

This document is confidential and is proprietary to the American Chemical Society and its authors. Do not copy or disclose without written permission. If you have received this item in error, notify the sender and delete all copies.

Tuning of Electrical and Optical Properties of Highly Conducting and Transparent Ta-Doped TiO₂ Polycrystalline Films

Journal:	<i>The Journal of Physical Chemistry</i>
Manuscript ID:	jp-2014-126156.R1
Manuscript Type:	Article
Date Submitted by the Author:	05-Mar-2015
Complete List of Authors:	Mazzolini, Piero; Politecnico di Milano, ; Center for Nano Science and Technology-IIT@PoliMI, Gondoni, Paolo; Politecnico di Milano, Russo, Valeria; Politecnico di Milano, Chrastina, Daniel; L-NESS Politecnico di Milano, Physics Casari, Carlo Spartaco; Politecnico di Milano, ; Center for Nano Science and Technology-IIT@PoliMI, Li Bassi, Andrea; Politecnico di Milano, ; Center for Nano Science and Technology-IIT@PoliMI,

SCHOLARONE™
Manuscripts

Tuning of Electrical and Optical Properties of Highly Conducting and Transparent Ta-Doped TiO₂ Polycrystalline Films

P. Mazzolini,^{1,2} P. Gondoni,¹ V. Russo,¹ D. Chrastina,³ C.S. Casari,^{1,2} A. Li Bassi^{1,2}*

¹ Department of Energy - Politecnico di Milano, via Ponzio 34/3, I-20133 Milano, Italy

² Center for Nano Science and Technology - IIT@PoliMI, Via Pascoli 70/3, I-20133 Milano, Italy

³ L-NESS, Dipartimento di Fisica, Politecnico di Milano, Via Anzani 42, I-22100 Como, Italy

* To whom correspondence should be addressed. E-mail: andrea.libassi@polimi.it; tel. +390223996316

Keywords: Transparent Conducting Oxides, Pulsed Laser Deposition, transport properties, anatase, Moss-Burstein effect, electron effective mass.

Abstract

We present a detailed study on polycrystalline transparent conducting Ta-doped TiO₂ films, obtained by room temperature pulsed laser deposition followed by an annealing treatment at 550°C in vacuum. The effect of Ta as a dopant element and of different synthesis conditions are explored in order to assess the relationship between material structure and functional properties, i.e. electrical conductivity and optical transparency. We show that for the doped samples it is possible to achieve low resistivity

(of the order of $5 \times 10^{-4} \Omega\text{cm}$) coupled with transmittance values exceeding 80% in the visible range, showing the potential of polycrystalline Ta:TiO₂ for application as a transparent electrode in novel photovoltaic devices. The presence of trends in the structural (crystalline domain size, anatase cell parameters), electrical (resistivity, charge carrier density and mobility) and optical (transmittance, optical band gap, effective mass) properties as a function of the oxygen background pressures and laser fluence used during the deposition process and of the annealing atmosphere is discussed, and points towards a complex defect chemistry ruling the material behavior. The large mobility values obtained in this work for Ta:TiO₂ polycrystalline films (up to $13 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$) could represent a definitive advantage with respect to the more studied Nb-doped TiO₂.

1. Introduction

Transparent conductive oxides (TCOs) play a crucial role in the technological evolution of new generation energy conversion devices, such as organic, hybrid or dye sensitized -perovskite solar cells.^{1,2} Their development requires new functional properties in order to improve efficiency and durability;³ in this direction, a high chemical stability and a proper band alignment throughout the material interfaces of the devices is highly desirable for a TCO.⁴⁻⁷

One material that could fit these requirements is titanium oxide. Because of its transparency to visible light, high chemical stability in reducing atmosphere and proper band alignment with respect to the active components of the device, TiO₂ is the most employed material as blocking layer and photoanode in dye sensitized solar cells (DSSCs).⁸ As a wide band gap oxide, anatase TiO₂ ($E_g = 3.2 - 3.4 \text{ eV}^{9,10}$) is not an intrinsically good electrical conductor. Interestingly it has been recently discovered that replacing a certain amount of Ti atoms in the titanium oxide tetragonal cell with group V elements, such as Nb or Ta, can strongly reduce the resistivity down to values of the order of $2 \times 10^{-4} \Omega\text{cm}$ for epitaxially grown thin films on SrTiO₃ substrates.^{11,12} Despite the lack of knowledge with respect to the

1
2
3 most studied and widely used TCOs like $\text{In}_2\text{O}_3:\text{Sn}$, $\text{SnO}_2:\text{F}$ and $\text{ZnO}:\text{Al}$, this new class of d-electron
4 based TCOs seems to have good potentialities.^{13,14} In particular, Ta is thought to have definite
5 advantages over Nb in doping TiO_2 for TCO purposes, mainly because of a considerably higher
6 solubility and lower theoretically predicted electron effective mass, which could in principle result in a
7 higher electron mobility.¹⁵ Despite these considerations $\text{Ta}:\text{TiO}_2$ (TaTO) is considerably less studied
8 than $\text{Nb}:\text{TiO}_2$ (NTO).^{11,16-27} In particular only a few works have been devoted to the control of
9 electrical/optical properties of TaTO polycrystalline thin films on cheap substrates (e.g. soda-lime
10 glass),²⁸⁻³⁰ which is a crucial requirement in view of a possible application in real devices, such as
11 DSSCs. To the best of our knowledge, the lowest resistivity values at room temperature obtained so far
12 for polycrystalline NTO and TaTO thin films are respectively $4.6 \times 10^{-4} \Omega\text{cm}$ and $8.7 \times 10^{-4} \Omega\text{cm}$.^{17,30} In
13 order to obtain highly conductive polycrystalline TiO_2 -based TCOs, it is necessary to use a two-step
14 approach, which involves the deposition of amorphous films (e.g. by pulsed laser deposition or
15 sputtering) followed by a post-deposition treatment in reducing atmosphere (vacuum or H_2). A
16 reducing deposition atmosphere is a basic requirement also for the synthesis of highly conductive
17 epitaxial films. However, the defect chemistry and its role in determining the functional properties are
18 still open issues for this new class of TCOs.^{27-29,31-34}

19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42 The typical limitations in polycrystalline TiO_2 -based TCOs are connected to lower obtainable mobility
43 values with respect to the epitaxially grown films. In fact, for an optimal extrinsic doping level (5% and
44 6% for Ta and Nb respectively^{12,35}), it is usually possible to obtain large charge carrier densities
45 (usually in the range of 5 to $15 \times 10^{20} \text{ cm}^{-3}$) for both epitaxial and polycrystalline thin films, while the
46 mobility value is more than halved due to the presence of different domain orientations (from values in
47 the range of $20 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$ for epitaxially grown films, down to values in the range of $5 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$ for
48 randomly oriented polycrystalline films).^{11,12,17-19,22,25,28,30,36-40}

1
2
3 One of the most suitable and versatile synthesis techniques in order to finely control the
4 structure/property relation in doped oxide thin films is pulsed laser deposition (PLD).⁴¹ In this work we
5 systematically explore the optimization of both the electronic transport and optical transparency of
6 polycrystalline TaTO thin films deposited via PLD, in order to unveil the real potentialities of TaTO,
7 with particular interest on the possible advantages with respect to the better explored NTO.¹⁵

8
9
10 TaTO films on soda-lime glass substrates have been obtained at room temperature by tuning the
11 oxygen background pressure and the laser fluence during the ablation process, and were then annealed
12 in different atmospheres (air and vacuum). We were able to obtain extremely low resistivity values of
13 the order of $5 \times 10^{-4} \Omega\text{cm}$ and transmittance in the visible region above 80%. Hall effect and absorbance
14 measurements were used to evaluate the electron conduction band effective mass. The obtained
15 electrical and optical properties are discussed in light of the material structural features, which can be
16 adjusted by tuning the different synthesis parameters (oxygen background pressure during the
17 deposition process, laser fluence, annealing atmosphere). The high electron mobility data are analyzed
18 in comparison to the most investigated Nb-doped TiO_2 .

37 38 **2. Experimental Section**

39
40 TaTO (Ta = 5% at.) and TiO_2 thin films were grown by PLD on soda-lime glass and Si(100) substrates
41 kept at room temperature, by ablating respectively $\text{Ta}_2\text{O}_5:\text{TiO}_2$ (molar ratio 0.025:0.975) and TiO_2 solid
42 targets with a ns-pulsed laser (Nd:YAG 4th harmonic, $\lambda = 266 \text{ nm}$, repetition rate $f_p = 10 \text{ Hz}$, pulse
43 duration $\sim 6 \text{ ns}$). The energy density was varied from 0.9 to 1.5 J/cm^2 by changing the focused area on
44 the surface of the solid target (laser pulse energy measured before the focusing lens was kept at 75 mJ),
45 while the target-to-substrate distance was fixed at 50 mm. Where not otherwise specified we will refer
46 in the following to a fluence of 0.9 J/cm^2 . All the depositions were made in oxygen background
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 pressure; TaTO films were deposited in the range 0.9 - 2.25 Pa, while TiO₂ films were deposited at 1
4
5 and 1.25 Pa.
6
7

8
9 All the as-deposited TiO₂-based thin films were amorphous. In order to obtain single phase
10
11 polycrystalline anatase, we exploited a post deposition annealing treatment in vacuum atmosphere ($p <$
12
13 4×10^{-5} Pa) in a home-made furnace at 550°C (10°C/min ramp, 1 hour dwell) and in air at 550°C in a
14
15 Lenton muffle furnace (8°C/min ramp, 1 hour dwell).
16
17

18
19 Film thickness was evaluated by means of scanning electron microscopy (Zeiss SUPRA 40 field-
20
21 emission SEM) on samples grown on silicon.
22
23

24
25 The crystalline structure has been determined by micro-Raman measurements (Renishaw In Via
26
27 spectrometer with Ar⁺ laser, $\lambda = 514.5$ nm – power on sample 1 mW) and x-ray diffraction
28
29 (PANalytical X'Pert PRO MRD high-resolution X-ray diffractometer, using CuK α 1 radiation ($\lambda =$
30
31 0.15406 nm) selected by a two-bounce Ge monochromator). XRD measurements were performed in
32
33 both θ -2 θ and grazing incident angle configuration (fixed incident angle $\omega = 5^\circ$). The lattice parameters
34
35 and the mean crystalline domain size were determined by evaluating peak positions and full width at
36
37 half maximum in grazing incidence angle configuration, since grazing scans gave more intensity for the
38
39 analyzed peaks. The thin film surfaces were investigated by means of optical microscopy (Leitz
40
41 orthoplan-pol).
42
43
44

45
46 The electrical characterization was performed in the 4-point probe configuration with a Keithley K2400
47
48 Source/Measure Unit as a current generator (from 100 nA to 10 mA), an Agilent 34970A voltage meter
49
50 and a 0.57 T Ecopia permanent magnet.
51
52
53

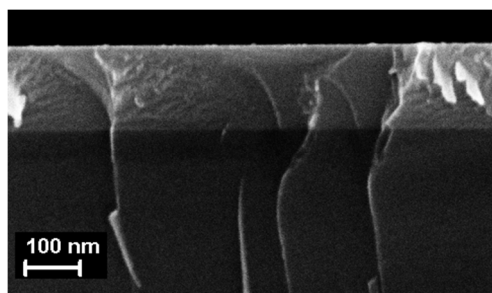
54
55 Optical transmittance and reflectance spectra (in the range 250 - 2000 nm) were evaluated with a UV-
56
57 vis-NIR PerkinElmer Lambda 1050 spectrophotometer with a 150 mm diameter integrating sphere. All
58
59
60

1
2
3 the acquired spectra were normalized with respect to the glass substrate contribution by setting to 1 the
4
5 intensity at the glass/film interface.
6
7

8 9 **3. Results**

10 11 *3.1. Structural properties*

12
13
14 TaTO and TiO₂ SEM micrographs show a similar ‘compact’ morphology for all the samples (a TaTO
15
16 micrograph is shown in Figure 1). All the TaTO thin films grown at different O₂ pressure are
17
18 characterized by the same thickness of 150 ± 5 nm, while the TiO₂ ones are 250 ± 5 nm and 150 ± 5 nm
19
20 thick (for 0.9 and 1.15 J/cm² laser fluence respectively).
21
22
23
24



36 **Figure 1.** SEM micrograph of 150 nm thick vacuum annealed TaTO (1.25 Pa oxygen background deposition pressure) on Si substrate.

