



# Swelling of Positronium Confined in a Small Cavity

Giovanni Consolati<sup>1\*</sup>, Fiorenza Quasso<sup>1</sup>, Davide Trezzi<sup>2</sup>

**1** Department of Aerospace Science and Technology, Politecnico di Milano, Milano, Italy, **2** Department of Physics, Università degli Studi di Milano, Milano, Italy

## Abstract

The electron density at the positron (contact density) in the ground state positronium (Ps) formed in condensed matter is generally found to be lower than in vacuum. This is usually attributed to microscopic electric fields which polarize Ps, by acting on the two particles of the atom. In this paper we quantitatively investigate an opposite effect. It is due to the confinement of Ps in small cavities existing in the host solid (e.g. free volume in polymers), which increases the contact density. Although this phenomenon is greater, the smaller is the size of the cavity, Ps polarization seems to play anyway a predominant role.

**Citation:** Consolati G, Quasso F, Trezzi D (2014) Swelling of Positronium Confined in a Small Cavity. PLoS ONE 9(10): e109937. doi:10.1371/journal.pone.0109937

**Editor:** Mark G. Kuzyk, Washington State University, United States of America

**Received:** May 27, 2014; **Accepted:** September 13, 2014; **Published:** October 14, 2014

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**Data Availability:** The authors confirm that all data underlying the findings are fully available without restriction. All relevant data are within the paper.

**Funding:** The authors have no support or funding to report.

**Competing Interests:** The authors have declared that no competing interests exist.

\* Email: giovanni.consolati@polimi.it

## Introduction

The bound electron-positron system, positronium (Ps), formed in condensed matter shares with Ps *in vacuo* various features; the most important ones are the presence in the ground state of two sublevels (triplet: ortho-Ps, *o*-Ps, parallel spins of the electron and positron; singlet: para-Ps, *p*-Ps, antiparallel spins) characterized by different lifetimes and an energy separation between them [1]. However, there are also significant differences, first of all the possibility to annihilate with an external electron in a relative singlet state ('pickoff' process), which changes the lifetimes with respect to those in vacuum (142 ns and 0.125 ns for *o*-Ps and for *p*-Ps, respectively); in particular, the *o*-Ps lifetime in condensed matter can be shortened up to a few ns [2]. Another feature generally found for Ps in matter is a different value, with respect to vacuum Ps, of the 'contact density', that is, the electron density at the positron, represented by  $|\psi(0)|^2$ , where  $\psi$  is the Ps wavefunction. Goldanskii [3] introduced for the first time a relative contact density:

$$\eta = \frac{|\psi(0)|_{\text{matter}}^2}{|\psi(0)|_{\text{vacuum}}^2} \quad (1)$$

to account for – in a phenomenological way – the perturbation of the positron wavefunction due to the presence of the surrounding matter. A value of the relative contact density less than unity has been attributed to polarization effects on Ps [4,5] due to local electric fields acting in opposite way on the pair; this effect is expected to be relevant in the presence of polar media [6]. A second factor influencing the Ps contact density is due to the confinement of Ps in cavities, such as defects in solids: in fact, squeezing of the Ps wavefunction should increase the contact density. The smaller the cavity hosting Ps, the stronger the effect should be. Ps can be hosted in a variety of volumes. In amorphous polymers Ps is formed in the free volume holes, which represent the fraction of the total empty space present in the structure, able

to accommodate an atomic or molecular probe [7]. In zeolites Ps is located in the cages, while in porous materials (like silica gels) Ps formed in the bulk may diffuse in the structure and eventually be trapped into a pore. In molecular materials the situation is more complex. In some pure crystals (e.g. *p*-terphenyl) Ps is not formed. By doping the host crystal with sufficiently small guest molecules (e.g. anthracene), an enlarged free space is generated in the neighborhood of the guest molecule, which allows Ps formation [8]. In some molecular crystals (such as succinonitrile and adamantane) Ps can be trapped in vacancies [9]. In other molecular crystals (e.g. naphthalene) there is evidence that Ps is not trapped into vacancies, but rather in intermolecular spaces which increase with the temperature [10]. In the present work we used the word 'cavity' to mean a generic volume of the solid structure in which Ps can be formed or trapped.

