

Thermal annealing and exposure to divertor-like deuterium plasma of tailored tungsten oxide coatings

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Abstract

In this work we produced tungsten (W) and W oxide (WO_x) films by pulsed laser deposition (PLD) with the aim of the addressing modifications of structure and morphology that occur after annealing treatments and high-flux deuterium plasma. Thanks to the high flexibility of PLD we produced nanostructured W containing non-bounded oxygen, different types of WO_x and multilayered films. W coatings are dense, non-porous and exhibit a nanocrystalline structure, resembling the coatings used as first wall in Tokamaks. The oxide films are nearly stoichiometric amorphous WO_x (x = 3) with different morphology from compact to porous. Depending on annealing temperature, nucleation of different crystalline phases (e.g. WO₃, W₁₈O₄₉) occurs. Exposure of films to high-flux ($\sim 10^{24} \text{ m}^{-2} \text{ s}^{-1}$) deuterium plasmas in Magnum-PSI at different surface temperatures ($T_{\text{max}} = 580 \text{ K}$) determines material modifications at the nanoscale (e.g. nanometric defects) but no delamination. In addition preliminary deuterium retention results are reported.

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1. Introduction

Due to its low erosion yield, high melting temperature, high thermal conductivity, low hydrogen isotope permeation and inventory, tungsten (W) is a candidate material for plasma facing components (PFCs) in fusion reactors [1]. W is used for PFCs in operating tokamaks (e.g. ASDEX-Upgrade and JET-ILW) and will be used in future devices (e.g. ITER). Despite its suitable properties in tokamak environment, W undergoes several changes, due to neutronic and ionic fluence, temperature and sputtering processes, that produces modifications to bulk W leading to its damaging, erosion, re/co-deposition. In these conditions W shows properties and compositions which can be different from bulk W. It can also form different types of compound materials during re/co-deposition process in presence of buffer (e.g. Nitrogen)[2] and impurity gases (e.g. Oxygen, O₂) [3]. Since O₂ and oxygen compounds can be present as contaminations in tokamak vacuum vessel, oxidation can occur on PFCs, due to the high temperature reached by components during operation, and eroded material from PFCs can be re-deposited with a different stoichiometry [3,4]. While many studies on W erosion and deuterium (D₂) retention and permeation for different kinds of W (e.g. bulk and **W coatings**) are available [5–7], the behavior of tungsten oxide (WO_x) in a tokamak environment is not well known. WO_x has different properties from W, such as higher erosion yield [8] and different hydrogen isotope retention mechanisms [9–11]. In literature, there are only few works in which tungsten trioxide (WO₃) coatings have been exposed to D₂ plasma. In these works WO₃ has been obtained either by thermal annealing in air of bulk W [10,11] or by evaporation of amorphous WO₃ coatings later annealed in vacuum up to 870 K [12]. The produced oxide layers are monoclinic, crystalline with

thickness ranging from hundreds of nanometers to microns and limited capability of varying stoichiometry. After exposure to D₂ plasma, these coatings exhibit a larger amount of retained D₂ if compared with Plasma Spray and CMSII W coatings [13,14] and they show loss of O₂ from surface due to chemical reaction of O₂ with D₂ [15]. These coatings are quite different from oxide layers that could be present in a tokamak because native oxide layers are thinner than hundreds of nanometers and, in addition, an amorphous structure has to be expected. Moreover, also re-deposited WO_x compounds will probably have an amorphous structure with peculiar stoichiometry [16–19]. To make major advances in our understandings of these important issues, it is essential to produce, in a controlled way, coatings that are able to simulate native oxide layer, co-deposited WO_x coatings and mixed layer structures. These coatings should be tested in ITER-relevant tokamaks or in linear machines plasma generators able to reproduce conditions that will be present in ITER divertor.

The aim of this work is to produce and test WO_x coatings that are similar to the oxide layers in tokamaks. The capabilities of Pulsed Laser Deposition (PLD) are used to control the properties of the coatings on the nanoscale, i.e., thickness, morphology, structure and stoichiometry [20,21]. The coatings modifications due to annealing treatments are investigated. In addition, exposures to D plasma at divertor-like condition are performed in Magnum-PSI. Preliminary results on D₂ post-exposure retention are presented.

