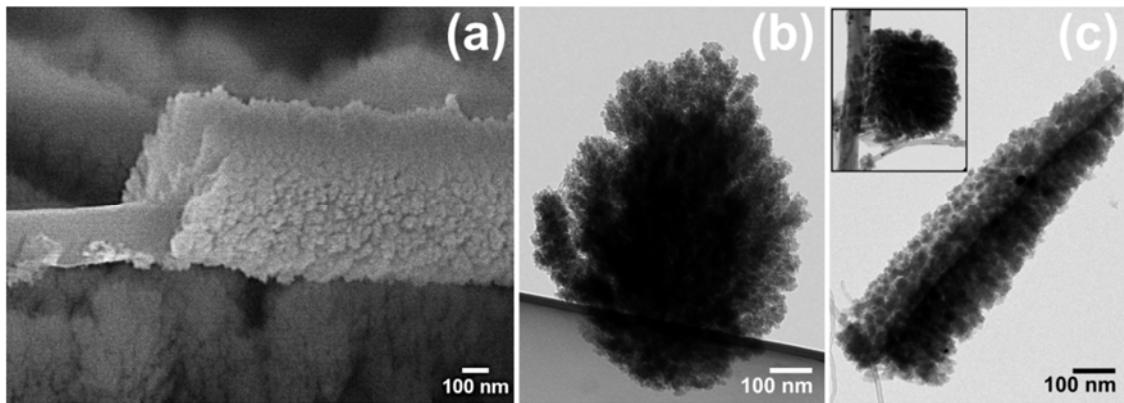


304

305 **Figure 1.** Top view SEM images of bare CNT paper (a) and Si-CNT (b). Cross-section  
306 images showing conformal covering of CNT by Si (c) and zoomed view of the Si  
307 morphology (d).