The effective contact density should result from a trade-off between confinement and polarization. Experimentally, it is generally found that the contact density for Ps in matter is lower than in a vacuum, although higher values are not ruled out, in principle [5,11]. In this work we aim to quantitatively estimate the change of the contact density consequent to the Ps confinement; to this purpose, we will adapt to Ps a theoretical approach already used for hydrogen [12–14].

## Theoretical background

We suppose a Ps atom in the ground state and enclosed into a spherical cavity with radius  $r_0$ . The potential well is infinite at  $r_0$  and zero for  $r < r_0$ . Therefore, the wavefunction must vanish at  $r_0$  instead of at infinity, as it is required for vacuum Ps. This boundary condition influences only the radial part  $R(r)$  of the wavefunction, whose equation is generally written in terms of the variable  $u(r) = rR(r)$  [15]:

$$\frac{d^2u}{dr^2} + \left[ \frac{1}{4} + \frac{n}{\rho} - \frac{l(l-1)}{\rho^2} \right] u = 0 \quad (2)$$

where:  $\rho = 2\kappa r$ ;  $\kappa = \frac{\sqrt{2\mu|E|}}{\hbar}$ ;  $n = \frac{1}{a_0\kappa}$ ;  $a_0 = \frac{4\pi\epsilon_0\hbar^2}{\mu e^2}$ ;  $l$  is the azimuthal quantum number and  $\mu$  is the Ps reduced mass. By looking for a solution of the form:

$$u(\rho) = \rho^{l+1} e^{-\frac{1}{2}\rho} F(l, \rho) \quad (3)$$

the following equation is obtained:

$$\rho \frac{d^2 F}{d\rho^2} + (2l+2-\rho) \frac{dF}{d\rho} + (n-1-l)F = 0 \quad (4)$$

for the confluent hypergeometric function  $F(l+1-n, 2l+2, \rho)$  [16].

By expressing  $F(l, \rho)$  as a power series:

$$F(l, \rho) = \sum_{j=0}^{\infty} b_j \rho^j \quad (5)$$

and inserting it into eq. (4) the recursion formula between the coefficients  $b_j$  is found:

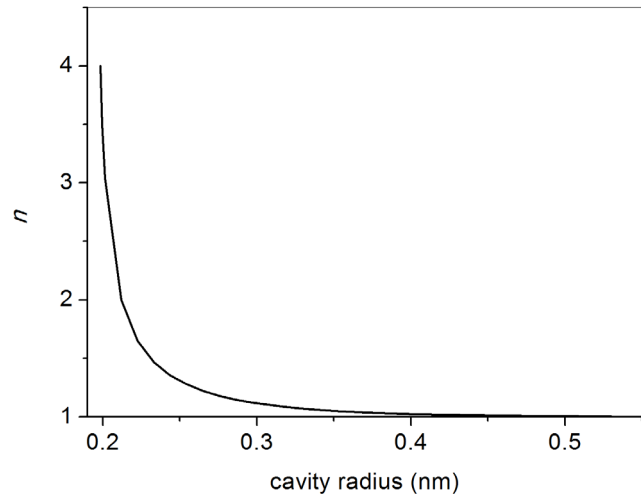
$$b_j = \frac{(j+l-n)}{j(j+2l+1)} b_{j-1} \quad (6)$$

As it is well known, when the node of the wavefunction is at  $r_0 = \infty$  the series expansion for  $F$  must break off: the number  $n$  must be integer (principal quantum number). In the present case this condition is not required and real values of  $n$  can be expected, with a corresponding shift of the energy levels [14,17]. In the above formulas  $l=0$ , as we will treat only ground state Ps.

## Results and Discussion

For  $n$  real it is necessary to find a relation between  $n$  and  $r_0$  which makes  $F(\rho)$  vanish at  $r=r_0$ ; this assures that Ps wavefunction is zero at  $r=r_0$ . This can be obtained numerically; in fact, a few dozens of terms in eq. 5 are sufficient to discriminate among values of  $F$  lower than  $10^{-6}$ . By requiring that  $F=0$  (or, more precisely,  $F < \epsilon$ , with  $\epsilon \approx 10^{-6}$ ) for several values of  $r_0$  we get the corresponding values of  $n$ . A plot of  $n$  versus  $r_0$  is shown in figure 1; it appears that  $n$  differs by at least 10% from unity only for  $r_0 < 0.30$  nm. Clearly, for large values of  $r_0$ ,  $n$  tends to its asymptotic limit, equal to 1. Knowledge of the numerical relationship between  $n$  and  $r_0$  is a prerequisite for the evaluation of the integral which appears in eq. 7 below. There is a limitation to the possible values of  $r_0$ , since by decreasing the radius of the cavity hosting Ps, the zero-point energy  $E_0$  of this last increases:  $E_0 = \hbar^2 / (8m_{\text{Ps}} r_0^2)$ . When  $E_0$  equals the Ps binding energy (amounting to  $-6.8$  eV in vacuum) Ps cannot be formed. Therefore, it is necessary that  $r_0 > 0.166$  nm, according to the adopted model.