2. Experimental

Our PLD system exploits a nanosecond laser pulse 2nd harmonic of a Nd:YAG laser ($\lambda=532$ nm, pulse duration 5-7 ns, repetition rate 10 Hz) focused on a W target (purity 99.9%). The laser spot is about 9.2 mm², pulse energy is 815 mJ. The species ablated from the target expand in a proper vacuum chamber (base pressure of 10⁻³ Pa), where an O₂ background gas (purity 99.999%) is present. O₂ pressure ranges between 5 and 100 Pa. The expanding species are collected on a

grounded rotating substrate of Silicon (100) 2.5 mm x 2.5 mm or polycrystalline W substrate (PLANSEE, 99.96% purity, Ø 20 mm, 1 mm thick) containing micrometer-sized grains and mechanically polished until mirror finish. The substrate is 60 mm away from the target at room temperature. Further details are given in a previous work [20].

The films are characterized with a Zeiss Supra 40 field emission scanning electron microscope (SEM, accelerating voltage 5 kV). In order to check film composition, energy dispersive X-ray spectroscopy (EDXS) is performed using an accelerating voltage of 15 kV, to excite K_α and M_α electronic levels of Oxygen (O) and W, respectively. K_α of Cobalt is used for system calibration. Every EDXS analysis is repeated three times on equivalent points of the sample, the number of X-ray counts for each measurement being 250000. The inspected thickness range is around 1 μm (data obtained with CASINO simulation using WO_3 bulk density and 15 keV electrons), depending on structural properties of different samples.

Micro-Raman measurements are performed to detect the presence and stoichiometry of WO_x , with a Renishaw InVia spectrometer and using the 514.5 nm wavelength of an Ar^+ laser.

To better understand WO_x thermal behavior, some specific samples, deposited on W substrates, are annealed in a high-vacuum chamber (10^{-5} Pa) for 1 h using a resistive heater stage. A K-Thermocouple is used to monitor the stage temperature.

WO_x coatings, deposited on W substrates and not annealed, were exposed to high-flux deuterium plasmas in the linear plasma generator Magnum-PSI at FOM-DIFFER [22,23] at low surface temperature (maximum temperature in center of plasma beam of $T_{\text{max}} = 580$ K). The electron density and temperature of the plasma were measured with Thomson scattering [24]. The flux on the target can be estimated by using Bohm criterion [25,26], leading to a maximum flux in the center of the plasma beam of $10^{24} \text{ m}^2 \text{ s}^{-1}$. The targets are subject to 6 plasma shots of 25 s each,

which leads to an average plasma fluence of typically $1 \times 10^{26} \text{ m}^{-2}$. During exposure, the targets are kept at floating voltage.

3. Results and Discussions

3.1. WO_x films deposition

In order to obtain films with controlled morphology, structure and stoichiometry, a wide range of pressures (from 5 Pa to 100 Pa of O_2) is investigated. In figure 1 an EDXS analysis of the samples as a function of the O_2 pressure during deposition is shown. **The O/W ratio in films increases with raising O_2 pressure from vacuum (10^{-3} Pa) to 30 Pa and in the end it stabilise at around 3, independently of O_2 pressure during deposition.** In figure 2a Raman spectra of the samples deposited at various background O_2 pressure are reported. For films deposited in an O_2 atmosphere below 30 Pa no Raman signal is detected, revealing their metallic nature. Above 30 Pa of O_2 , Raman spectra show a low frequency band in the range $100\text{-}500 \text{ cm}^{-1}$, associated to O-W-O bending modes, a band in the range $600\text{-}900 \text{ cm}^{-1}$, associated to W-O stretching modes and a band at about 960 cm^{-1} attributed to W=O stretching modes at grain boundaries, therefore related to the presence of nano-grains [21]: this typical broad band feature in Raman spectra implies that the deposited oxide films are nearly stoichiometric amorphous tungsten oxide (a- WO_3) [21]. In these spectra a narrow peak at 521 cm^{-1} due to the first order scattering from silicon substrate is also evident: this confirms the optical transparency properties of these films due to their oxide nature. In figure 3 a SEM cross section analysis is reported. The film morphology changes, from compact to porous, varying O_2 pressure from 30 Pa to 100 Pa (figure 3a and 3b respectively). In addition, in figure 4a a plain view SEM image of the sample deposited at 30 Pa O_2 atmosphere is presented. Furthermore, thanks to the great versatility of PLD, multi-layer structures have also been deposited: in figure 3c, the dark layer on the top of a