37
38 Raman spectroscopy was employed to analyze the structure of the films. All the acquired spectra for
39
40 the room temperature deposited samples are similar, showing broad bands typical of amorphous /
41
42 highly disordered titanium oxide (black line in Figure 2). On the other hand, five distinct peaks related
43
44 to the expected anatase active modes are present in the Raman spectra of all annealed films, regardless
45
46 of the annealing atmosphere. In Figure 2 (red line) it is possible to identify the three E_g modes at 144
47
48 cm⁻¹, 197 cm⁻¹ and 639 cm⁻¹, the B_{1g} mode at 399 cm⁻¹ and the peak at 519 cm⁻¹, due to superposition
49
50 of the remaining A_{1g} and B_{1g} modes.^{42,43} We may conclude that both the annealing processes in air and
51
52 vacuum atmosphere (T = 550°C) are efficient to crystallize the TiO₂-based films in the anatase phase.
53
54
55
56
57
58
59
60

Moreover for all the acquired spectra there is no evidence of rutile Raman modes, excluding the formation of this phase under the used annealing conditions.

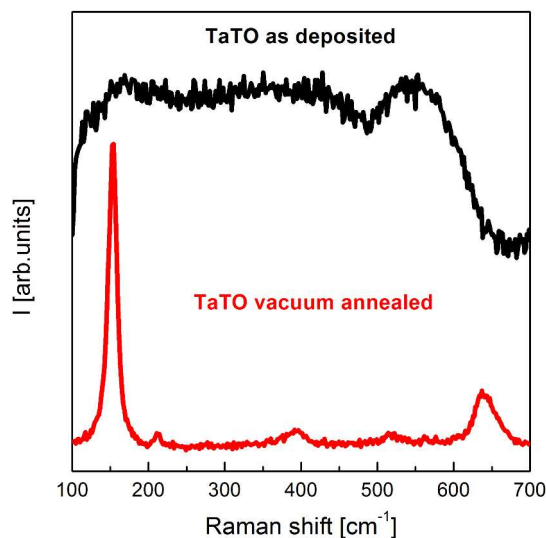


Figure 2. Raman spectra of as-deposited (black line) and vacuum annealed (red line) TaTO thin films deposited at 1.25 Pa oxygen background pressure.

The presence of anatase crystalline phase only for all the annealed films is confirmed by XRD measurements. As an example, selected XRD diffraction patterns for vacuum annealed samples deposited at two different oxygen background pressures (1.25 and 2 Pa) are shown in Figure 3 (a), together with the XRD pattern of an undoped TiO₂ film. No rutile peaks are observed, as well as tantalum oxide (Ta₂O₅) or metallic tantalum segregations. The crystalline quality of the vacuum annealed TaTO films seems to be dependent on the oxygen background pressure during the deposition process. In fact we are able to identify a narrow window of deposition pressures between 1.25 and 1.75 Pa in which both the most intense diffraction peaks of the anatase phase ((101) and (200)) are observed in θ -2 θ configuration (see e.g. red line in Figure 3 (a)), while at lower (1 Pa) and higher (2 – 2.5 Pa) deposition pressures a significant decrease of the (101) anatase reflection intensity is detected (see e.g. purple line in Figure 3 (a)).

Moreover, for the most crystalline samples deposited in the above mentioned pressure window (1.25 – 1.75 Pa), the introduction of Ta atoms in the crystal lattice seems to increase the intensity of the (101) peak and simultaneously decrease the (200) peak with respect to undoped TiO_2 , in the θ - 2θ configuration (see e.g. red line and blue line in Figure 3 (a)). For undoped titanium oxide it is known that, since the $\langle 101 \rangle$ surface has the lowest surface energy,⁴⁴ anatase thin films grown on amorphous substrates can result in either randomly oriented or $\langle 101 \rangle$ preferred-oriented polycrystalline forms.⁴⁵ In our case, the variation in the intensity ratio of anatase x-ray reflections could be related to an effect of doping or defect-induced lattice distortion or disordering (see Section 4 - Discussion).

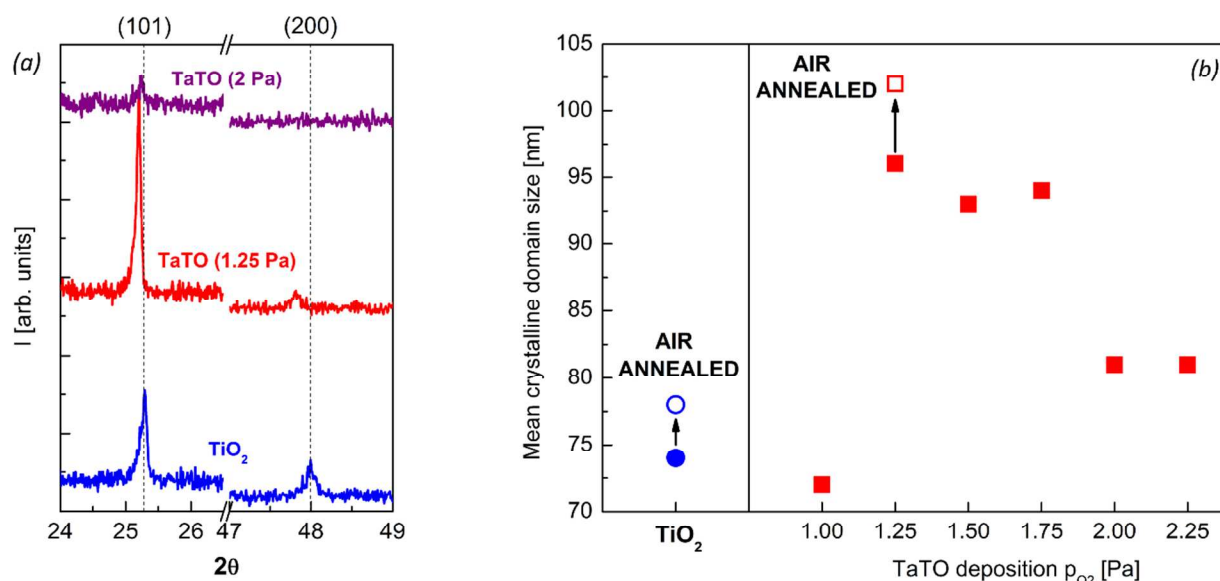


Figure 3. (a) XRD diffraction patterns of vacuum annealed TiO_2 (blue line) and TaTO thin films deposited at different oxygen background pressures (1.25 – 2 Pa respectively red and purple line). (b) Mean crystalline domain size evaluated from XRD of TiO_2 (blue spot on the left side) and TaTO samples deposited at different oxygen partial pressure (red squares), vacuum (filled) and air (empty) annealed at 550°C.

Furthermore, comparing XRD peak positions for vacuum annealed undoped and Ta doped TiO_2 samples, it is possible to identify the presence of a shift to smaller angles for the doped samples (as visible in Figure 3 (a)). This could be partially ascribable to a lattice expansion due to Ta incorporation

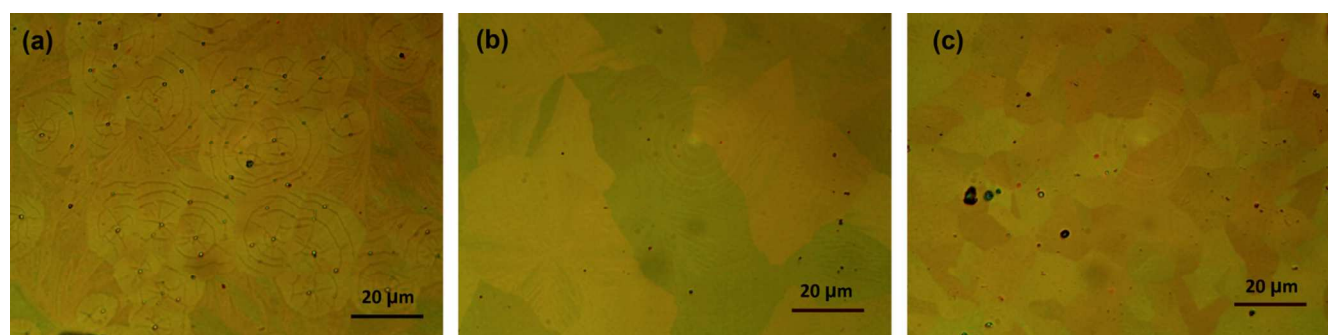
1
2
3 in substitutional Ti sites, since the ionic size of Ta⁺⁵ (0.064 nm) is slightly larger than that of Ti⁺⁴
4 (0.061 nm).^{12,46} Nevertheless the shift is also dependent on the oxygen background pressure used in the
5
6 deposition process; in particular the shift reduces as the oxygen deposition pressure is increased. This
7
8 will be further discussed in Section 4 (Discussion).
9
10

11
12
13 In grazing incident configuration (fixed incidence angle $\omega = 5^\circ$), both (101) and (200) diffraction peaks
14 are detected for all the range of TaTO deposition pressures, with a stronger intensity with respect to θ -
15
16 2θ measurements (not shown), and were employed to evaluate the mean crystalline domain size
17
18 according to Scherrer equation (Figure 3 (b)). The above mentioned narrow window of deposition
19
20 pressures (between 1.25 and 1.75 Pa) is confirmed to be the one with the higher diffraction peaks
21
22 intensity also in grazing incident configuration.
23
24
25
26
27