In order to determine the relative contact density it is necessary to find the radial Ps wavefunction at the origin. Indeed,  $\psi(r) = R(r)Y_{0,0}$ , where  $Y_{0,0}$  is the spherical harmonic corresponding to  $l=0$ ; this is obviously valid for the ground Ps state, the only one of interest in the present work. It follows that  $\eta = |\psi(0)|_{\text{matter}}^2 / |\psi(0)|_{\text{vacuum}}^2 = |R(0)|_{\text{matter}}^2 / |R(0)|_{\text{vacuum}}^2$ . Now  $R(r) = u(r)/r$  and, by using eq. (3), we obtain:  $R(r) = \frac{2}{na_0} \exp\left(-\frac{r}{na_0}\right) F\left(\frac{2r}{na_0}\right)$  or, by expressing the radial



**Figure 1. Dependence of the number  $n$  on the radius  $r_0$  of the cavity trapping Ps.**

doi:10.1371/journal.pone.0109937.g001

coordinate  $r$  in the reduced unit  $r' = r/a_0$ :  $R(r') = \frac{2}{n} \exp\left(-\frac{r'}{n}\right) F\left(\frac{2r'}{n}\right)$ .

Function  $F(l, \rho)$  in eq. (5) is defined up to an arbitrary constant  $b_0$ , which can be fixed by requiring that the integral of the radial part of the wavefunction:

$$\int_0^{r_0/a_0} r'^2 |R(r')|^2 dr' = \int_0^{r_0/a_0} \frac{4}{n^2} r'^2 \exp\left(-\frac{2r'}{n}\right) F^2\left(\frac{2r'}{n}\right) dr' = A^2 \quad (7)$$

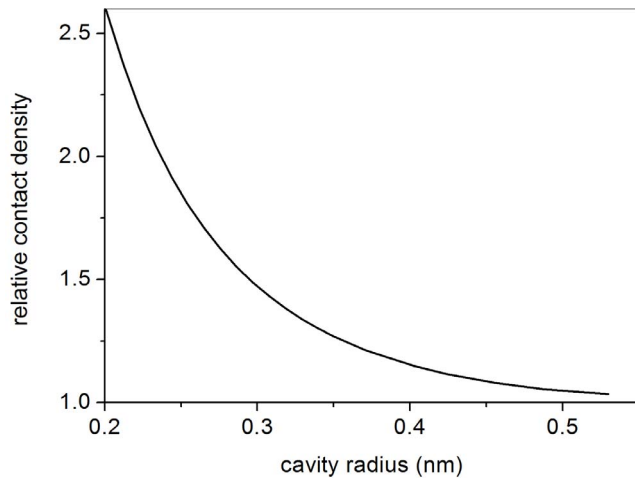
(where  $A$  depends on both  $n$  and  $r_0$ ) is normalized. This condition is satisfied when all the coefficients appearing in eq. (5) are multiplied by  $A^{-1}$ ; then  $F(0) = b_0 = 1/A$  and  $R(0) = \frac{2}{n} F(0) = \frac{2}{nA}$ . We note that  $F(0)$ , and whence  $R(0)$  has a different value for any  $r_0$ , as a consequence of the dependence of  $A$  on  $r_0$ .

Since  $|R(0)|_{\text{vacuum}}^2 = 4$  (in reduced units  $r' = r/a_0$ ), the relative contact density can be expressed as follows:

$$\eta = \frac{1}{n^2 A^2} \quad (8)$$

Calculation of the constant  $A$  appearing in eq. (7) has been carried out numerically for several values of  $r_0$  – and consequently of  $n$ , according to figure 1. The resulting relative contact density is displayed in figure 2 versus  $r_0$  and shows that  $\eta$  increases by decreasing the cavity radius  $r_0$ , when only confinement effects are taken into account. We point out that the present discussion is based on the assumption of spherical cavities. Another geometry would imply a different quantitative relationship between  $r_0$  and  $n$ . But the general trend of the relative contact density as displayed in figure 2 is not expected to change. Indeed, squeezing of the Ps wavefunction due to confinement increases its value at the origin, whatever is the adopted geometry for the cavity, if the normalization of the wavefunction is required.