nanostructured columnar W film (c-W) is a thin a-WO₃ layer (50 nm), while figure 3d shows a similar a-WO₃ film buried between two c-W layer.

Different types of W and WO₃ coatings, with an high thickness control, can be produced using PLD: in particular, the deposition of metallic W films with a controlled amount of not bonded O is achievable below a threshold pressure (30 Pa). It is feasible to obtain various morphologies of a-WO₃, from compact to porous, above 30 Pa, by tuning pressure during deposition. Finally, the oxide layer can be put in multi-layer structures, which can be useful for simulating different kinds of co-redeposited coatings that could be present on PFCs.

For the aims of this work, 30 Pa O₂ atmosphere is selected as reference pressure to produce compact a-WO₃ films, than can mimic WO_x co-redeposition present in tokamak environment, with a thickness of 1 μm, high uniformity and planarity for subsequent analysis.

3.2. Annealing

The expected working conditions in the ITER divertor involve temperatures ranging from about 400 K to 1150 K [27,28] and high vacuum ($\sim 10^{-3}$ Pa). To better understand a-WO₃ thermal behavior, films deposited at 30 Pa O₂ have been annealed in vacuum for 1 h at 570 K, 870 K and 1150 K. It is interesting to note that at 1150 K WO₃ starts to volatilize even in air [15]. The film deposited at room temperature and the one annealed at 570 K keep the same surface morphology (figure 4a) and structure (Raman spectrum (2) in figure 2b and Raman spectrum (30 Pa) in figure 2a). These results show that, below 570 K, a-WO₃ is not subjected to modifications induced by temperature. At higher temperature, modifications occur. In figure 4b a plain SEM view of a sample annealed at 870 K is reported. In this case, morphology modifications are evident, such as cracks that interrupt film continuity. Raman spectrum (3) in figure 2b presents narrow peaks at 135 cm⁻¹, 270 cm⁻¹, 721 cm⁻¹ and 812 cm⁻¹, showing that, at this temperature, a-WO₃ crystallizes as monoclinic WO₃ [29,30]. In figure 4c SEM plain view of a film annealed at 1150 K is

reported; an evident morphological modification occurs with nano-rods and nano-sheets growing on the surface sample. These morphological features, together with the very complex peaks pattern reported in the Raman spectrum (4) in figure 3b, attest the formation of the Magneli phase, that is a sub-stoichiometric phase of WO_3 , defined as $\text{W}_{18}\text{O}_{49}$ (or $\text{WO}_{2.75}$) [29,31]: the decrease in the O content is caused by the annealing treatment at 1150 K in vacuum.