28
29 Domain size values for the vacuum annealed samples are reported in Figure 3 (b) (filled symbols),
30 showing larger values for deposition pressures in the range 1.25 – 1.75 Pa, thus confirming the
31
32 discussed ‘good crystallinity’ window. We also observe that the domain size for Ta-doped films is
33
34 larger or comparable with that found for undoped TiO₂. The same analysis shows that a post
35
36 crystallization process in the presence of oxygen (annealing in air) increases the mean domain size with
37
38 respect to the vacuum annealing process (domain size respectively 78 and 102 nm for TiO₂ and TaTO
39
40 deposited at an oxygen background pressure of 1.25 Pa, empty symbols in Figure 3 (b)).
41
42
43
44
45

46
47 While the Scherrer equation provides an evaluation of the mean crystalline domain size in the almost-
48
49 vertical direction (which is basically limited by the film thickness), an optical microscopy analysis of
50
51 the thin film surface of vacuum annealed TaTO samples deposited at different oxygen pressures allows
52
53 the evaluation of the grain size in the horizontal plane (Figure 4). Circular macro-cracks are only
54
55 observed on the surface of TaTO deposited at 1 Pa (Figure 4 (a)), while for higher deposition pressures
56
57 the presence of macro-grains is clearly visible, as already observed by Pore *et al.*⁴⁷ for the post
58
59
60

1
2
3 crystallization from amorphous phase of Nb- and Ta-doped TiO₂ films; in particular the macro-grain
4 dimension is following the trend obtained from the XRD analysis (Figure 3 (b)): in the narrow window
5 of deposition pressures for which the XRD crystalline domain size is larger, the grain size is of the
6 order of 25-30 μm (1.25-1.5-1.75 Pa, Figure 4 (b)), while for higher deposition pressures (2-2.25 Pa)
7 the typical grain size decreases down to 10-15 μm (Figure 4 (c)). This evidences are providing an
8 indication of the good crystallinity of the films.
9
10
11
12
13
14
15
16
17



29
30
31 **Figure 4.** Surface images captured using polarized light through optical microscope of vacuum annealed TaTO films deposited at 1 Pa
32 (a), 1.5 Pa (b) and 2.25 Pa (c).
33
34

35 3.2. Electrical properties

36

37
38 The as-deposited films have a typical resistivity value in the order of 10 Ωcm without showing any
39 particular tendency with respect to the presence of doping or different oxygen deposition pressure. For
40 such measurements it was not possible to discern the contributions of charge carrier concentration and
41 charge mobility because of the highly scattered values obtained in Hall measurements.
42
43
44
45
46
47

48
49 The post-deposition crystallization process in vacuum drastically decreases the resistivity values for all
50 the samples, as reported in Figure 5 where we show resistivity for undoped TiO₂ and TaTO (1.25 Pa
51 O₂), as-grown and after annealing. Undoped TiO₂ exhibits a drop of nearly 3 orders of magnitude,
52 reaching a resistivity value comparable with the best results obtained for PLD epitaxially grown
53 films.⁴⁸ The TaTO film with the best electrical properties (grown at 1.25 Pa O₂) showed a resistivity
54
55
56
57
58
59
60

value of $\rho = 5.73 \times 10^{-4} \Omega\text{cm}$ (see Figure 5) and sheet resistance $R_S = 38 \Omega/\square$; to the best of our knowledge, the resistivity value for the TaTO thin film is the lowest reported in literature for a tantalum-doped TiO_2 polycrystalline film and in line with the best results obtained with the most investigated NTO.^{17,28,30,49}

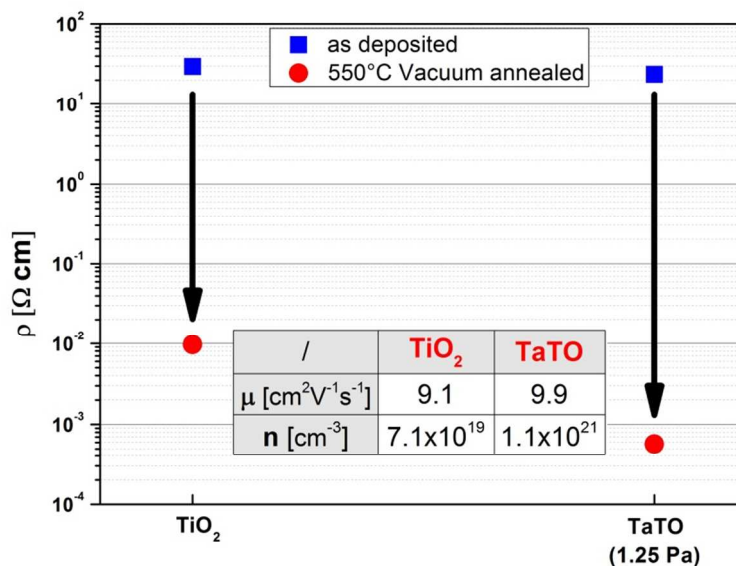


Figure 5. Effect on resistivity of the crystallization process exploited in vacuum atmosphere for TiO_2 and TaTO (deposited in an oxygen background pressure of 1.25 Pa). The resistivity values for the as deposited films are represented with blue squares, while the crystallized ones with red circles. In the inset table are reported mobility and electron density of the vacuum annealed samples.

The data reported in the inset of Figure 5 show that the difference between the electrical properties of vacuum annealed doped and undoped thin films mainly resides in the charge carrier density, which for TaTO is more than one order of magnitude higher, while the measured mobility is comparable. Still it is interesting to underline the possibility to obtain such a high concentration of free electrons in undoped TiO_2 : due to the absence of an extrinsic dopant, this evidence should be strictly related to the presence in the anatase cell of a large amount of oxygen vacancies, which are known to effectively act as electron donors.⁵⁰

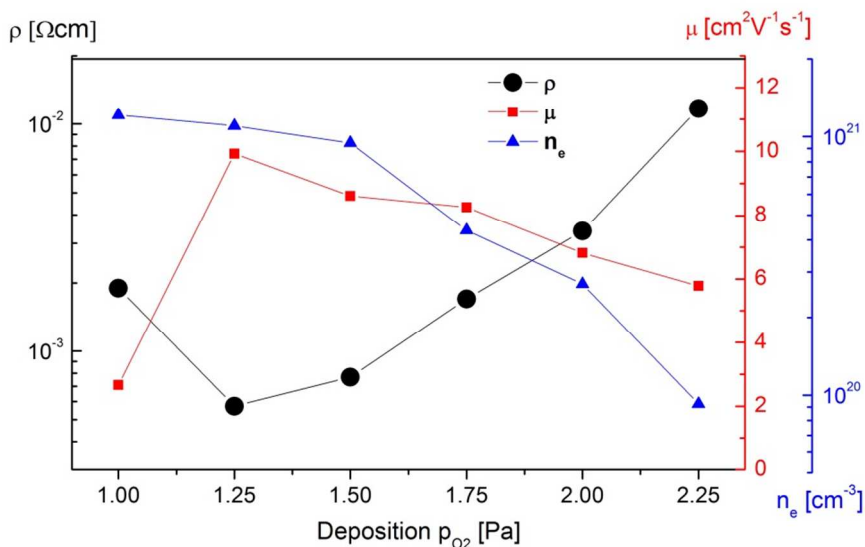


Figure 6. Electrical properties of polycrystalline TaTO thin films (150 nm) deposited in different oxygen partial pressures and annealed in vacuum atmosphere (550°C). The black spots represent the resistivity, the blue triangles represent the electron density, and the red squares represent the electron mobility evaluated via Hall measurements.

Furthermore, for TaTO vacuum annealed samples, we found a strong dependence between the electrical properties and the oxygen background pressure used during the deposition process. As visible in Figure 6, resistivity values decrease by more than one order of magnitude while oxygen pressure decreases from 2.25 Pa to 1.25 Pa and increase again with a further pressure reduction to 1 Pa. In the same figure the variation of conduction electron density and mobility versus oxygen deposition pressure is also reported. The decrease in electrical properties of the film deposited at 1 Pa is connected to the drop of electron mobility (respectively 2.7 and 9.9 $\text{cm}^2\text{V}^{-1}\text{s}^{-1}$ for 1 and 1.25 Pa), in spite of the highest conduction electron density (respectively 1.21×10^{21} and $1.1 \times 10^{21} \text{ cm}^{-3}$ for 1 and 1.25 Pa).

In order to explore the optimization of the deposition process, the effect of a different laser fluence for TaTO and TiO_2 thin films was also explored. As the amount of ablated material is related to the energy density on target surface, it was necessary also to tune the oxygen background pressure during the ablation process in order to optimize electrical properties. Finally it was found that for 1.15 and 1.5

1
2
3 J/cm² laser fluences the best electrical properties for TaTO vacuum annealed films could be obtained
4
5 for 1 Pa oxygen pressure (with respect to the 1.25 Pa optimum condition found for the 0.9 J/cm²
6
7 fluence). The use of a higher laser fluence slightly increases the resistivity of the films (from 5.73×10⁻⁴
8
9 to 6.47×10⁻⁴ Ωcm, see Figure 7) due to a small decrease in the free electron density. On the other hand
10
11 by tuning the fluence it is possible to strongly enhance the electron mobility: for a TaTO 150 nm thick
12
13 deposited at a laser fluence of 1.15 J/cm² (1 Pa O₂) it was possible to obtain μ = 12.5 cm²V⁻¹s⁻¹ (ρ =
14
15 5.90×10⁻⁴ Ωcm, n = 8.49×10²⁰ cm⁻³, R_S = 39.3 Ω/□). For comparison we studied also a vacuum
16
17 annealed TiO₂ 150 nm thick film deposited at the same laser fluence, which showed as well a
18
19 noticeable increase in the electron mobility (ρ = 6.46×10⁻³ Ωcm, n = 6.16×10¹⁹ cm⁻³, μ = 15.7 cm²V⁻¹s⁻¹
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
1).

