



Functionalization of graphene related materials with biosourced C-3 and C-6 building blocks. From synthesis to applications

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Politecnico di Milano, Department of Chemistry, Materials and Chemical Engineering “G. Natta”

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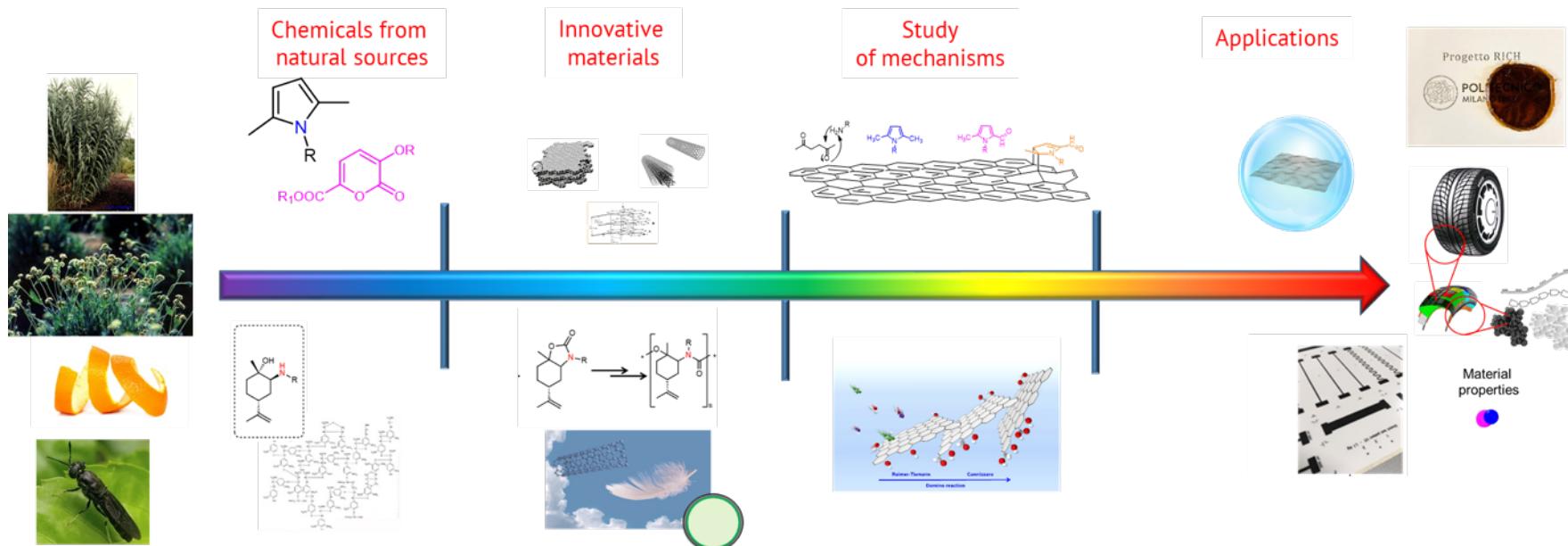
ISCaMaP

*Innovative Sustainable Chemistry and Materials and Proteomics
Group*

Politecnico di Milano, Department of Chemistry, Materials and Chemical Engineering “G. Natta”

ISCaMaP

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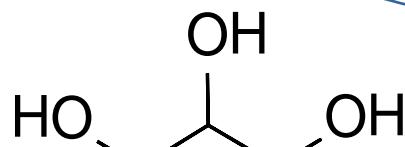


ISMATERIALS group

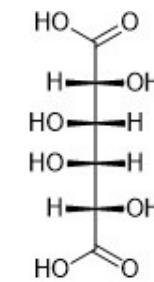
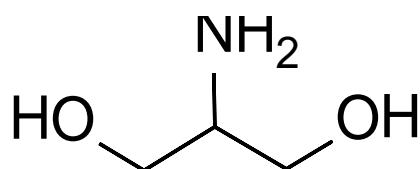
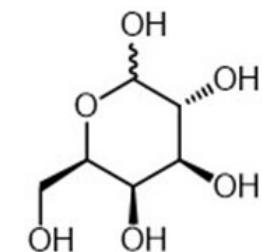
ISMATERIALS group

instagram: @ismaterials.polimi

The ISCaMaP Group. Some examples



Glycerol, Sugars



...

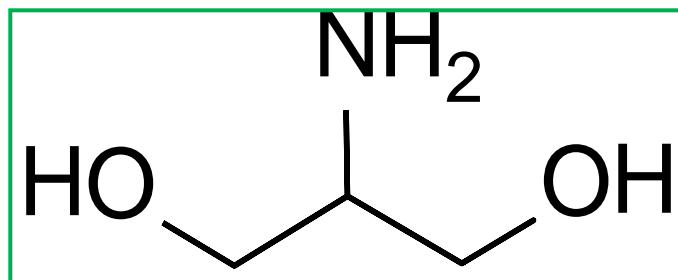
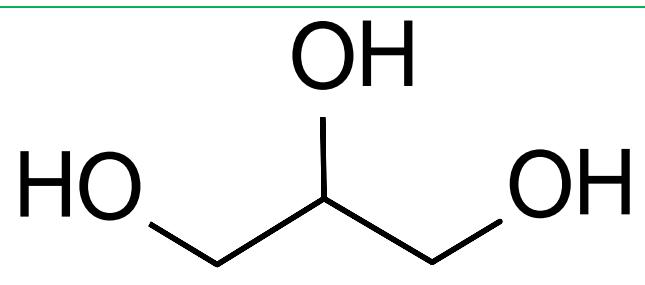


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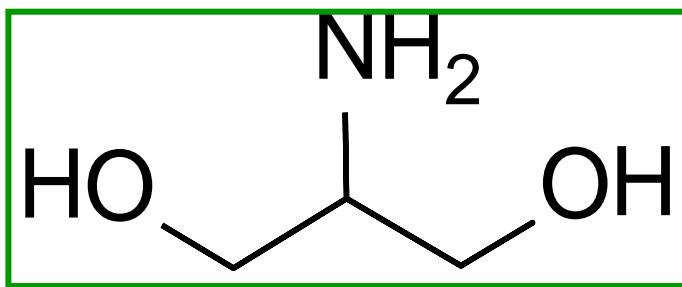
C3 building blocks