3.3. *Magnum-PSI exposure*

In figure 4d, a SEM plain view of the sample deposited at 30 Pa O_2 and exposed to D plasma at 580 K films is reported. Surface modifications occur, such as nano-cracks and point-like features. The Raman signal dramatically decreases (spectrum (1) in figure 2b) pointing out loss of O from the film, probably due to the reduction reaction that occurs between WO_x and D during plasma exposure. It is known that reduction of WO_3 by H_2 starts at around 850 K [15]; in the Magnum-PSI plasma, D_2 is present in dissociated and ionized form (D) and this probably allows the reduction starting at lower temperatures. In the exposed sample, the O/W ratio, as obtained with EDXS, is about 2.5, confirming the O loss. Thermal desorption spectroscopy (TDS) has been performed to investigate the retention of deuterium in the thin films. The TDS signal of D_2 for the 1 μm a- WO_3 film exposed to D plasma is displayed in figure 5, together with the TDS signal for the 1 μm c-W film exposed at similar conditions. The spectrum of the a- WO_3 film shows the presence of two desorption peaks at 500 K and 780 K, while the c-W film shows a broad TDS spectrum with its maximum at about 580 K and a shoulder at higher temperatures. The small peak at ~ 700 K is most likely caused by an experimental artifact during the heating procedure. The total number of D atoms desorbed from the a- WO_3 film is $6.1 \times 10^{19} \text{ D m}^{-2}$, which is a factor 20 lower than $1.3 \times 10^{21} \text{ m}^{-2}$ for the c-W film. These total numbers include the deuterium desorbing as HD **but not the contribution due to heavy waters (HDO and D_2O) desorption.** **For this reason values reported can be affected to an underestimation of total D retained.**

Nevertheless, it should be noted that whereas the total deuterium desorption for the c-W is dominated by D₂, for the oxide film the D₂ signal is a factor of 1.4 smaller than the HD signal. Moreover D retained in a-WO₃ coating is 2 times lower than D retained in monoclinic thermally grown WO₃ [10].

4. Conclusions

WO_x films were deposited using PLD technique **in order to simulate re-deposited compound layers that can be present onto PFCs**. It is shown how films morphology and stoichiometry can be controlled changing background O₂ during deposition. Two different growth regimes are determined with SEM and Raman analysis. In particular, below 30 Pa O₂ metallic W films are deposited, with different concentrations of not bonded O depending on background deposition pressure; above 30 Pa, a-WO₃ films with different morphologies, from compact to porous, are obtained. Compact a-WO₃ films have been annealed in vacuum at different temperatures to investigate their thermal behavior: at 570 K no changes are observed; at 870 K monoclinic phase of WO₃ is formed; at 1150 K Magneli phase (W₁₈O₄₉) is obtained. 1 μm compact a-WO₃ film and 1 μm c-W film are exposed to high-flux D plasma in Magnum PSI at low temperature. After exposure, a-WO₃ exhibits changes, such as loss in O content and nano-defects. Comparison between D₂ retention in c-W and a-WO₃ highlights that retention in oxide film is 20 times lower than that in c-W.

In the future, different oxide coatings will be tested in plasma simulators (e.g. Magnum PSI) or in tokamaks, to better understand their retention properties.

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Figure captions

Figure 1. Elemental composition: Oxygen on Tungsten ratio (O/W) calculated using EDXS versus deposition background O₂ pressure.

Figure 2. (a) Raman spectra of the films deposited on Silicon substrates varying background O₂ pressure (star marker indicates Silicon peak at 521 cm⁻¹); (b) Raman spectra of the film deposited on W substrate at 30 Pa O₂ background pressure annealed at 570 K (2), 870 K (3), 1150 K (4) and exposed at 580 K D plasma (1).

Figure 3. SEM cross section analysis of a-WO₃ films deposited varying O₂ deposition pressure (a)-(b) and multi-layer films (W + WO_x) (c)-(d).

Figure 4. SEM plain view of the a-WO₃ films deposited in 30 Pa O₂ background atmosphere: at room temperature and annealed at 570 K (a), annealed at 870 K (b) at 1150 K (c) and exposed to 580 K D₂ plasma in Magnum-PSI (d).

Figure 5. TDS Spectrum of 1 μm a-WO₃ and of 1 μm c-W, deposited in 30 Pa O₂ background pressure and vacuum (10⁻³ Pa) respectively, exposed to 580 K D plasma (signal related to mass 4, D₂).

Figures

Figure 1.