Experimentally, the relative contact density in media where Ps formation is allowed can be obtained by various methods, the most



**Figure 2. Relative contact density versus the cavity radius  $r_0$ ; only confinement effects are considered.**  
doi:10.1371/journal.pone.0109937.g002

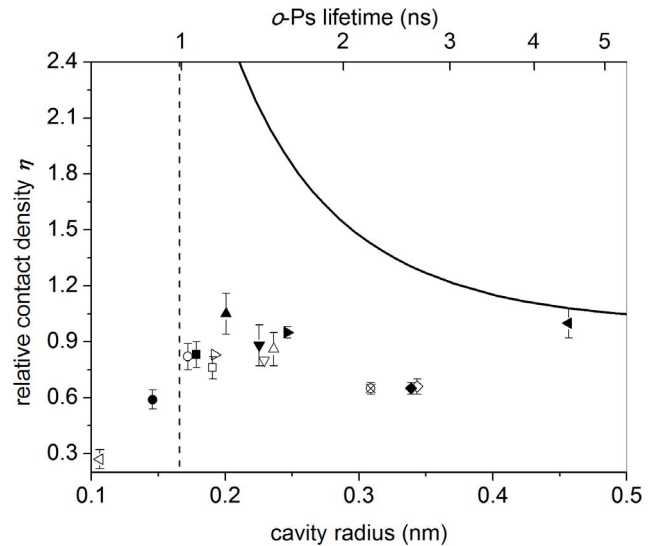
used being the ‘magnetic quenching’: in the presence of a static magnetic field the Ps sublevels with magnetic quantum number  $m=0$  are mixed and the lifetime of the corresponding triplet sublevel is quenched [18]. It is then possible to determine the relative contact density by means of a fitting procedure on the time annihilation lifetime spectra collected at different applied fields by extracting a ‘quenching ratio’ [19] or by looking for the dependence of the perturbed  $o$ -Ps lifetime on the magnetic field [20]. The three quantum yield was used, too, to obtain an estimation of the relative contact density [21]; this method turned out useful in the presence of ‘anomalous’ magnetic quenching effects [22]. Comparison between the different procedures showed good agreement, within the experimental uncertainties [21].

Unfortunately, any adopted experimental method cannot split the confinement and the polarization effects on the relative contact density. Nevertheless, it is instructive to recall the experimental results obtained. They confirm that the relative contact density is almost always found to be lower than unity, as already noted in the introduction. The only cases, at the best of our knowledge, where values significantly higher than unity were measured are those of naphthalene [23] and quartz [24]. However, such results were not confirmed [25–27]. On the basis of the previous discussion we conclude that Ps polarization overcomes the effect of confinement; the last one could be anyway significant only in small cavities (Figure 2).

Table 1 and figure 3 display experimental data as obtained in some polymers and molecular solids. These last materials are often characterized by short  $o$ -Ps lifetimes [28], of the order of 1 ns or even less; therefore, confinement effects should play a significant role. The relative contact densities found in polymers should be considered as rough estimates. Indeed, the presence of a distribution of the free volume holes sizes makes hard to extract a distribution of relative contact densities.

In figure 3 the relative contact densities are reported versus both the  $o$ -Ps lifetime (upper  $x$ -axis) and the cavity size (lower  $x$ -axis), as obtained by using the Tao-Eldrup equation [29–31]:

$$\lambda_p = \lambda_0 \left[ \frac{\Delta R}{R + \Delta R} + \frac{1}{2\pi} \sin \left( 2\pi \frac{R}{R + \Delta R} \right) \right] \quad (9)$$



**Figure 3. Relative contact density and  $o$ -Ps lifetime as measured in some molecular solids and polymers (symbols are explained in Table 1).**  
doi:10.1371/journal.pone.0109937.g003