Finally we verified that the electrical properties are almost independent with respect to the TaTO film thickness: this provides the possibility of tuning the sheet resistance. In fact by almost doubling the thickness (295 nm) and maintaining the same laser fluence (1.15 J/cm²), we were able to obtain almost the same resistivity (ρ = 5.57×10⁻⁴ Ωcm, μ = 13.0 cm²V⁻¹s⁻¹, n = 8.61×10²⁰ cm⁻³) and consequently the sheet resistance dropped down to 18.9 Ω/□. It is interesting to note that the mobility values reported in this work are significantly larger than those reported in literature for polycrystalline films of doped TiO₂.^{17,19,37,38,51}

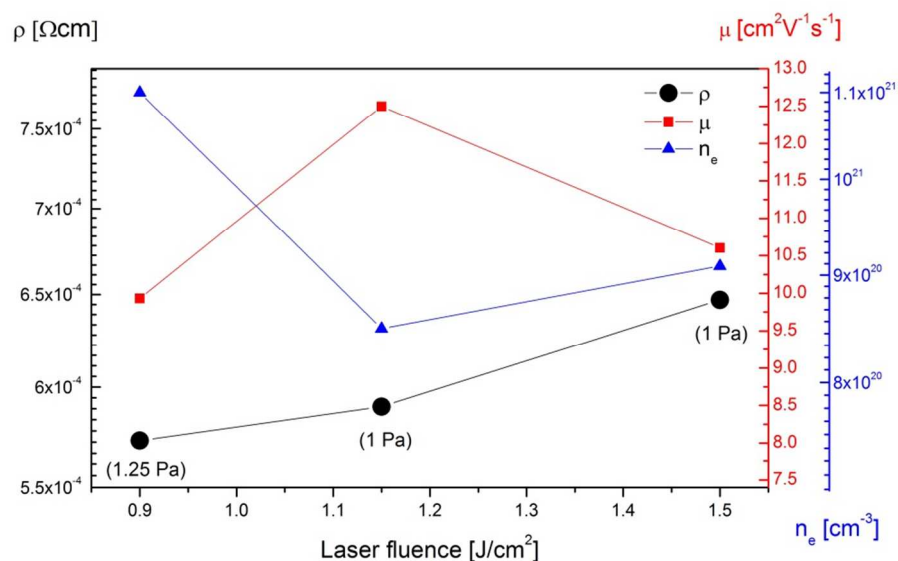


Figure 7. Electrical properties of TaTO films (150 nm) deposited at different laser fluence and vacuum annealed at 550°C. In the brackets are reported the oxygen deposition pressures. The black spots represent the resistivity, the blue triangles represent the electron density and the red squares represent the electron mobility evaluated via Hall measurements.

As clear from the behavior of vacuum annealed TaTO films deposited at different background pressures, oxygen clearly plays a key role in determining the electrical properties; this is confirmed by performing the crystallization process in air, which is known to be highly detrimental for both TiO₂ and TaTO thin films.³⁵ In fact, with our experimental setup, it was not possible to measure the resistivity (as well as the resistance in 2-point configuration measurements), suggesting a resistance of all the air-annealed films higher than 10 GΩ.

3.3. Optical properties

As well as structural and electrical properties, the optical properties of TaTO thin films also appear to be related to the oxygen background pressure during the deposition process. In Figure 8 total transmittance spectra for vacuum annealed TaTO films are reported, highlighting a tendency towards higher transmittance for higher oxygen background pressures during the deposition process. This is particularly evident in the near-IR region, where the absorption is due to plasma oscillations of the

“free” conduction band electrons (according to the Drude model), and lower transmittance at lower pressures should be related to higher free electron density values (see Figure 6). Moving to lower wavelengths it is possible to observe the presence of interference fringes in the visible range, whose position and spacing is related to the film thickness and refractive index, and then a sharp drop in the UV region because of the crossing of the optical band gap.

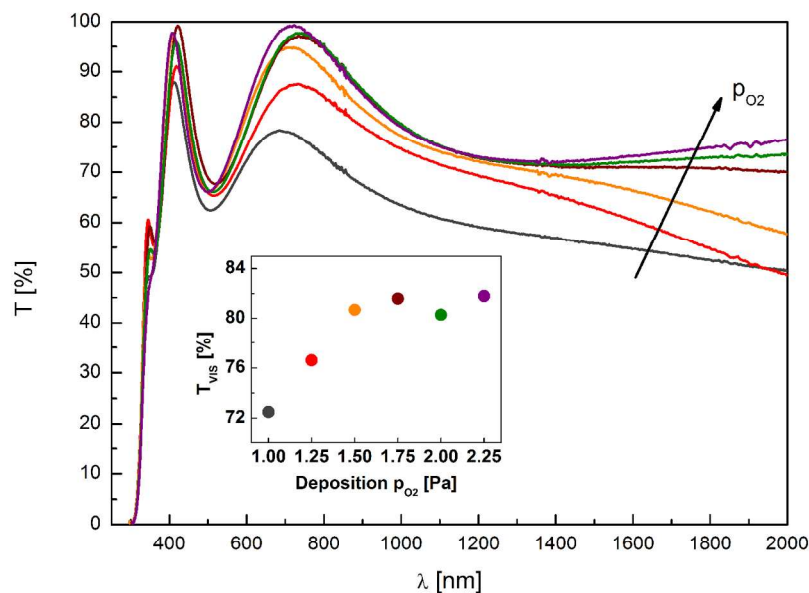


Figure 8. Total transmittance spectra of 150 nm thick TaTO films deposited with different oxygen partial pressures and annealed in vacuum atmosphere (550°C). In the inset graph the mean transmittance values evaluated in the visible region ($\lambda = 400 - 700$ nm) are reported as a function of different oxygen deposition pressures.

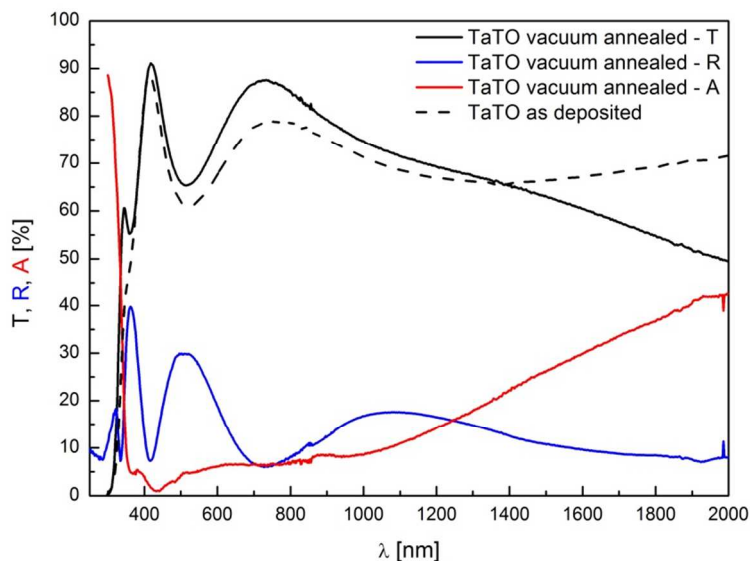


Figure 9. Total transmittance (black line), reflectance (blue line) and absorbance (red line) spectra of 150 nm thick TaTO film deposited at an oxygen partial pressure of 1.25 Pa annealed in vacuum atmosphere (550°C). The black dotted line refers to the total transmittance of the same sample before the crystallization process.

From the transmittance spectra the mean transmittance in the visible range (T_{VIS} for $\lambda = 400 - 700$ nm) can be calculated, finding in general higher values for higher oxygen background pressure during the deposition process (inset of Figure 8). For the TaTO vacuum annealed film deposited at 1.25 Pa, which shows the best electrical properties ($\rho = 5.73 \times 10^{-4} \Omega\text{cm}$), T_{VIS} is 76.6%. For this ‘best’ film we report in Figure 9 the transmittance, reflectance, and absorbance spectra, together with the transmittance spectrum of the amorphous as deposited film. The comparison reveals that the obtained mean transmittance value in the visible region is strongly dominated by reflectance ($R_{\text{VIS}} = 18.7\%$), while absorption is as low as 5%. We also note that the as-deposited amorphous film (dotted line in Figure 9) does not show the absorption in the near-IR due to plasma oscillations of the “free” conduction band electrons.

We also studied the effect of the annealing process in air on the optical properties, finding that for TaTO films deposited at 1 and 1.25 Pa, T_{VIS} is enhanced up to 79.9% and 80.0% respectively, while for

oxygen background pressures higher than 1.25 Pa, T_{VIS} values remained almost unchanged regardless of the annealing atmospheres. This fact could be related to the presence of a higher concentration of oxygen vacancy defects for the vacuum annealed TaTO deposited at lower oxygen pressures (1 – 1.25 Pa), as it is known that this occurrence usually leads to a decrease in the total transmittance values.^{52,53}

Finally we investigated the effect of the laser fluence on the optical properties of 150 nm thick TaTO films. In particular, for the sample deposited at a laser fluence of 1.15 J/cm² (see Figure 10 (a)), we were able to obtain an increase of the total transmittance in the visible range of nearly 5% ($T_{\text{VIS}} = 76.6\%$ and 81.1% respectively for 0.9 J/cm² and 1.15 J/cm²), even though they maintain almost the same electrical resistivity ($\rho = 5.73 \times 10^{-4} \text{ } \Omega\text{cm}$ and $5.90 \times 10^{-4} \text{ } \Omega\text{cm}$ respectively for 0.9 J/cm² and 1.15 J/cm²).

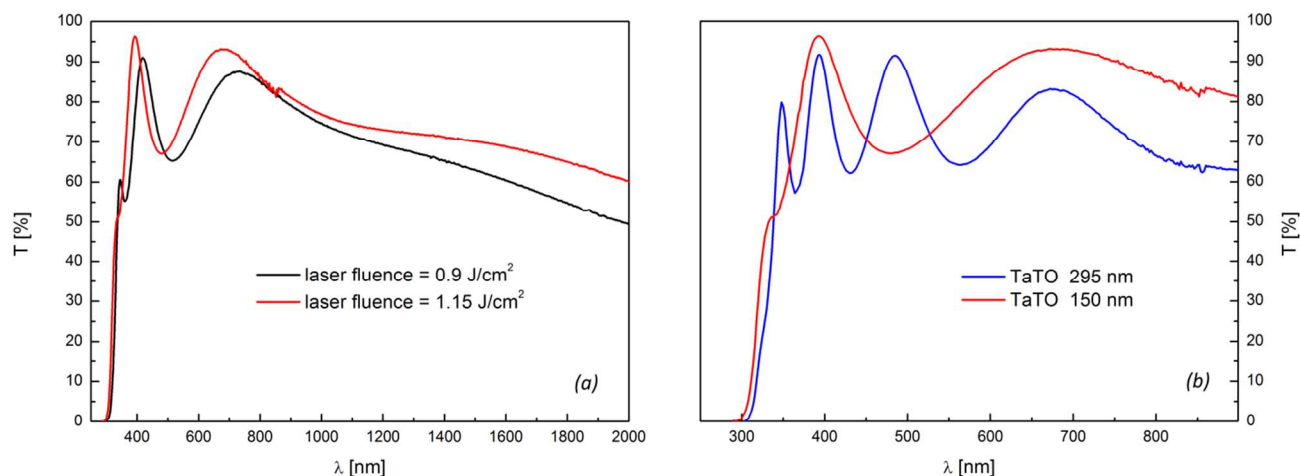


Figure 10. (a) Total transmittance spectra of 150 nm thick TaTO films deposited at a different laser fluence (0.9 and 1.15 J/cm²) annealed in vacuum atmosphere (550°C). (b) Total transmittance spectra of 150 and 295 nm thick TaTO films deposited at a laser fluence of 1.15 J/cm² annealed in vacuum atmosphere (550°C).