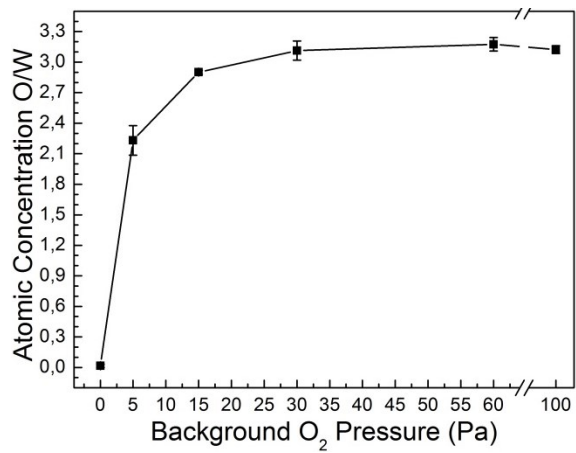


Figure 2.

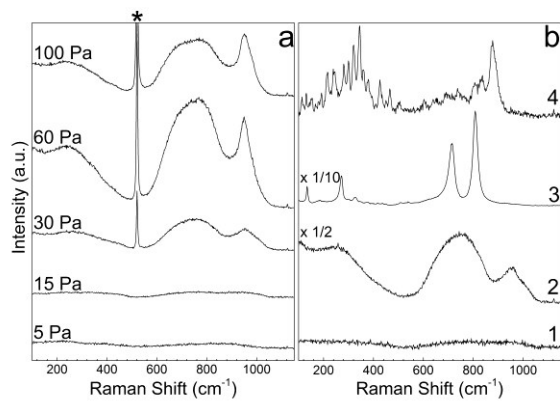


Figure 3.

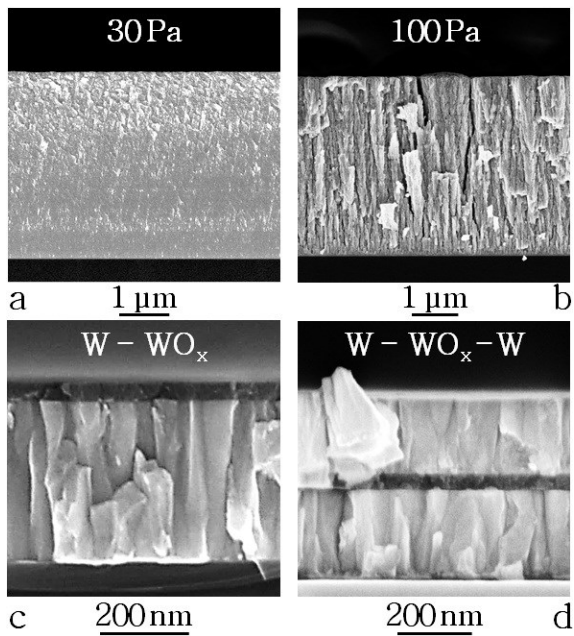


Figure 4.

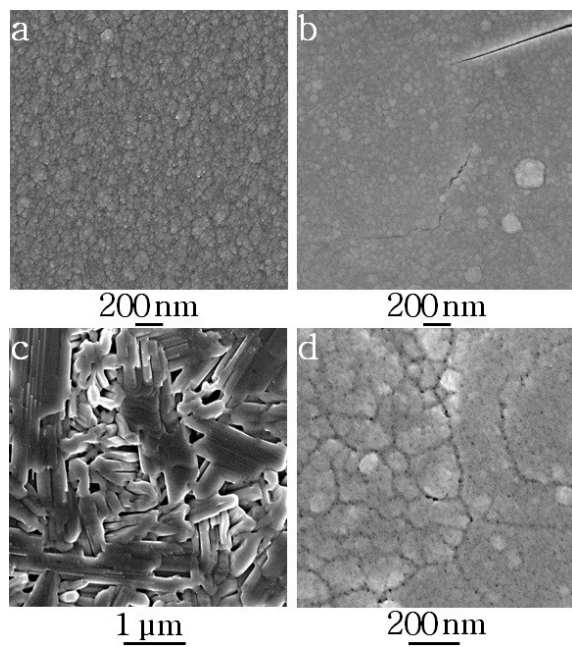


Figure 5.