This equation is based on the assumption of a spherical cavity - like the model used in the present work - with effective radius  $R$ . For convenience of calculations the depth is assumed as infinite, but the radius is increased to  $R + \Delta R$ ,  $\Delta R$  being an empirical parameter which describes the penetration of Ps wave function into the bulk. The electron density is supposed to be zero inside the cavity and constant from  $R$  to  $R + \Delta R$ . In eq. (9)  $\lambda_0 = 2 \text{ ns}^{-1}$  is the spin-averaged annihilation rate of  $p$ -Ps ( $8 \text{ ns}^{-1}$ ) and  $o$ -Ps ( $1/142 \text{ ns}^{-1}$ ) in vacuum,  $\lambda_p$  is the pickoff decay rate. The measured  $o$ -Ps lifetime  $\tau_3$  is the reciprocal of the total decay rate  $\lambda_3$ , sum of the pickoff decay rate and the intrinsic decay rate  $\lambda_i$  ( $1/142 \text{ ns}^{-1}$ ):

$$\tau_3 = 1/\lambda_3 = 1/(\lambda_p + \lambda_i) \quad (10)$$

The  $\lambda_i$  contribution, included in Eq. (10) for the sake of completeness, is negligible when  $o$ -Ps lifetimes are of the order of a few ns. The vertical dashed line drawn in figure 3 represents the minimum value of the cavity radius (0.166 nm) compatible with the Ps zero point energy limitation, as discussed above and dependent on the adopted cavity model.

A reduction of the relative contact density  $\eta$  with respect to unity, corresponding to a swelling of Ps atom, is observed in figure 3, on the average. Therefore, polarization is the main effect on Ps contact density, even when the atom is confined in small cavities. To point out this last conclusion we reported in figure 3 also the behaviour of  $\eta$  due to bare confinement, already shown in figure 2. Qualitatively, polarization effects can be explained in terms of Van der Waals interactions between Ps and molecules of the medium. Although Ps does not have an electric dipole moment in its ground state, when the surrounding molecules are polar a Ps polarization is expected, through an induced dipole force (Debye force) between the permanent dipole and the Ps induced dipole. Nevertheless, also in non-polar media instantaneous dipole-induced dipole forces (London dispersion forces) can occur. Such forces are weak, but the effect on Ps can be noticeable when the Ps-molecule distance is very short, as in the case of a small sized cavity; this could justify the predominance of polarization effects

**Table 1.** Relative contact density  $\eta$  and *o*-Ps lifetime  $\tau$  in some molecular solids and polymers.

| Material                            | $\tau$ (ns) | $\eta$    | Symbol in fig.3 | Ref  |
|-------------------------------------|-------------|-----------|-----------------|------|
| PTFE <sup>a</sup>                   | 4.47±0.09   | 1.00±0.08 | ◀               | [35] |
| PPA <sup>b</sup>                    | 2.66±0.05   | 0.66±0.04 | ◇               | [22] |
| PE <sup>c</sup>                     | 2.60±0.05   | 0.60±0.06 | ◆               | [35] |
| PMMA <sup>d</sup>                   | 2.23±0.04   | 0.65±0.03 | ⊗               | [22] |
| a-SiO <sub>2</sub>                  | 1.59±0.02   | 0.95±0.03 | ▶               | [27] |
| Octadecane                          | 1.50±0.02   | 0.86±0.09 | Δ               | [36] |
| p-terphenyl (doped with anthracene) | 1.44        | 0.80      | ▽               | [37] |
| Butyl-PBD <sup>e</sup>              | 1.41±0.07   | 0.88±0.11 | ▼               | [38] |
| PPD <sup>f</sup>                    | 1.22±0.02   | 1.05±0.11 | ▲               | [38] |
| p-terphenyl (doped with chrysene)   | 1.16        | 0.83      | ▷               | [37] |
| Byphenil                            | 1.15±0.02   | 0.76±0.06 | □               | [26] |
| PPO <sup>g</sup>                    | 1.07±0.03   | 0.78±0.04 | ■               | [38] |
| Naftalene                           | 1.03±0.01   | 0.82±0.07 | ○               | [26] |
| Acenaphtene                         | 0.88±0.01   | 0.59±0.05 | ●               | [39] |
| Thorium phosphate                   | 0.70        | 0.32–0.22 | ◁               | [37] |

<sup>a</sup>Polytetrafluoroethylene.

<sup>b</sup>Atactic polypropylene.

<sup>c</sup>Polyethylene.

<sup>d</sup>Poly(methyl methacrylate).