As for the thickness effect on the total transmittance, a nearly doubled thickness of 295 nm with respect to the 150 nm thick TaTO already shown in Figure 10 (a) (laser fluence 1.15 J/cm²) results in a decrease in T_{VIS} of 6% (from 81.1% to 75.2%, see Figure 10 (b)). Nevertheless we note that it could be

possible to engineer the light transmittance at certain wavelengths in a very precise way taking advantage of the position of the well-defined and peaked thickness fringes (e.g. total transmittance higher than 90% in the regions of wavelength close to 390 and 480 nm for the 295 nm thick TaTO film, blue line in Figure 10 (b)).

4. Discussion

Optical and electrical properties of TaTO films are strongly connected to the oxygen background pressure during the PLD process, even though an annealing process is needed to crystallize the films in the anatase phase. Oxygen vacancies (V_O) are known to play a fundamental role in the electrical properties of TiO_2 because they should represent an effective doubly negative charged donor state,⁵⁰ but the reported trends in the TaTO functional properties presented above are likely related to a more complex defect chemistry. For instance, the abrupt increase in resistivity from the vacuum to the air annealing process (from $5.73 \times 10^{-4} \Omega\text{cm}$ up to a resistance value higher than 10 G Ω) is not easily explained only with the absence of V_O resulting from an oxygen-rich atmosphere. In fact, the incorporation of Ta as an active dopant (5% at.), due to the substitution of Ti atoms in the anatase TiO_2 cell (Ta_{Ti}), should provide a free electron concentration⁵⁴ in the order of 10^{21} cm^{-3} (in line with our data shown in Figure 5 for vacuum annealed TaTO); the amount of free electrons due to the formation of doubly positive charged oxygen vacancies (V_O) during a vacuum annealing process has been proved to be in the order of 10^{19} cm^{-3} (i.e. for vacuum annealed undoped TiO_2 , see Figure 5). Therefore, in agreement with what is supported by Huy *et al.*,¹⁵ the lack (or the reduced amount) of V_O that should be associated to oxygen-rich conditions cannot explain alone the highly insulating properties obtained for TaTO air annealed films or the drop in conductivity for higher oxygen deposition pressures (see Figure 6); nevertheless, we are aware that this has to be just considered as a too simple approach to a complex problem. For instance, the possibility that the insertion of Ta in the anatase matrix could itself

1
2
3 affect the concentration of produced oxygen vacancies in reducing atmosphere with respect to the
4 undoped TiO₂ should also be taken into account. One possible explanation to the oxygen-related loss
5 of conductivity suggested by Huy *et al.*,¹⁵ is that the solubility of Ta in the anatase lattice is reduced in
6 oxygen-rich conditions, leading to the formation of the non-doping Ta₂O₅ phase. However, O-rich
7 conditions are expected to reduce only slightly the Ta solubility (and the number of Ta_{Ti} defects);^{28,31,32}
8 we observe that in our structural characterization we did not collect any experimental evidence of Ta
9 segregated phases (see Section 3.1).
10
11
12
13
14
15
16
17
18

19
20 In this work we are not aiming to fully clarify this aspect, but still we can infer that there should be a
21 non-obvious interplay of different defects (*e.g.* oxygen - titanium - tantalum) governing the electrical
22 (and optical) properties. In a recent work Qi *et al.*³⁴ have discussed how Ti vacancies could play an
23 important role in the defect chemistry of TaTO. These defects can actually act as “electron killers” and
24 it has been shown that growing TiO₂-based films under oxygen rich condition should promote the
25 occurrence of cationic vacancies and oxygen interstitials.^{27,55} The coupling between a decreasing
26 concentration of V_O and the occurrence of an increasing amount of electron “killer-defects” in the
27 anatase cell as the oxygen deposition pressure is increased could be related to the decrease in the
28 charge carrier density reported in Figure 6 (from 1.21×10²¹ cm⁻³ to 9.26×10¹⁹ cm⁻³ when moving from
29 1 Pa to 2.25 Pa). Still it is difficult to state if the defect concentration in the crystal lattice is fixed by
30 the deposition process, and to which extent defects can be ruled by the annealing process. A study on
31 thermodynamics of defects formation at a given annealing temperature and in different oxygen partial
32 pressures would be recommended for this material, in order to clarify the kind of defects involved and
33 the kinetics of the formation process.
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52

53
54 On the other hand, also the mobility of the vacuum annealed TaTO samples has been shown to be
55 connected to the oxygen background pressure during the deposition process (Figure 6). This trend
56
57
58
59
60

could be linked to the structural properties of the films: the presence of circular macro-cracks only observed on the surface of the TaTO deposited at 1 Pa (Figure 4 (a)), could explain its significant drop in electron mobility; the decrease seen for deposition pressures higher than 1.25 Pa, is likely related to the decrease of the crystalline domain size in both vertical and horizontal direction revealed by XRD and optical microscope analysis (Figure 3 (a) and Figure 4). Nevertheless the role of an increased concentration of the above mentioned “killer-defects” with the increase of oxygen deposition pressure could also contribute to both the decrease in the electron mobility and the mean crystalline size and should be considered.

In Section 3.1 we mentioned the presence of a shift of XRD reflections for vacuum annealed TaTO with respect to undoped TiO₂ films, dependent on the oxygen deposition pressure (see Figure 3 (a)). In Figure 11 we report the anatase cell parameters calculated from (101) and (200) XRD peak positions evaluated in grazing incident angle configuration for vacuum and air annealed films (TiO₂ and TaTO 1.25 Pa).

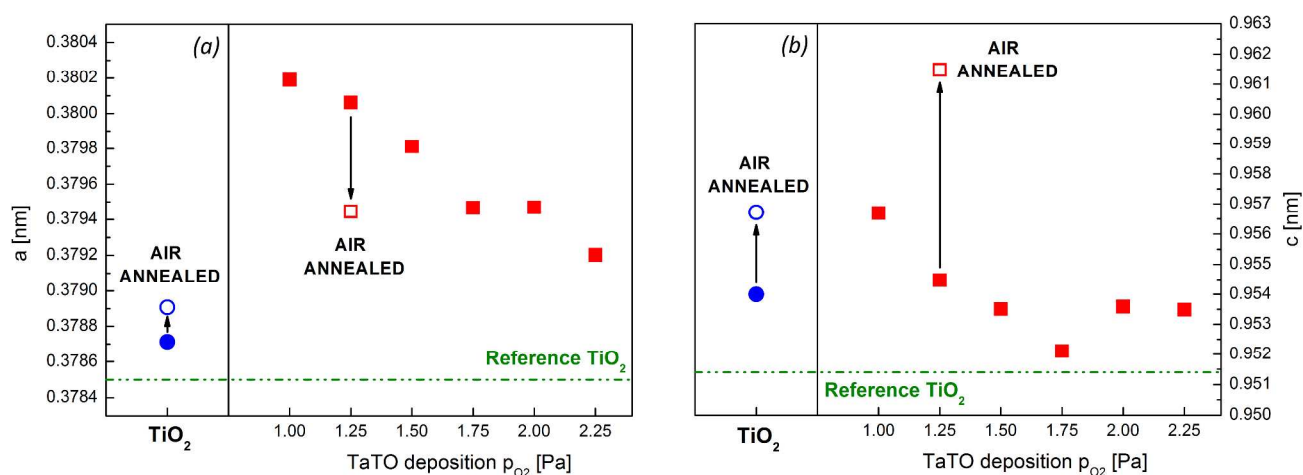


Figure 11. Anatase cell parameters (*a* and *c* respectively in (a) and (b)) evaluated for TiO₂ (blue circles) and TaTO (red squares) vacuum (filled) and air (TiO₂ and 1.25 Pa TaTO - empty) annealed films. TaTO cell parameters are plotted as a function of the oxygen

1
2
3 background pressure during the deposition process. The green dotted line represent the reference values for a nominally pure anatase
4
5 TiO_2 .⁵⁶
6
7

8 A comparison between doped and undoped vacuum annealed films shows that both a and c anatase cell
9 parameters (presented in Figure 11 (a) and (b) respectively) are generally larger for TaTO films: this
10 could be related to a lattice expansion due to Ta incorporation in Ti substitutional sites, as commented
11 above. Nevertheless, it is possible to notice that both a and c parameters are slightly decreasing
12 (approaching the values for the vacuum annealed TiO_2 cell) with increasing p_{O_2} ; this observation could
13 be related to a decreasing amount of oxygen vacancies as the deposition pressure is increased in TaTO
14 thin films. It is known from theoretical calculations that V_{O} are thought to give a small lattice distortion
15 in the TiO_2 anatase cell, but it is plausible that a strong interplay between V_{O} and substitutional Ta
16 takes place.^{57,58} The presence of a decreasing amount of V_{O} as the deposition pressure is increased is in
17 line also with the transmittance data shown in Figure 8. Nevertheless when the crystallization process is
18 performed in air, on TaTO (1.25 Pa p_{O_2}) there is an increment in c coupled with a decrease in a; in this
19 case the change in the lattice parameters is difficult to explain only as a lack of V_{O} in the crystal lattice
20 and it is likely related to the presence of different defects induced by the annealing in the presence of
21 oxygen (e.g. Ti vacancies, O interstitials). Finally, looking at the undoped TiO_2 , it is possible to note
22 that the mismatch with respect to the nominally pure anatase cell parameters (green line in Figure 11⁵⁶),
23 is increasing for the sample annealed in air atmosphere; this evidence could again be related to the
24 above mentioned presence of oxygen-induced defects.
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47

48 We also evaluated the optical gap for the vacuum annealed TaTO films by means of Tauc plots.
49 Because of the indirect band gap of TiO_2 , $(\alpha h\nu)^{0.5}$ versus $h\nu$ was plotted in the proximity of the
50 absorption onset. The absorption coefficient α was evaluated from transmittance and reflectance spectra
51 via the Lambert-Beer law (with $\alpha = -1/d (T/(1-R))$). Thus the optical band gap ΔE_{G} was extrapolated
52 with the intercept in the energy axis using a linear fit. The obtained optical energy gaps for the TaTO
53
54
55
56
57
58
59
60

vacuum annealed films deposited at different oxygen background pressures are plotted as a function of the charge carrier density (Figure 12), since we aim at evaluating the possible presence of a Moss-Burstein effect due to the filling of the conduction band by free carriers.⁵⁹

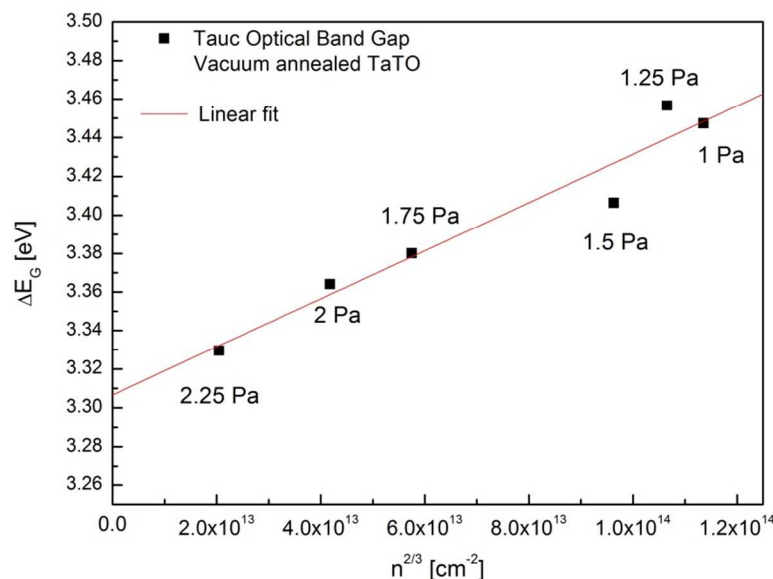


Figure 12. Tauc optical band gap as a function of $n^{2/3}$ for vacuum annealed TaTO films deposited at different oxygen partial pressures.