From glycerol to serinol



Selection of the building block: serinol



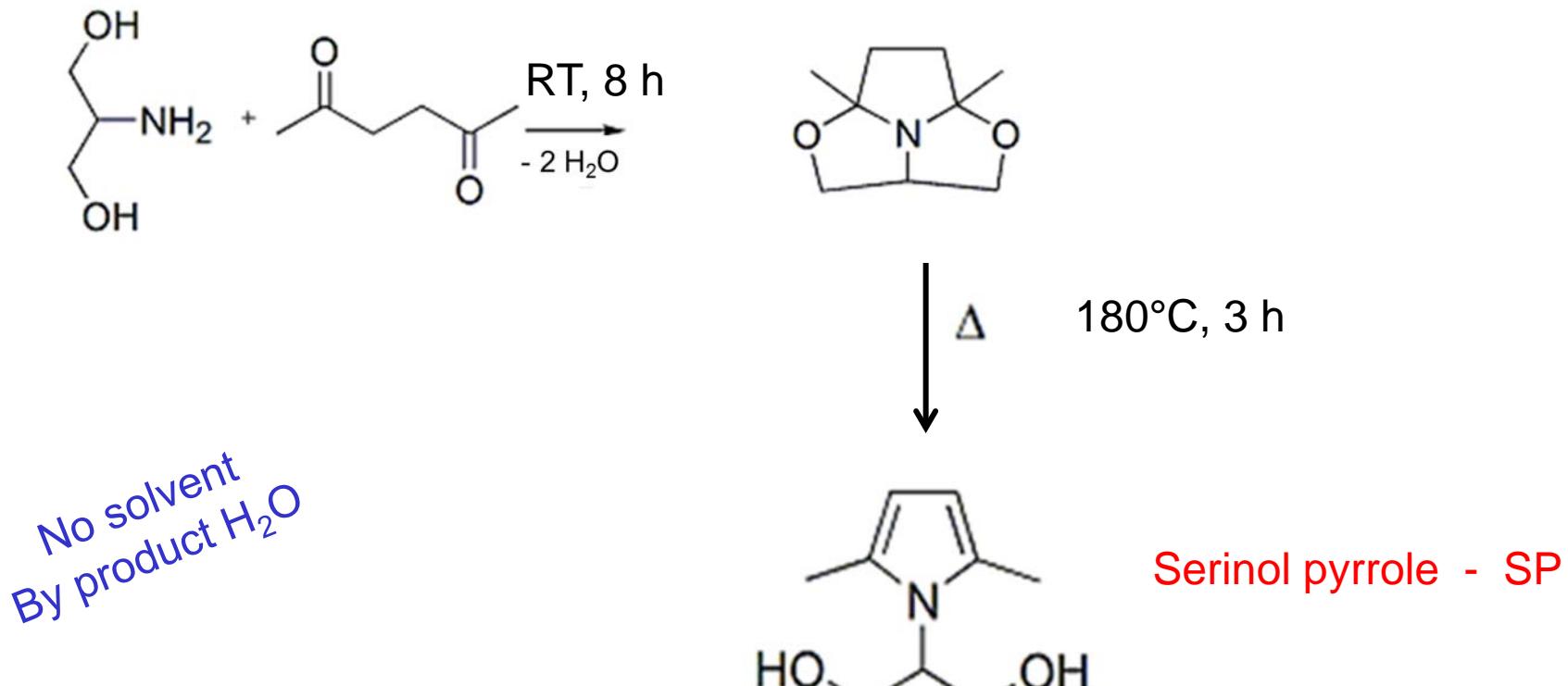
- ☞ Starting building block for many reaction pathways: many derivatives
- ☞ Chemoselectivity



Reactions of the amino group with carbonyl compounds

Reaction of serinol with dicarbonyl compound

Sustainable
synthesis



2-(2,5-dimethyl-1*H*-pirrol-1-yl)-1,3-propanediol

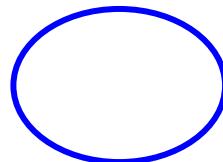
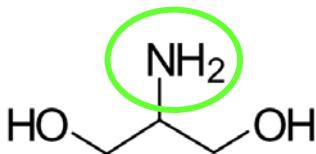
V. Barbera, A.Citterio, M. Galimberti, G. Leonardi, R. Sebastiani, S.U.Shisodia, A.M. Valerio WO 2015 189411 A1

M. Galimberti, V. Barbera, A. Citterio, R. Sebastiani, A. Truscello, A. M. Valerio, L. Conzatti, R. Mendichi, *Polymer*, vol 63, 20 April 2015, Pages 62–70

M. Galimberti, V. Barbera, S. Guerra, L. Conzatti, C. Castiglioni, L. Brambilla, A. Serafini,, *RSC Adv.*, 2015, 5, 81142-81152 DOI: 10.1039/C5RA11387C

V. Barbera, S. Musto, A. Citterio, L. Conzatti, M. Galimberti,, *eXPRESS Polymer Letters* 2016, 10 (7) 548–558

Neat synthesis of Serinol pyrrole



- ☞ Yield: at least 96%
- ☞ Atom efficiency: 85%
- ☞ Easy procedure
- ☞ No solvent
- ☞ By product: H₂O

Pyrrole compounds (PyC) from neat Paal Knorr reaction

Same reaction conditions used for SP

PyC

Yield %

80

75

62

73

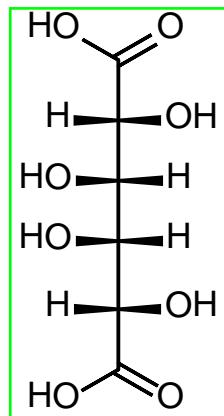
80

70

C6 building blocks



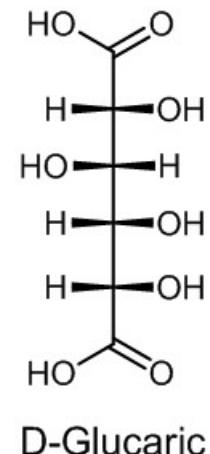
Chemicals from sugars. Galactaric Acid



IUPAC: (2R,3S,4R,5S)-2,3,4,5-tetrahydroxyhexanedioic acid
Common nomenclature: **mucic acid**
Formula: C₆H₁₀O₈
210.14 Da