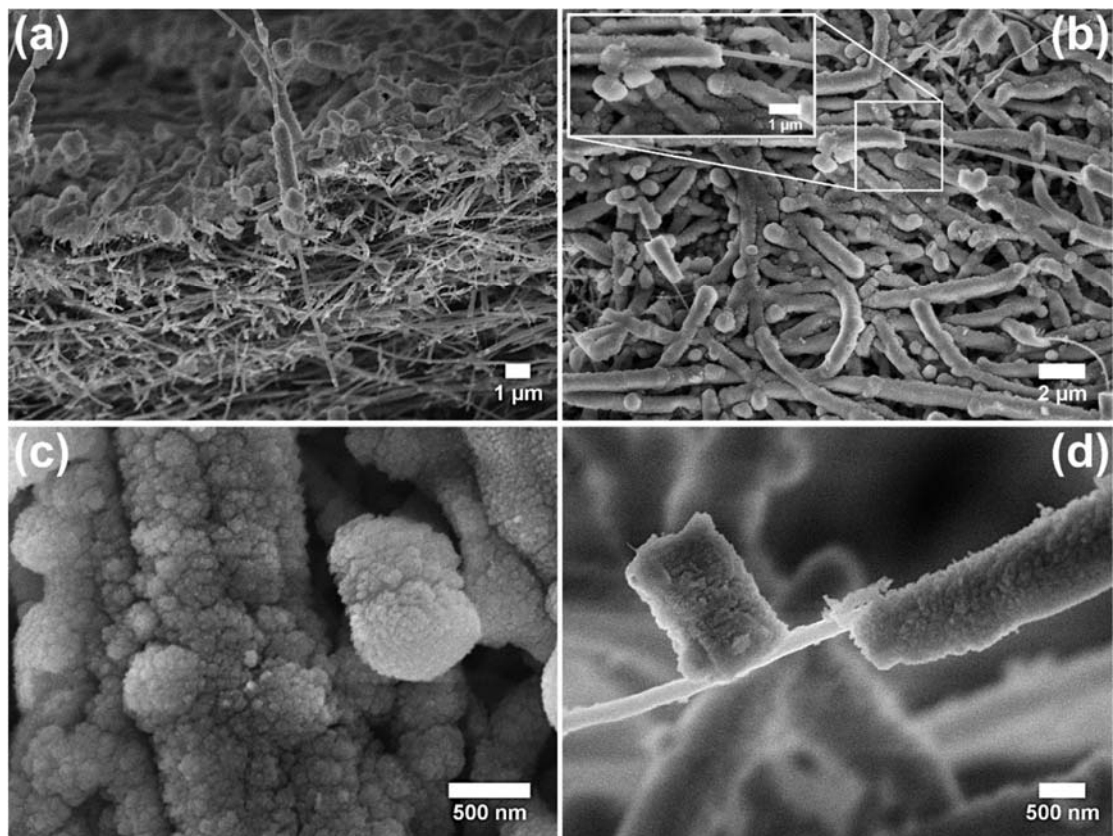
308

309  
310  
311  
312  
313  
314



315  
316  
317  
318  
319  
320  
321  
322  
323

**Figure 2.** SEM (a) and TEM (b and c) images of Si covering a carbon nanotube.



324

325

326

327

328

329

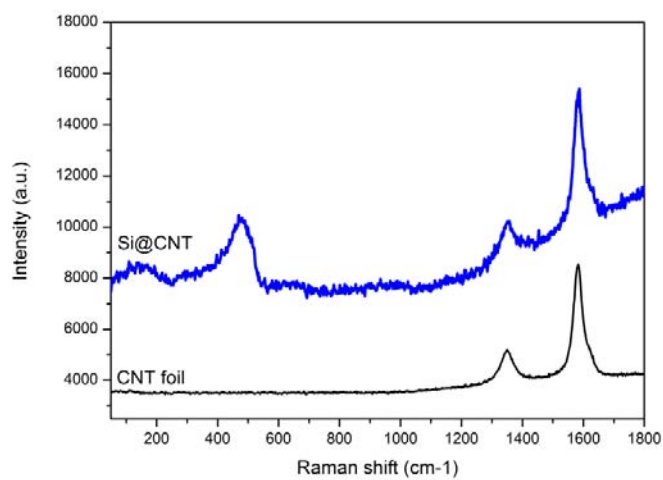
330

331

332

**Figure 3.** HRSEM images of Si-CNT after mechanical test. No visible damage is observed either in cross section (a) or plan view (b) The inset in (b) shows a zoomed view of the Si film covering a CNT. (c) Zoomed view of the silicon clustered film. (d) In some spots, silicon is detached from CNT.





333

334

335

**Figure 4.** Raman spectrum of a Si-CNT sample with bare CNT foil spectrum as a comparison.

336

337

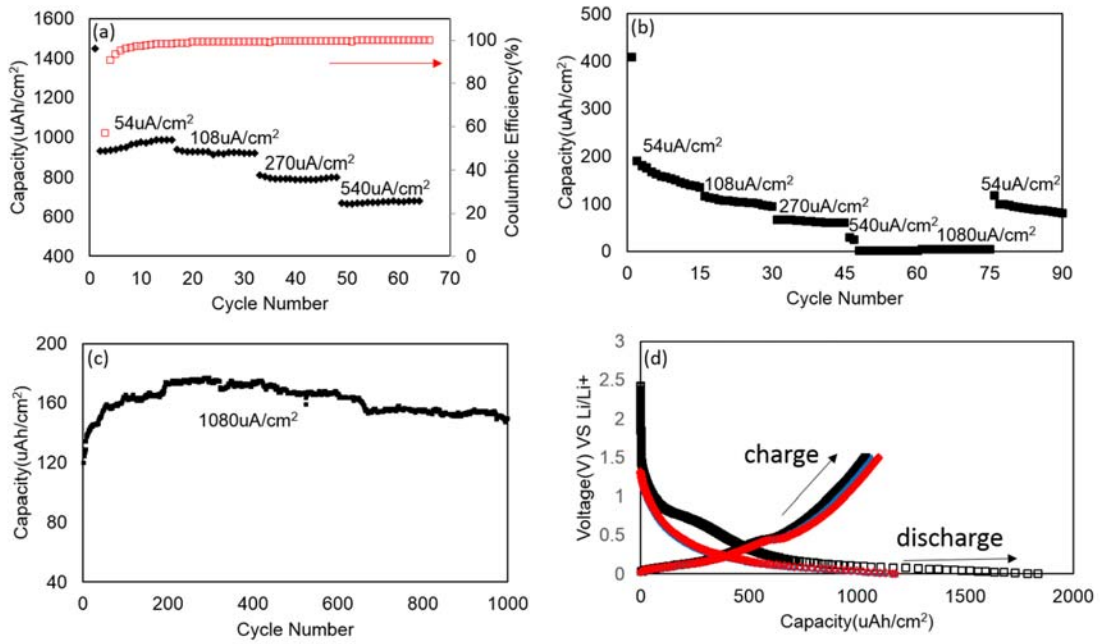
338

339

340

341

342



343

344

**Figure 5.** (a) Rate test and coulombic efficiency (b) Si film deposited onto Cu current collector under the same conditions (c) long life test, showing stable behavior up to 1000 cycles (d) Galvanostatic charge/discharge curves for the first three cycles.

345

346

347

348

349

350

351

352

353

354

355

356

357

358

359

360

361

362

Table1. Comparison of thin film battery anodes performance with this work

<b>Materials</b>	<b>Capacity(<math>\mu\text{Ah cm}^{-2}</math>)</b>	<b>Synthesis Technique</b>	<b>Reference</b>
Annealed TiO <sub>2</sub> nanotubes	460	two-step anodization followed by annealing	20
3D Ni/SnO <sub>x</sub> /C nanostructured arrays	470	hydrothermal method followed by a calcination-reduction process	21
3D Ni/TiO <sub>2</sub> nanowire network	170	Electrodeposition followed by atomic layer deposition	22
3 $\mu\text{m}$ Si on stainless steel	60	pulsed laser deposition	23
Fe/Si multi-layer thin film	130	electron-beam evaporation	24
CVD carbon coated nanoporous Si film on Cu	175	Pulsed laser deposition followed by chemical vapor deposition	14
<b>3.5 <math>\mu\text{m}</math> nanoporous Si film on CNT paper</b>	<b>1000</b>	<b>pulsed laser deposition</b>	<b>This work</b>