The charge carrier densities were taken from Hall effect measurements reported in Figure 6.

As observed in Figure 12, ΔE_G shows a trend as a function of the charge carrier density. The reliability of these data is confirmed by the evaluation of the energy value obtained by the y-axis intercept of the linear fit, i.e. 3.31 eV. In fact this value is in line with the band gap of anatase TiO_2 , which is known to be around 3.2 – 3.4 eV.^{9,10} Since in a simplified Drude model picture the shift in the optical band gap should obey the equation:

$$\Delta E_G = \frac{\hbar^2}{2m^*} (3\pi^2 n)^{2/3}$$

from the slope of the linear fit we are able to estimate the electron effective mass to be $m^* = 2.9m_0$, where m_0 is the electron rest mass. The evaluated m^* should be considered as a combination of the two different values of effective mass reported for the anisotropic anatase cell, the orthogonal m_x^* and

1
2
3 parallel m_z^* (with respect to the tetragonal axis), since it is likely that the electron path is almost
4
5 randomly oriented with respect to the cell. While m_x^* was evaluated to be in the range 0.4 - 0.6 m_0 ,
6
7 which is actually the usual TCO range,⁶⁰ m_z^* is definitely higher since no value smaller than 3.5 m_0 has
8
9 ever been theoretically deduced from anatase band structure calculations.^{15,61} According to these
10
11 values, the electron effective mass evaluated in this work from optical measurements seems to be
12
13 consistent. Nonetheless this value should be read just as an estimate since it does not take into account
14
15 the many-body effects and the nonparabolicity of the conduction band, which usually lead an
16
17 overestimate of the deduced m^* . Moreover, based on the comparison between experimental data and
18
19 DFT calculations, Huy *et al.*^{15,61} found a severe increase of m_z^* with increasing charge carrier
20
21 concentration for both NTO and TaTO. The enhancement of the optical effective mass with carrier
22
23 density in the direction parallel to the main axis of the Ta-doped system was calculated to be less than
24
25 60% of that in Nb-doped anatase.¹⁵ Since a higher m^* should be detrimental for charge carrier mobility,
26
27 the large values of the mobility reported in this work for TaTO polycrystalline films with respect to
28
29 NTO ones could be possibly explained in this way (up to $\mu = 13.0 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$ for $n = 8.61 \times 10^{20} \text{ cm}^{-3}$).
30
31 This represents a 40% improvement with respect to the highest mobility values reported for a
32
33 polycrystalline NTO film ($\mu = 9.1 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$ for $n = 1.7 \times 10^{21} \text{ cm}^{-3}$), interestingly obtained for a (001)
34
35 preferential orientation.⁵¹ Of course we are aware that in polycrystalline films scattering by grain
36
37 boundaries can play a relevant role in determining electron mobility; however, no evidence of mobility
38
39 values higher than $8 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$ has been reported for randomly oriented polycrystalline NTO films.
40
41
42
43
44
45
46
47
48

49 Finally we discuss the light absorption by the gas of conduction carriers: the plasma wavelength λ_p can
50
51 be evaluated from the relation:
52
53
54

$$\lambda_p = 2\pi c \left(\frac{\epsilon_0 \epsilon_\infty m^*}{n_e e^2} \right)^{1/2}$$

1
2
3 where ϵ_0 , ϵ_∞ , c , m^* , and e denote the dielectric constant of vacuum, the high-frequency permittivity, the
4 speed of light, the effective mass, and the electronic charge, respectively. Since ϵ_∞ was reported to be
5 equal to 5.9 for TiO_2 ,⁶² and the electron effective mass was estimated in this work to be $m^* = 2.9m_0$,
6 the maximum absorption by electrons in the conduction band can be estimated to be at a wavelength of
7 4160 nm for $n = 1.1 \times 10^{21} \text{ cm}^{-3}$. This value seems to be consistent with the transmittance spectra shown
8 in Figure 9, although it was not possible to experimentally verify the maximum λ_p absorption, due to
9 the upper limit of our experimental setup ($\lambda_{\text{max}} = 2000 \text{ nm}$). This high value of λ_p is definitely an
10 advantage for TCO applications, since for other materials such as ZnO , In_2O_3 and SnO_2 based TCOs,
11 the useful number of the free electrons is limited by a shift ($\propto \sqrt{m^*}$) of λ_p in the visible region.⁶³ This
12 makes TaTO a very interesting material for application as a TCO in solar cells, in which an enhanced
13 transmittance at longer wavelengths with respect to the visible region (400 – 1100 nm) is a preferred
14 property.¹⁴

32 33 **5. Conclusions**

34
35
36 We have performed a thorough investigation of the electrical and optical properties of Ta-doped TiO_2
37 polycrystalline films in order to understand and control the real potentialities of this novel TCO. It has
38 been shown how it is possible to combine low resistivity values of the order of $5 \times 10^{-4} \Omega\text{cm}$ with mean
39 transmittance in the visible region exceeding 80%. We discussed the effect of the oxygen background
40 pressure during the deposition process and the role of the post annealing atmosphere.
41
42
43
44
45
46
47
48

49 We were able to estimate the optical electron effective mass and we analyzed the obtained data in view
50 of the high mobility values obtained for our polycrystalline thin films (up to $13 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$). This shows
51 how Ta could have definite advantages with respect to the most investigated Nb as a TiO_2 dopant for
52 TCO applications.
53
54
55
56
57
58
59
60

References:

1. O'Regan, B.; Gratzel, M. A Low-Cost, High-Efficiency Solar Cell Based on Dye-Sensitized Colloidal TiO₂ Films. *Nature*, **1991**, *353*, 737-740.
2. Lee, M.M.; Teuscher, J.; Miyasaka, T.; Murakami, T. N.; Snaith, H. J. Efficient Hybrid Solar Cells Based on Meso-Superstructured Organometal Halide Perovskites. *Science*, **2012**, *338*, 643-647.
3. Gondoni, P.; Mazzolini, P.; Russo, V.; Petrozza, A.; Srivastava, A. K.; Li Bassi, A.; Casari, C. S. Enhancing Light Harvesting by Hierarchical Functionally Graded Transparent Conducting Al-Doped ZnO Nano- and Mesoarchitectures. *Sol. Energ. Mat. and Sol. C.*, **2014**, *128*, 248-253.
4. Gong, J.W., Liang, J.; Sumathy, K. Review on Dye-Sensitized Solar Cells (DSSCs): Fundamental Concepts and Novel Materials. *Renew. Sust. Energ. Rev.*, **2012**, *16*, 5848-5860.
5. Rühle, S.; Dittrich T. Investigation of the Electric Field in TiO₂/FTO Junctions Used in Dye-Sensitized Solar Cells by Photocurrent Transients. *J. Phys. Chem. B*, **2005**, *109*, 9522-9526.
6. Snaith, H.J.; Grätzel M. The Role of a "Schottky Barrier" at an Electron-Collection Electrode in Solid-State Dye-Sensitized Solar Cells. *Adv. Mater.*, **2006**, *18*, 1910-1914.
7. Xue, G.; Guo, Y.; Yu, T.; Guan, J.; Yu, X.; Zhang, J.; Liu, J.; Zou, Z. Degradation Mechanisms Investigation for Long-Term Thermal Stability of Dye-Sensitized Solar Cells. *Int. J. Electrochem. Sc.*, **2012**, *7*, 1496-1511.
8. Hagfeldt, A.; Boschloo, G.; Sun, L.; Kloo, L.; Pettersson, H. Dye-Sensitized Solar Cells. *Chem. Rev.*, **2010**, *110*, 6595-6663.
9. Tang, H.; Berger, H.; Schmid, P. E.; Lévy, F.; Burri, G. Photoluminescence in TiO₂ Anatase Single Crystals. *Solid State Commun.*, **1993**, *87*, 847-850.

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
10. Tang, H.; Lévy, F.; Berger, H.; Schmid, P. E. Urbach Tail of Anatase TiO₂. *Phys. Rev. B*, **1995**, *52*, 7771-7774.
 11. Furubayashi, Y.; Hitosugi, T.; Yamamoto, Y.; Inaba, K.; Kinoda, G.; Hirose, Y.; Shimada, T.; Hasegawa, T. A Transparent Metal: Nb-Doped Anatase TiO₂. *Appl. Phys. Lett.*, **2005**, *86*, 252101.
 12. Hitosugi, T.; Furubayashi, Y.; Ueda, A.; Itabashi, K.; Inaba, K.; Hirose, Y.; Kinoda, G.; Yamamoto, Y.; Shimada, T.; Hasegawa, T. Ta-Doped Anatase TiO₂ Epitaxial Film as Transparent Conducting Oxide. *Jpn. J. Appl. Phys. 2*, **2005**, *44*, L1063-L1065.
 13. Minami, T. Transparent Conducting Oxide Semiconductors for Transparent Electrodes. *Semicond. Sci. Tech.*, **2005**, *20*, S35-S44.
 14. Ellmer, K. Past Achievements and Future Challenges in the Development of Optically Transparent Electrodes. *Nat. Photonics*, **2012**, *6*, 809-817.
 15. Huy, H.A.; Aradi, B.; Frauenheim, T.; Deák, P. Comparison of Nb- and Ta-Doping of Anatase TiO₂ for Transparent Conductor Applications. *J. Appl. Phys.*, **2012**, *112*, 016103.
 16. Furubayashi, Y.; Yamada, N.; Hirose, Y.; Yamamoto, Y.; Otani, M.; Hitosugi, T.; Shimada, T.; Hasegawa, T. Transport Properties of d-Electron-Based Transparent Conducting Oxide: Anatase Ti_{1-x}Nb_xO₂. *J. Appl. Phys.*, **2007**, *101*, 093705.
 17. Hitosugi, T.; Ueda, A.; Nakao, S.; Yamada, N.; Furubayashi, Y.; Hirose, Y.; Shimada, T.; Hasegawa, T., Fabrication of Highly Conductive Ti_{1-x}Nb_xO₂ Polycrystalline Films on Glass Substrates Via Crystallization of Amorphous Phase Grown by Pulsed Laser Deposition. *Appl. Phys. Lett.*, **2007**, *90*, 212106.
 18. Yamada, N.; Hitosugi, T.; Hoang, N. L. H.; Furubayashi, Y.; Hirose, Y.; Shimada, T.; Hasegawa, T. Fabrication of Low Resistivity Nb-Doped TiO₂ Transparent Conductive