Production (ton/a) 38,000 (4-10 \$/kg)*

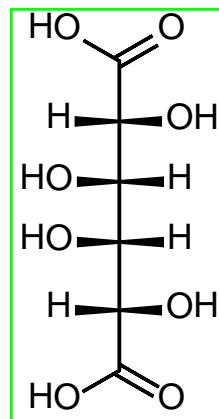
- ☞ Symmetrical structure: achiral
- ☞ Platform molecule
- ☞ Not toxic, Biocompatible
- ☞ 25 times less expensive than glucaric acid



Production (ton/a) 100,000 (6-40 \$/kg)*

J. LI – PhD thesis 2019 – Politecnico di Milano

Galactaric Acid



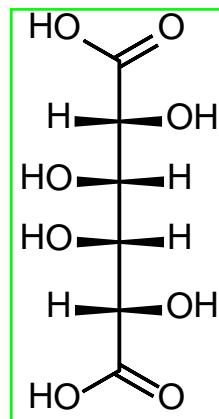
IUPAC: (2R,3S,4R,5S)-2,3,4,5-tetrahydroxyhexanedioic acid
Common nomenclature: mucic acid
Formula: $C_6H_{10}O_8$
210.14 Da

However ...

pretty high price



Galactaric Acid



IUPAC: (2R,3S,4R,5S)-2,3,4,5-tetrahydroxyhexanedioic acid
Common nomenclature: mucic acid
Formula: $C_6H_{10}O_8$
210.14 Da

Improvements in the production techniques would unlock its potential as a platform chemical

pretty high price



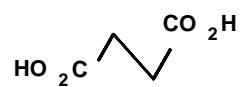
low price



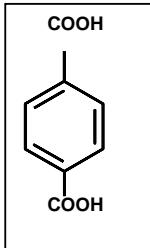
Galactaric Acid - Platform molecule

Derivatives from galactaric acid

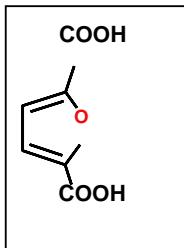
Chemicals from sugar - Target molecules



succinic acid

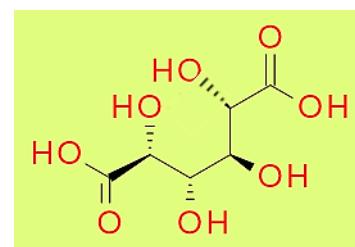
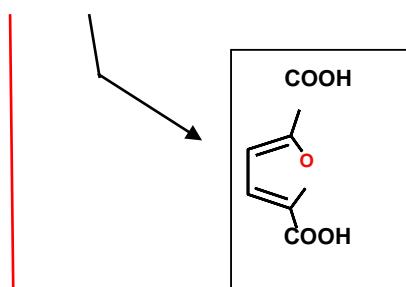
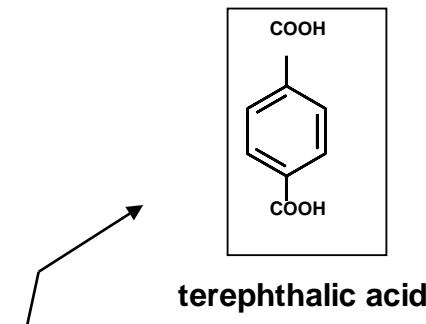
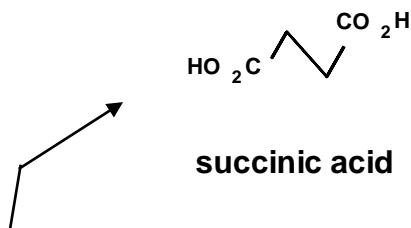