<sup>e</sup>2,4-tert-Butylphenyl)-5-(4-byphenyl)1,3,4-oxadiazol.

<sup>f</sup>2,5-Diphenyl-1,3,4-oxadiazol.

<sup>g</sup>2,5-Diphenyl-oxazol.

doi:10.1371/journal.pone.0109937.t001

with respect to the squeezing of Ps wavefunction due to confinement.

## Conclusions

The change of the Ps contact density with respect to Ps in a vacuum, as generally observed, should be attributed to the opposite effects of Ps confinement and Ps polarization, this last being induced by the surrounding molecules. The last effect is stronger than the confinement occurring when Ps is trapped in a small cavity, which squeezes Ps, by increasing the value of its wavefunction at the origin. The results here obtained are based on a simplified model of the cavity, which is supposed to be spherical. Anyway, the experimental relationship between contact density and *o*-Ps lifetime does not imply any particular assumption on the geometry of the cavity: *o*-Ps lifetime can be considered a (non-linear) probe of the cavity size, since by decreasing this last the lifetime decreases, too, due to the increased pickoff rate. Generally, real cages trapping Ps are not spherical; this is not a severe constraint, at least as far as the relationship between *o*-Ps lifetime

and cavity size is concerned [32]. Use of a non-spherical geometry [33,34] would imply a different solution of the Schrödinger equation (*e.g.* in cylindrical coordinates) with respect to the treatment here adopted. In spite of this limitation, the conclusions are not expected to change by using other geometries.

## Acknowledgments

G.C. is grateful to prof. Tomasz Goworek (Institute of Physics, Maria Curie Skłodowska University, Lublin, Poland) and to prof. Fabrizio Castelli (Department of Physics, University of Milano, Milan, Italy) for stimulating discussions.

## Author Contributions

Conceived and designed the experiments: GC FQ DT. Performed the experiments: GC DT. Analyzed the data: GC FQ DT. Contributed reagents/materials/analysis tools: GC FQ. Wrote the paper: GC FQ DT. Designed the software used in the analysis: GC FQ. Bibliographic researches: FQ DT.

## References

- Berko S, Pendleton HN (1980) Positronium. *Ann Rev Nucl Part Sci* 30: 543–581.
- Jean YC, Mallon PE, Schrader DM (2003) Introduction to positron and positronium chemistry. In: Jean YC, Mallon PE, Schrader DM, editors. *Principles and applications of positron and positronium chemistry*. World Scientific:Singapore. p. 3.
- Goldanskii VI (1968) *Physical Chemistry of the Positron and Positronium*. Atomic Energy Review 6: 3–148.
- Dupasquier A (1983) Quasi-positronium systems in solids. In: Brandt W, Dupasquier A, editors. *Positron Solid-State Physics*. North-Holland: Amsterdam. p. 510.
- McMullen T, Stott MT (1983) Dependence of the positronium hyperfine interaction on the environment – a simple model. *Can J Phys* 61: 504–507.
- Consolati G, Quasso F (2013) Polarization of positronium in amorphous polar polymers: a case study. *J Chem Phys* 139: 204901.
- Hofmann D, Entrialgo-Castano M, Lebre A, Heuchel M, Yampolskii Y (2003) Molecular Modeling Investigation of free Volume Distributions in Stiff Chain Polymers with Conventional and Ultrahigh Free Volume: Comparison between Molecular Modeling and Positron Lifetime Studies. *Macromolecules* 36: 8528–8538.
- Goworek T, Rybka C, Wawryszczuk J (1978) Positronium in mixed molecular crystals. *Phys. Stat. Solidi (b)* 89: 253–259.
- Eldrup M, Pedersen NJ, Sherwood JN (1979) Positron annihilation study of defects in succinonitrile. *Phys. Rev. Lett.* 43: 1407–1410.