- 1
2
3 Polycrystalline Films on Glass by Reactive Sputtering. *Jpn. J. Appl. Phys. 1*, **2007**, *46*, 5275-
4
5 5277.
6
7
8 19. Dabney, M.S.; van Hest, M. F. A. M.; Teplin, C. W.; Arenkiel, S. P.; Perkins, J. D.; Ginley, D.
9
10 S. Pulsed Laser Deposited Nb Doped TiO₂ as a Transparent Conducting Oxide. *Thin Solid*
11
12 *Films*, **2008**, *516*, 4133-4138.
13
14
15 20. Hitosugi, T; Kamisaka, H.; Yamashita, K.; Nogawa, H.; Furubayashi, Y.; Nakao, S.; Yamada,
16
17 N.; Chikamatsu, A.; Kumigashira, H.; Oshima, M.; et al. Electronic Band Structure of
18
19 Transparent Conductor: Nb-Doped Anatase TiO₂. *Appl. Phys. Express*, **2008**, *1*, 1112031-
20
21 1112033.
22
23
24 21. Hitosugi, T.; Ueda, A.; Nakao, S.; Yamada, N.; Furubayashi, Y.; Hirose, Y.; Konuma, S.;
25
26 Shimada, T.; Hasegawa, T. Transparent Conducting Properties of Anatase Ti_{0.94}Nb_{0.06}O₂
27
28 Polycrystalline Films on Glass Substrate. *Thin Solid Films*, **2008**, *516*, 5750-5753.
29
30
31 22. Yamada, N.; Hitosugi, T.; Hoang, N. L. H.; Furubayashi, Y.; Hirose, Y.; Konuma, S.; Shimada,
32
33 T.; Hasegawa, T. Structural, Electrical and Optical Properties of Sputter-Deposited Nb-Doped
34
35 TiO₂ (TNO) Polycrystalline Films. *Thin Solid Films*, **2008**, *516*, 5754-5757.
36
37
38 23. Lee, S.; Noh, J. H.; Han, H. S.; Yim, D. K.; Kim, D. H.; Lee, J.-K.; Kim, J. Y.; Jung, H. S.;
39
40 Hong, K. S. Nb-Doped TiO₂: A New Compact Layer Material for TiO₂ Dye-Sensitized Solar
41
42 Cells. *J. Phys. Chem. C*, **2009**, *113*, 6878-6882.
43
44
45 24. Tonooka, K.; Chiu, T.-W.; Kikuchi, N. Preparation of Transparent Conductive TiO₂:Nb Thin
46
47 Films by Pulsed Laser Deposition. *Appl. Surf. Sci.*, **2009**, *255*, 9695-9698.
48
49
50 25. Yamada, N.; Hitosugi, T.; Kasai, J.; Hoang, N. L. H.; Nakao, S.; Hirose, Y.; Shimada, T.;
51
52 Hasegawa, T. Direct Growth of Transparent Conducting Nb-Doped Anatase TiO₂
53
54 Polycrystalline Films on Glass. *J. Appl. Phys.*, **2009**, *105*, 123702.
55
56
57
58
59
60

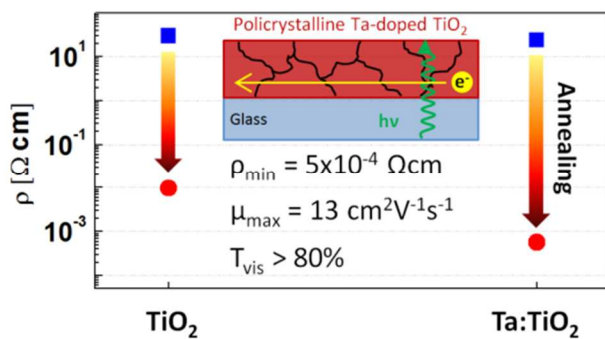
- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
26. Nogawa, H.; Chikamatsu, A.; Hirose, Y.; Nakao, S.; Kumigashira, H.; Oshima, M.; Hasegawa, T. Carrier Compensation Mechanism in Heavily Nb-Doped Anatase $\text{Ti}_{1-x}\text{Nb}_x\text{O}_{2+\delta}$ Epitaxial Thin Films. *J. Phys. D Appl. Phys.*, **2011**, *44*, 365404.
27. Lee, H.-Y.; Robertson J. Doping and Compensation in Nb-Doped Anatase and Rutile TiO_2 . *J. Appl. Phys.*, **2013**, *113*, 213706.
28. Neubert, M.; Cornelius, S.; Fiedler, J.; Gebel, T.; Liepack, H.; Kolitsch, A.; Vinnichenko, M. Overcoming Challenges to the Formation of High-Quality Polycrystalline TiO_2 :Ta Transparent Conducting Films by Magnetron Sputtering. *J. Appl. Phys.*, **2013**, *114*, 083707.
29. Ok, K.-C.; Park, Y.; Chung, K.-B.; Park, J.-S. The Effect of Ta Doping in Polycrystalline TiO_x and the Associated Thin Film Transistor Properties. *Appl. Phys. Lett.*, **2013**, *103*, 213501.
30. Wu, B.B.; Pan, F. M.; Yang, Y. E. Annealing Effect of Pulsed Laser Deposited Transparent Conductive Ta-Doped Titanium Oxide Films. *Chinese Phys. Lett.*, **2011**, *28*, 118102.
31. Osorio-Guillén, J.; Lany, S.; Zunger, A. Atomic Control of Conductivity Versus Ferromagnetism in Wide-Gap Oxides Via Selective Doping: V, Nb, Ta in Anatase TiO_2 . *Phys. Rev. Lett.*, **2008**, *100*, 036601.
32. Rusydi, A.; Dhar, S.; Barman, A. R.; Ariando; Qi, D.-C.; Motapothula, M.; Yi, J. B.; Santoso, I.; Feng, Y. P.; Yang, K.; et al. Cationic-Vacancy-Induced Room-Temperature Ferromagnetism in Transparent, Conducting Anatase $\text{Ti}_{1-x}\text{Ta}_x\text{O}_2$ ($x \sim 0.05$) Thin Films. *Philos. T. Roy. Soc. A*, **2012**, *370*, 4927-4943.
33. Yamamoto, T.; Ohno T. Screened Hybrid Density Functional Study on Nb- and Ta-Doped TiO_2 . *Phys. Rev. B*, **2012**, *85*, 033104.
34. Qi, D.C.; Barman, A. R.; Debbichi, L.; Dhar, S.; Santoso, I.; Asmara, T. C.; Omer, H.; Yang, K.; Krüger, P.; Wee, A. T. S.; et al. Cationic Vacancies and Anomalous Spectral-Weight

- 1
2
3 Transfer in $Ti_{1-x}Ta_xO_2$ Thin Films Studied Via Polarization-Dependent Near-Edge X-Ray
4 Absorption Fine Structure Spectroscopy. *Phys. Rev. B*, **2013**, *87*, 245201.
5
6
7
8 35. Hitosugi, T.; Yamada, N.; Nakao, S.; Hirose, Y.; Hasegawa, T. Properties of TiO_2 -Based
9 Transparent Conducting Oxides. *Phys. Status Solidi A*, **2010**, *207*, 1529-1537.
10
11
12 36. Hitosugi, T.; Ueda, A.; Furubayashi, Y.; Hirose, Y.; Konuma, S.; Shimada, T.; Hasegawa, T.
13 Fabrication of TiO_2 -Based Transparent Conducting Oxide Films on Glass by Pulsed Laser
14 Deposition. *Jpn. J. Appl. Phys. 2*, **2007**, *46*, L86-L88.
15
16
17 37. Hoang, N.L.H.; Yamada, N.; Hitosugi, T.; Kasai, J.; Nakao, S.; Shimada, T.; Hasegawa, T.
18 Low-Temperature Fabrication of Transparent Conducting Anatase Nb-Doped TiO_2 Films by
19 Sputtering. *Appl. Phys. Express*, **2008**, *1*, 1150011-1150013.
20
21
22 38. Hitosugi, T.; Yamada, N.; Hoang, N. L. H.; Kasai, J.; Nakao, S.; Shimada, T.; Hasegawa, T.
23 Fabrication of TiO_2 -Based Transparent Conducting Oxide on Glass and Polyimide Substrates.
24 *Thin Solid Films*, **2009**, *517*, 3106-3109.
25
26
27 39. Sato, Y.; Sanno, Y.; Tasaki, C.; Oka, N.; Kamiyama, T.; Shigesato, Y. Electrical and Optical
28 Properties of Nb-Doped TiO_2 Films Deposited by DC Magnetron Sputtering Using Slightly
29 Reduced Nb-Doped TiO_{2-x} Ceramic Targets. *J. Vac. Sci. Technol. A*, **2010**, *28*, 851-855.
30
31
32 40. Liu, J.; Zhao, X.; Duan, L.; Cao, M.; Sun, H.; Shao, J.; Chen, S.; Xie, H.; Chang, X.; Chen, C.
33 Influence of Annealing Process on Conductive Properties of Nb-Doped TiO_2 Polycrystalline
34 Films Prepared by Sol-Gel Method. *Appl. Surf. Sci.*, **2011**, *257*, 10156-10160.
35
36
37 41. Kim, H. Transparent Conducting Oxide Films. In *Pulsed laser deposition of thin films:
38 applications-led growth of functional materials*; Eason, R.; John Wiley & Sons, Inc.: Hoboken,
39 NJ, 2007; pp. 236-260.
40
41
42 42. Balachandran, U.; Eror N. G. Raman Spectra of Titanium Dioxide. *J. Solid State Chem.*, **1982**,
43 *42*, 276-282.
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
43. Li Bassi, A.; Cattaneo, D.; Russo, V.; Bottani, C. E.; Barborini, E.; Mazza, T.; Piseri, P.; Milani, P.; Ernst, F. O.; Wegner, K.; et al. Raman Spectroscopy Characterization of Titania Nanoparticles Produced by Flame Pyrolysis: The Influence of Size and Stoichiometry. *J. Appl. Phys.*, **2005**, *98*, 074305.
44. Oliver, P. M.; Watson, G. W.; Toby Kelsey, E.; Parker, S. C. Atomistic Simulation of the Surface Structure of the TiO₂ Polymorphs Rutile and Anatase. *J. Mater. Chem.*, **1997**, *7*, 563-568.
45. Yang, C.; Hirose, Y.; Nakao, S.; Hasegawa, T. C-Axis-Oriented Growth of Anatase TiO₂ Thin Films on Glass Substrate with SrTiO₃/TiN Template. *J. Cryst. Growth*, **2013**, *376*, 66-69.
46. Dy, E.; Hui, R.; Zhang, J.; Liu, Z.-S.; Shi, Z. Electronic Conductivity and Stability of Doped Titania (Ti_{1-x}M_xO₂, M = Nb, Ru, and Ta)—A Density Functional Theory-Based Comparison. *J. Phys. Chem. C*, **2010**, *114*, 13162-13167.
47. Pore, V.; Ritala, M.; Leskelä, M.; Saukkonen, T.; Järn, M. Explosive Crystallization in Atomic Layer Deposited Mixed Titanium Oxides. *Cryst. Growth Des.*, **2009**, *9*, 2974-2978.
48. Tachikawa, T.; Minohara, M.; Nakanishi, Y.; Hikita, Y.; Yoshita, M.; Akiyama, H.; Bell, C.; Hwang, H.Y. Metal-to-Insulator Transition in Anatase TiO₂ Thin Films Induced by Growth Rate Modulation. *Appl. Phys. Lett.*, **2012**, *101*, 022104.
49. Chen, D.-m.; Xu, G.; Miao, L.; Nakao, S.; Jin, P. Sputter Deposition and Computational Study of M-TiO₂ (M = Nb, Ta) Transparent Conducting Oxide Films. *Surf. Coat. Tech.*, **2011**, *206*, 1020-1023.
50. Deák, P.; Aradi, B.; Frauenheim, T. Quantitative Theory of the Oxygen Vacancy and Carrier Self-Trapping in Bulk TiO₂. *Phys. Rev. B*, **2012**, *86*, 195206.
51. Yamada, N.; Shibata, T.; Taira, K.; Hirose, Y.; Nakao, S.; Ngoc, L. H. H.; Hitosugi, T.; Shimada, T.; Sasaki, T.; Hasegawa, T. Enhanced Carrier Transport in Uniaxially (001)-Oriented