terephthalic acid



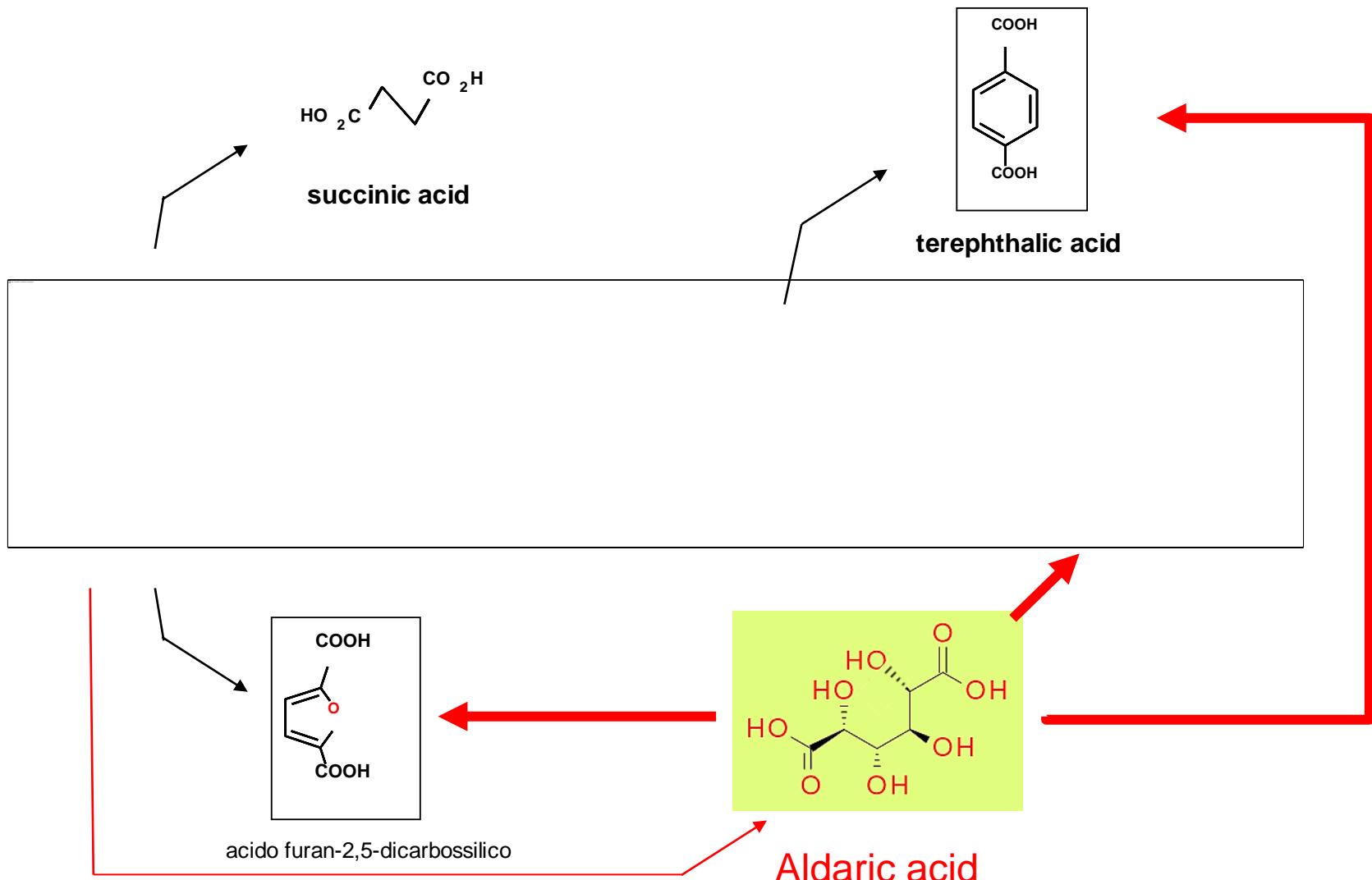
acido furan-2,5-dicarbossilico

Dicarboxylic Acids from Hydrolyzed Biomasses

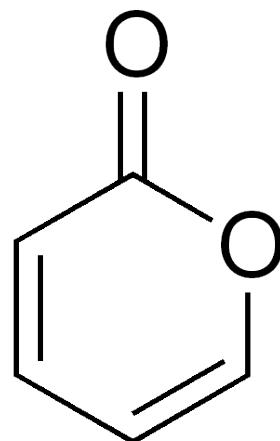


tetrahydroxyl derivatives
of adipic acid

Dicarboxylic Acids from Hydrolyzed Biomasses



From aldaric acids to pyrones



Synthesis of Pyrone Derivatives from Aldaric Acids @ ISCaMaP

3-Acetoxy-2-oxo-2*H*-
pyran-6-carboxylic salt
(GAB-OAc)

- Easy procedure
- No solvent
- No catalyst
- High Conversion
- High Atom efficiency

Synthesis of Pyrone Derivatives from Aldaric Acids @ ISCaMaP

First step



Yield = 99%; A.E. = 77%

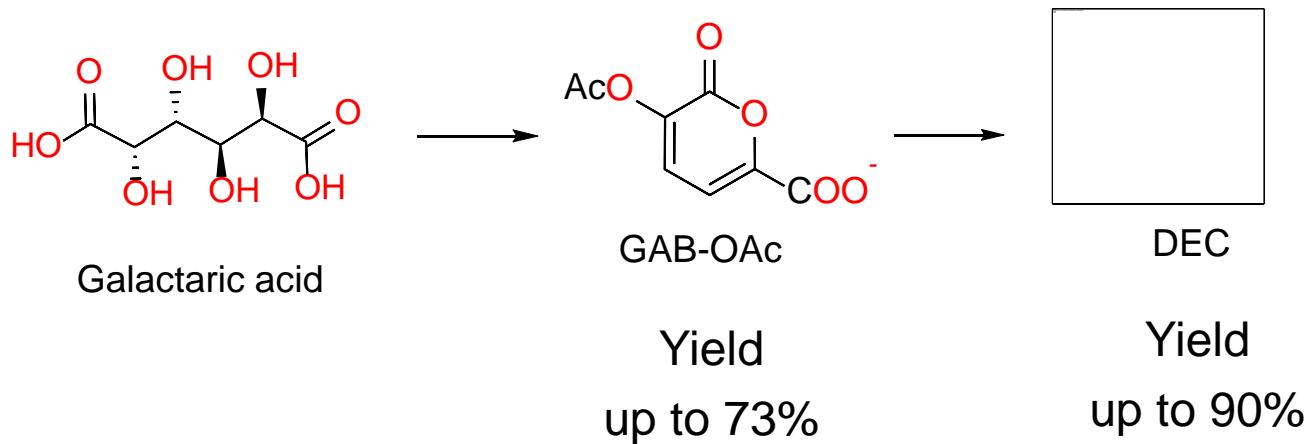
Second step



Yield = 65%; A.E. = 62%

Leonardi G., Li, J., Righetti, G. I. C., Truscello, A. M., Gambarotti, C., Terraneo, G., Citterio, A., Sebastiano, R.,
Eur. J. Org. Chem. **2020**, 241–251