10. Goworek T, Rybka C, Wasiewicz R, Wawryszczuk J (1982) Positronium trapping at thermally produced defects in naphthalene. *Phys. Stat. Solidi.(b)* 113: K9–K13.
11. Brandt W, Berko S, Walker WW (1960) Positronium decay in molecular substances. *Phys Rev* 120: 1289–1295.
12. Michels A, de Boer J, Bijl A (1937) Remarks concerning molecular interaction and their influence on the polarizability. *Physica* 4: 981–994.
13. Sommerfeld A, Welker H (1938) Künstliche Grenzbedingungen beim Keplerproblem. *Ann Physik* 32: 56–65.
14. De Groot SR, Ten Seldam CA (1946) On the energy levels of a model of a compressed hydrogen atom. *Physica* 12: 669–682.
15. Griffiths DJ (2005) Introduction to quantum mechanics. Pearson Prentice Hall, Upper Saddle River: New Jersey, cap. 4.
16. Slater IJ (1960) Confluent hypergeometric functions, Cambridge University Press: Cambridge, UK.
17. Goldmann S, Joslin C (1992) Spectroscopic properties of an isotropically compressed hydrogen atom. *J Phys Chem* 96: 6021–6027.
18. Halpern O (1954) Magnetic quenching of the positronium decay. *Phys Rev* 94: 904–907.
19. Consolati G (1996) Magnetic quenching of positronium. *J Radioanal Nucl Chem* 210: 273–292.
20. Rochanakij S, Schrader DM (1988) Magnetic quenching of positronium in organic solutions. *Radiat Phys Chem* 32: 557–561.
21. Consolati G, Quasso F (1991) The experimental determination of the Ps contact density in condensed matter. *Appl Phys A52*: 295–298.
22. Consolati G, Quasso F (1988) A comparison between the magnetic quenching of positronium in atactic polypropylene and that in isotactic polypropylene, *J Phys C: Solid State Phys.* 21: 4143–4151.
23. Gorniak W, Goworek T (1991) Electron contact density for positronium in naphthalene. *Chem Phys Lett* 177: 23–26.
24. Bisi A, Consolati G, Quasso F, Zappa L (1988) Positroniumlike system in quartz. *Il Nuovo Cimento* 10D: 1069–1077.
25. Wawryszczuk J, Goworek T (1996) Does the p-Ps lifetime agree with expectations from magnetic quenching? *J Radioanal Nucl Chem* 210: 161–164.
26. Consolati G, Quasso F (1994) Magnetic quenching of the three photon annihilation rate in some molecular solids. *Phys Rev B* 50: 5848–5853.
27. Nagashima Y, Nagai Y, Hyodo T (2001) Positronium contact density in crystalline and amorphous SiO<sub>2</sub>. *Mat Sci Forum* 363–365: 567–569.
28. Goworek T (1987) Defects in molecular solids – positron studies. *Phys Stat Solidi A - Appl Res* 102: 511–526.
29. Tao S J (1972) Positronium Annihilation in Molecular Substances. *J Chem Phys* 56: 5499–5510.
30. Eldrup M, Lightbody D, Sherwood N (1981) The Temperature Dependence of Positron Lifetimes in Solid Pivalic Acid. *J Chem Phys* 63: 51–58.
31. Nakanishi H, Wang SJ, Jean YC (1988) Microscopic surface tension studied by positron annihilation. In: Sharma SC, editor. *Positron Annihilation Studies of Fluids*, World Scientific: Singapore, p. 292–298.
32. Consolati G (2002) Positronium trapping in small voids: influence of their shape on positron annihilation results. *J Chem Phys* 117: 7279–7283.
33. Jasinska B, Koziol AE, Goworek T (1996) Ortho-positronium lifetimes in nonspherical voids. *J Radioanal Nucl Chem* 210: 617–623.
34. Consolati G, Quasso F, Simha R, Olson BG (2005) On the Relation between Positron Annihilation Lifetime Spectroscopy and Lattice-Hole Theory Free Volume. *J Polym Sci B* 43: 2225–2229.
35. Consolati G, Quasso F (1990) On the origin of the intermediate component in the positron lifetime spectra in polymers. *Appl Phys A50*: 43–48.
36. Consolati G, Quasso F (1990) Anomalous magnetic quenching of quasi-positronium in solid octadecane. *J Phys: Cond Matter* 2: 3941–3946.
37. Goworek T, Badia A, Duplatre G (1994) Normal and anomalous positronium states in ionic and molecular solids investigated *via* magnetic field effects. *J Chem Soc Faraday Trans* 90: 1501–1506.
38. Consolati G, Gambarà N, Quasso F (1991) Positron annihilation in some organic scintillators: magnetic quenching and three gamma spectroscopy results. *Z Phys D* 21: 259–264.
39. Consolati G, Quasso F (1988) Evidence for positronium formation in some aromatic hydrocarbons. *Solid State Comm* 67: 1113–1117.