- 1
2
3 Anatase $\text{Ti}_{0.94}\text{Nb}_{0.06}\text{O}_2$ Films Grown on Nanosheet Seed Layers. *Appl. Phys. Express*, **2011**, *4*,
4 045801.
5
6
7
8 52. Brudnik, A.; Czternastek, H.; Zakrzewska, K.; Jachimowski, M. Plasma-Emission-Controlled
9 DC Magnetron Sputtering of TiO_{2-x} Thin Films. *Thin Solid Films*, **1991**, *199*, 45-58.
10
11
12 53. Poelman, H.; Tomaszewski, H.; Poelman, D.; Depla, D.; De Gryse, R. Effect of the Oxygen
13 Deficiency of Ceramic TiO_{2-x} Targets on the Deposition of TiO_2 Thin Films by DC Magnetron
14 Sputtering. *Surf. Interface Anal.*, **2004**, *36*, 1167-1170.
15
16
17
18 54. Deák, P.; Aradi, B.; Frauenheim, T. Polaronic Effects in TiO_2 Calculated by the HSE06 Hybrid
19 Functional: Dopant Passivation by Carrier Self-Trapping. *Phys. Rev. B*, **2011**, *83*, 155207.
20
21
22
23 55. Zhang, S.; Ogale, S. B.; Yu, W.; Gao, X.; Liu, T.; Ghosh, S.; Das, G. P.; Wee, A. T. S.; Greene,
24 R. L.; Venkatesan, T. Electronic Manifestation of Cation-Vacancy-Induced Magnetic Moments
25 in a Transparent Oxide Semiconductor: Anatase Nb:TiO₂. *Adv. Mater.*, **2009**, *21*, 2282-2287.
26
27
28
29 56. Hanaor, D.A.H.; Sorrell, C. C. Review of the Anatase to Rutile Phase Transformation. *J. Mater.*
30 *Sci.*, **2011**, *46*, 855-874.
31
32
33
34 57. Peng, H.; Li, J.; Li, S.-S.; Xia, J.-B. Possible Origin of Ferromagnetism in Undoped Anatase
35 TiO₂. *Phys. Rev. B*, **2009**, *79*, 092411.
36
37
38
39 58. Kim, D.; Hong, J.; Park, Y. R.; Kim, K. J. The Origin of Oxygen Vacancy Induced
40 Ferromagnetism in Undoped TiO₂. *J. Phys. Condens. Matter*, **2009**, *21*, 195405.
41
42
43
44 59. Burstein, E. Anomalous Optical Absorption Limit in InSb. *Phys. Rev.*, **1954**, *93*, 632-633.
45
46
47
48 60. Gondoni, P.; Ghidelli, M.; Di Fonzo, F.; Carminati, M.; Russo, V.; Bassi, A. L.; Casari, C. S.
49 Structure-Dependent Optical and Electrical Transport Properties of Nanostructured Al-Doped
50 ZnO. *Nanotechnology*, **2012**, *23*, 365706.
51
52
53
54
55
56
57
58
59
60

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
61. Huy, H.A.; Aradi, B.; Frauenheim, T.; Deák, P. Calculation of Carrier-Concentration-Dependent Effective Mass in Nb-Doped Anatase Crystals of TiO₂. *Phys. Rev. B*, **2011**, *83*, 155201.
62. Gonzalez, R. J.; Zallen, R.; Berger, H. Infrared Reflectivity and Lattice Fundamentals in Anatase TiO₂. *Phys. Rev. B*, **1997**, *55*, 7014-7017.
63. Gondoni, P.; Ghidelli, M.; Di Fonzo, F.; Russo, V.; Bruno, P.; Marti-Rujas, J.; Bottani, C. E.; Bassi, A. L.; Casari, C. S. Structural and Functional Properties of Al:ZnO Thin Films Grown by Pulsed Laser Deposition at Room Temperature. *Thin Solid Films*, **2012**, *520*, 4707-4711.



For Table of Contents Only

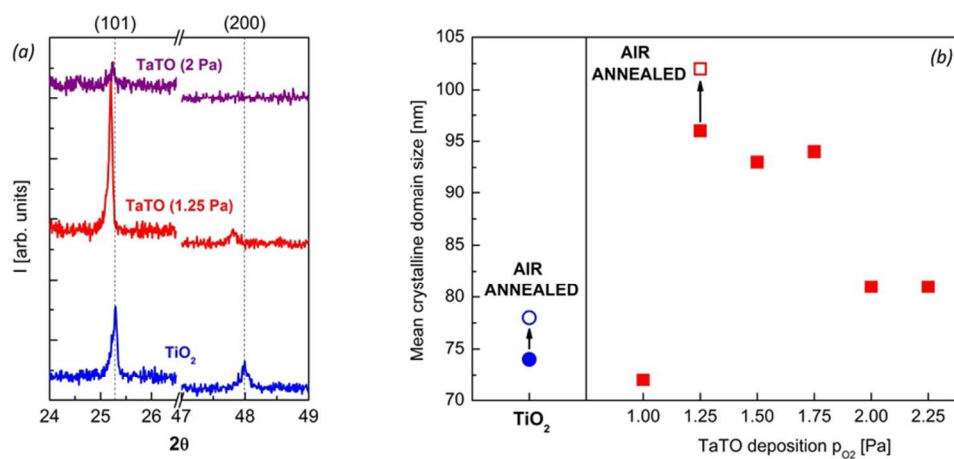


Fig 3
79x36mm (300 x 300 DPI)

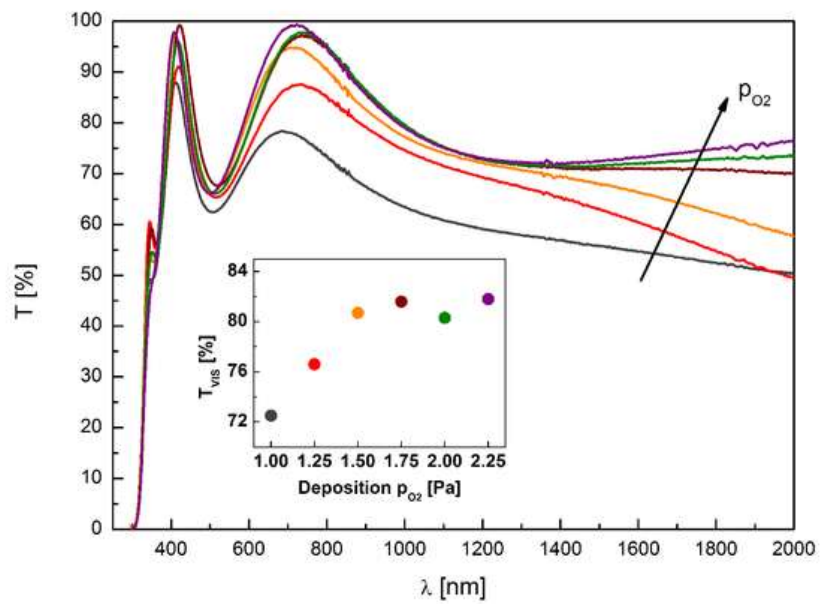
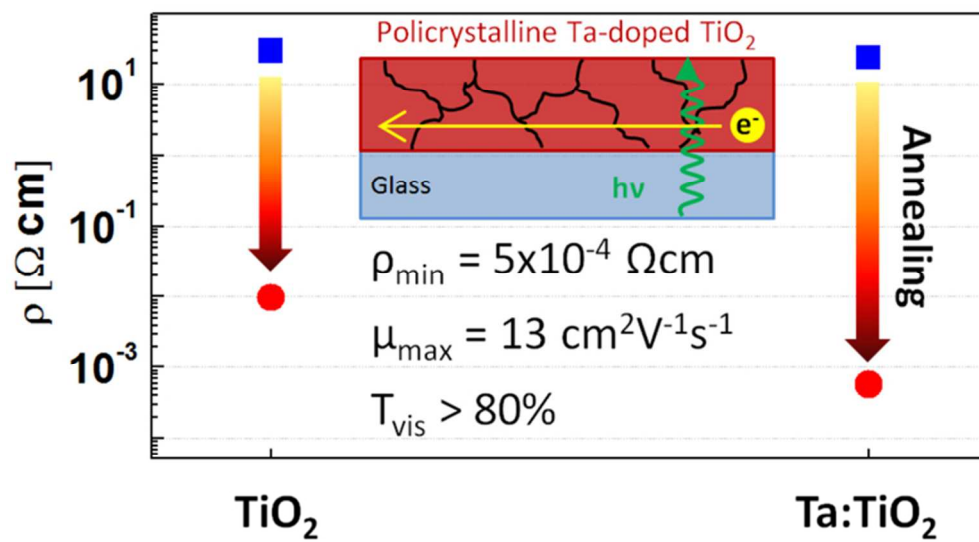


Fig 8
56x39mm (300 x 300 DPI)



TOC graphics
33x18mm (600 x 600 DPI)