Synthesis of Pyrone Derivatives from Aldaric Acids @ ISCaMaP



Synthesis of Pyrone - Scale up



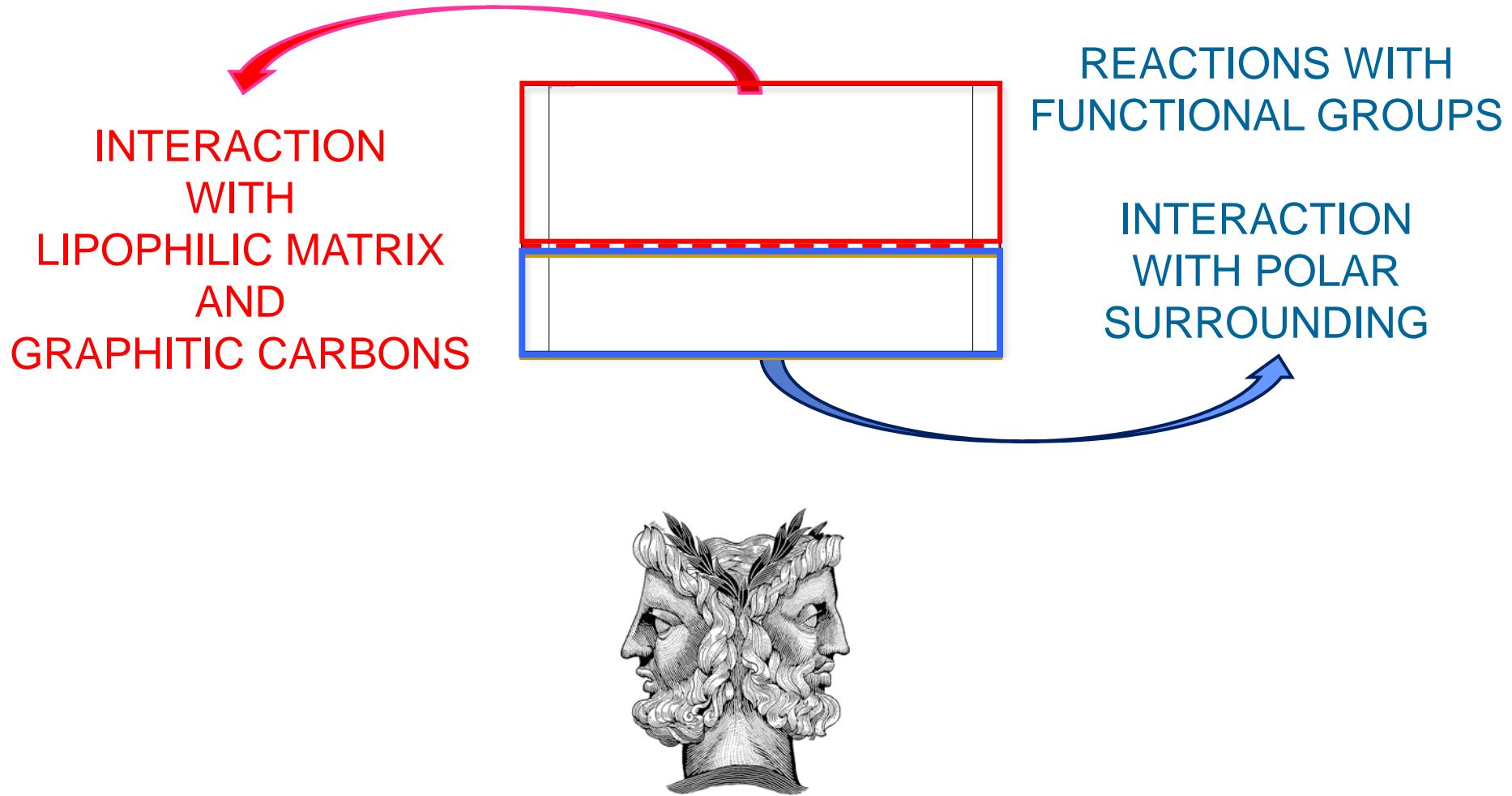
One Pot

2 hours

Yield = 75%

Functionalization of sp² carbon allotropes

Serinolpyrrole: Janus molecule

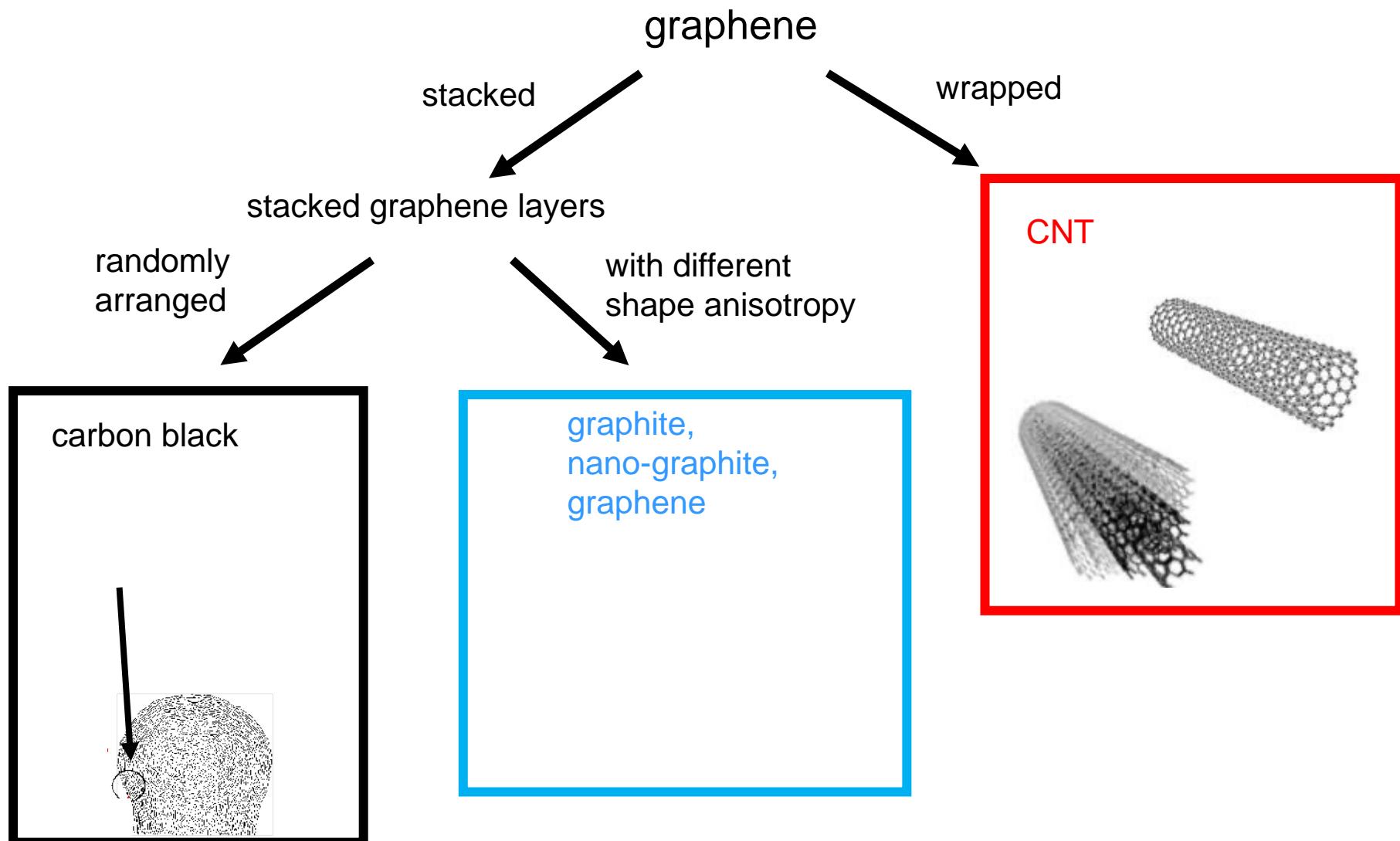


Objective of the research

Sustainable and versatile functionalization
of unperturbed graphene layers



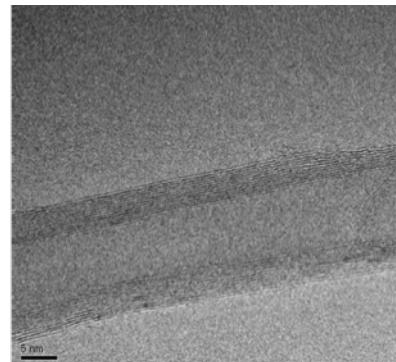
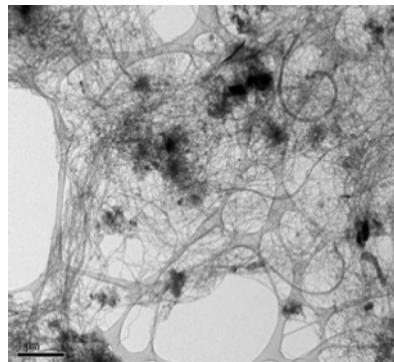
Carbon fillers from a layer of sp²-bonded carbon atoms



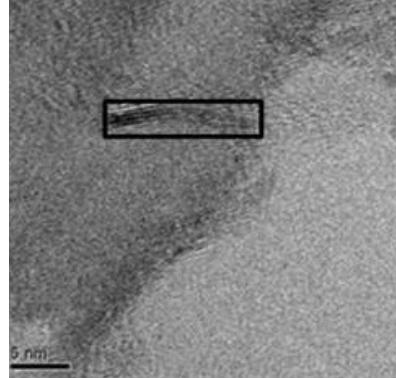
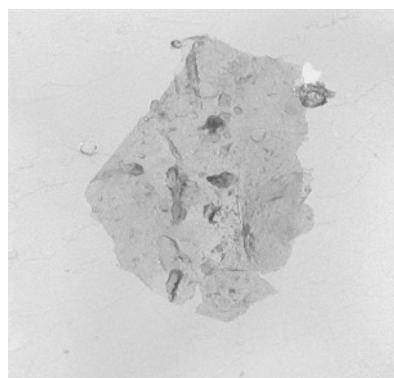
Carbon allotropes (CA)

CB

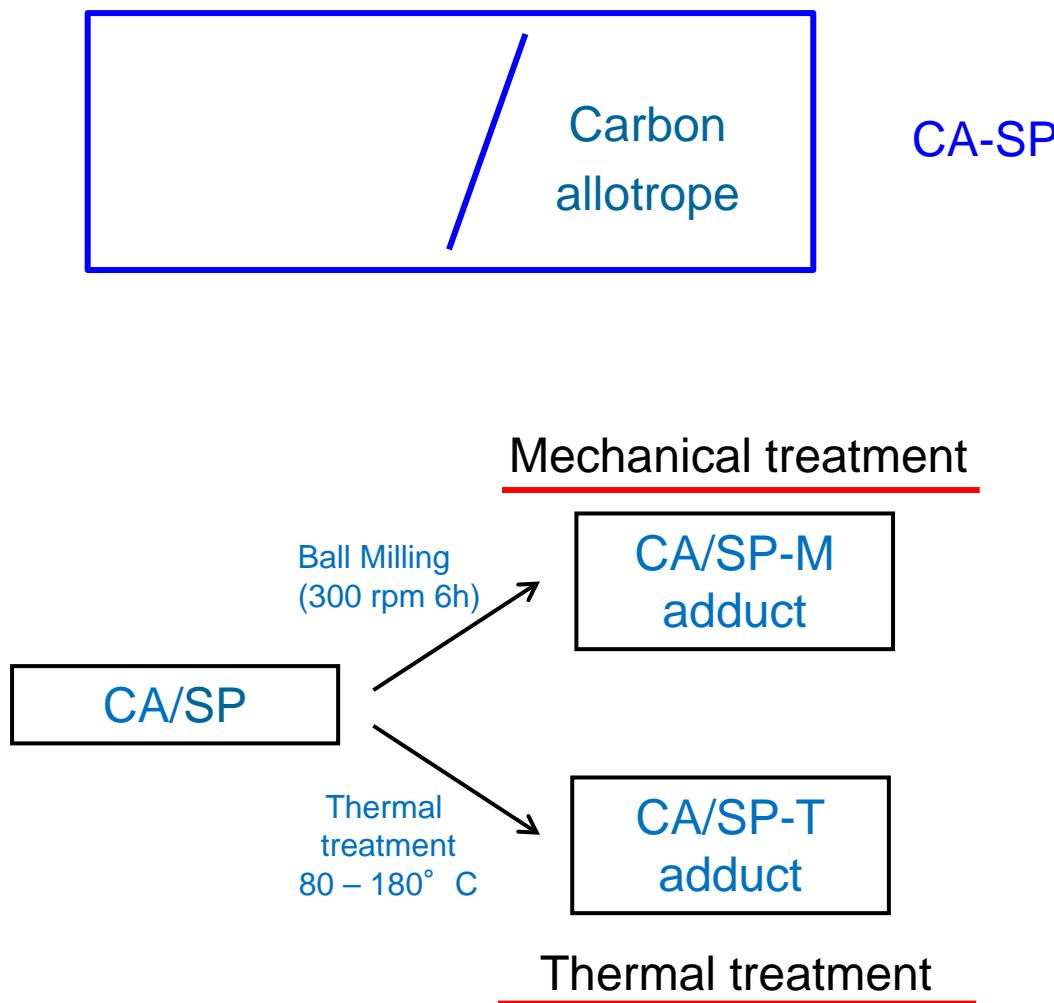
CNT



FEW LAYERS
GRAPHENE

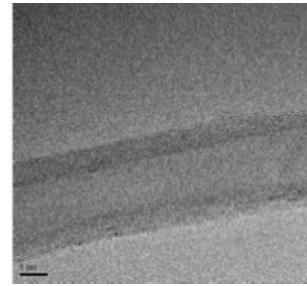
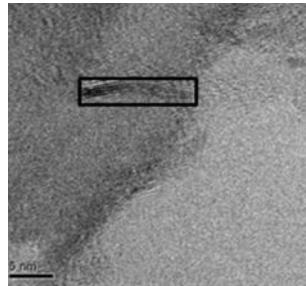


Adducts of SP with CA - Preparation



Barbera, V., Citterio, A., Galimberti, M., Leonardi, G., Sebastiani, R., Shisodia, S.U., Valerio A.M. WO 2015 189411 A1
Galimberti, M., Barbera, V., Guerra, S., Conzatti, L., Castiglioni, C., Brambilla, L., A. Serafini, RSC Advances, 5(99), (2015) 81142-81152

High yield functionalization!



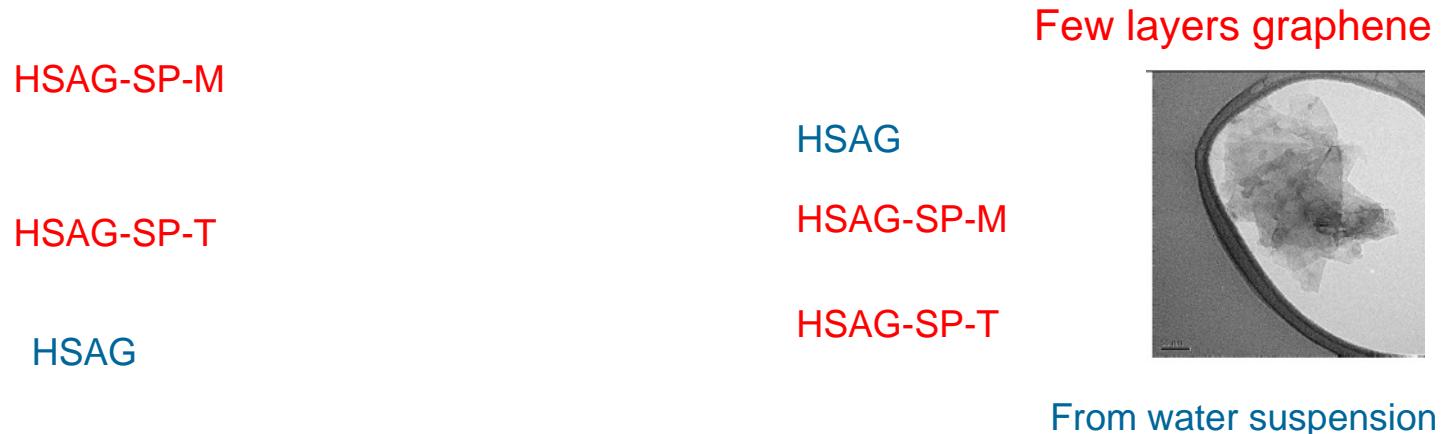
BET Surface area: [m ² /g]	300	77	275
Initial functional groups: [mmol/g]	1.7	0.9	2.0
Yields (%)*:	96	82	92

*was estimated through:

$$\text{Functionalization Yield (\%)} = 100 * \frac{\text{SP mass \% in (CA-SP adduct) after acetone washing}}{\text{SP mass \% in (CA-SP adduct) before acetone washing}}$$

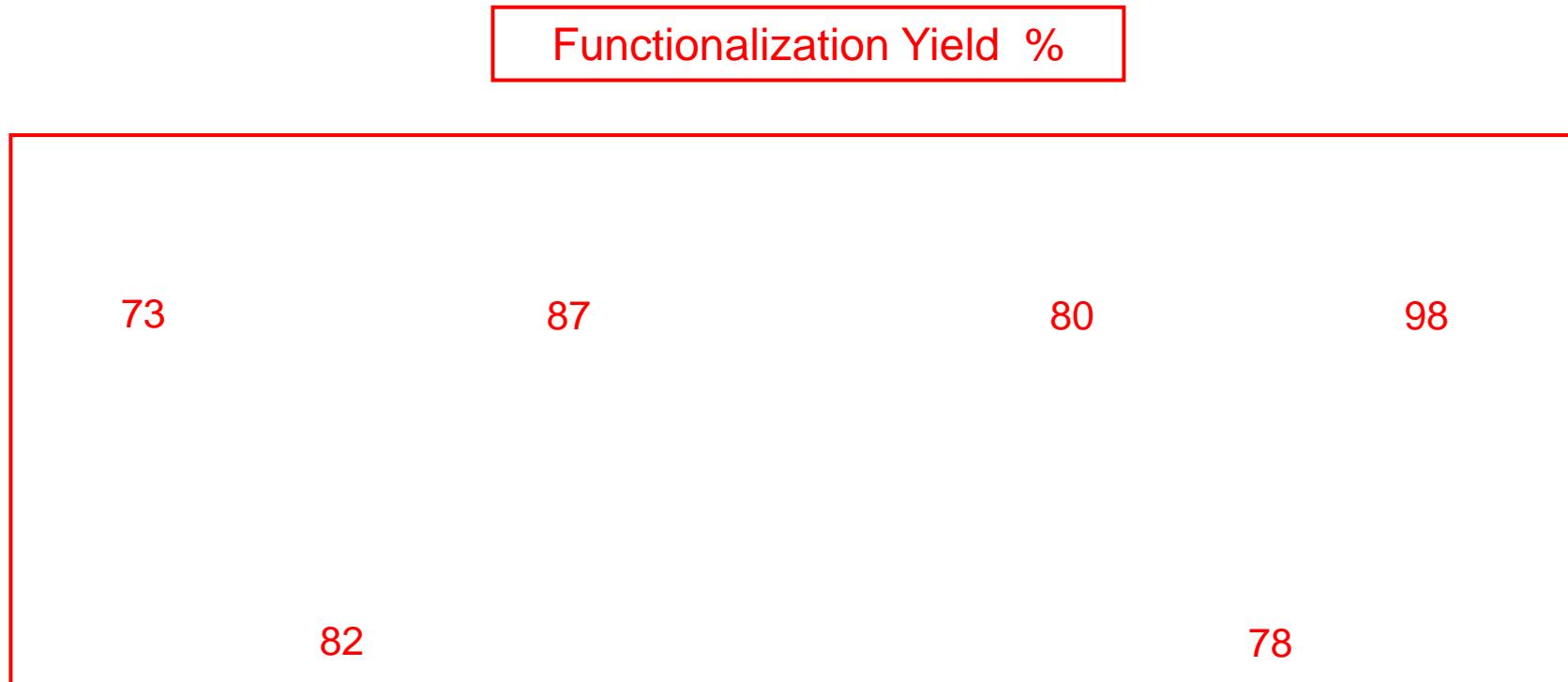
Adducts of SP with HSAG

- ☞ Functional groups up to 20%
- ☞ In plane order substantially unaltered
- ☞ No expansion of interlayer distance



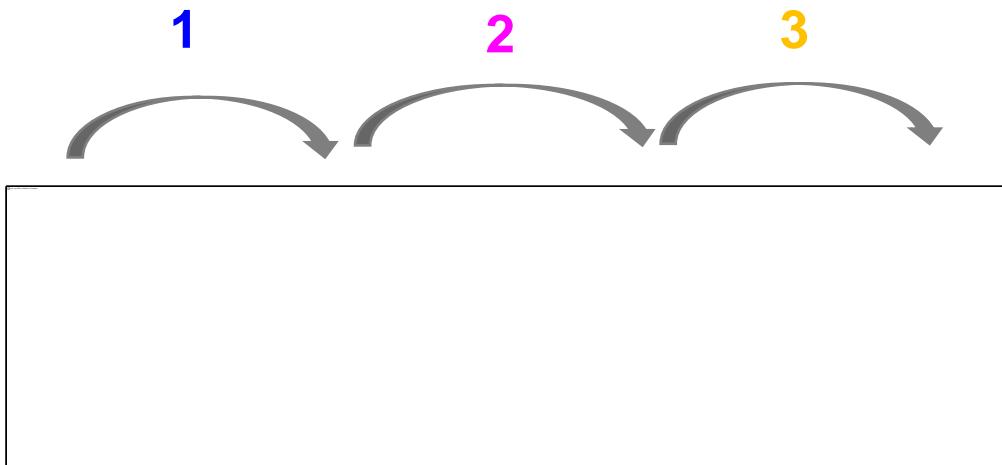
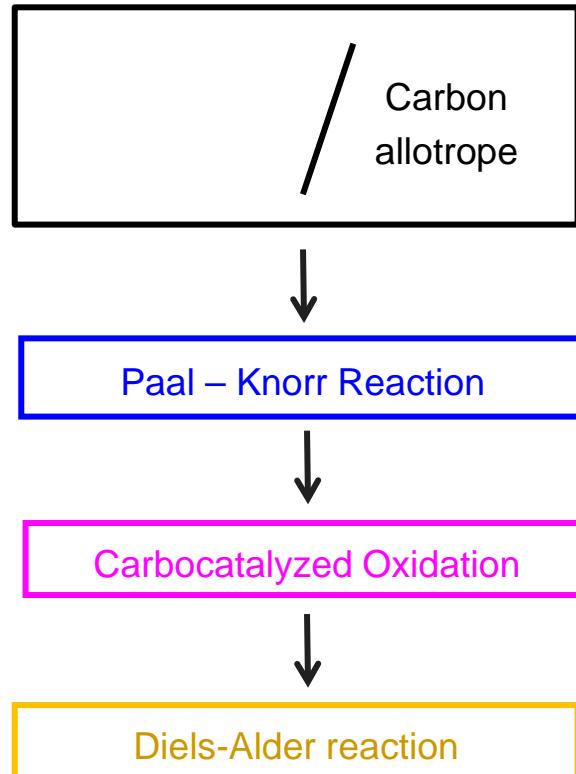
Results from elemental, TGA, IR, XPS, Raman, XRD, HRTEM analysis

HSAG / PyC adducts



Mechanism for the formation of CA/PyC adducts

Domino reaction



Thanks to the carbon allotrope!

- ☞ **Support:** absorption of pyrrole ring thanks to π - π interaction
- ☞ **Oxidation catalyst:** protection of pyrrole ring and oxidation of lateral substituent
- ☞ **Substrate** for the cycloaddition reaction, i.e. for functionalization



The CA/PyC adducts

- ☞ Functional group:
from few % to 20%
- ☞ Functionalization yield:
from 85% to quantitative
- ☞ Covalent bond
between functional group
and carbon allotrope
- ☞ Bulk structure of graphitic materials:
substantially unaltered



V. Barbera, A. Citterio, M. Galimberti, G. Leonardi, R. Sebastiani, S.U. Shisodia, A.M. Valerio. [US10329253B2](#)

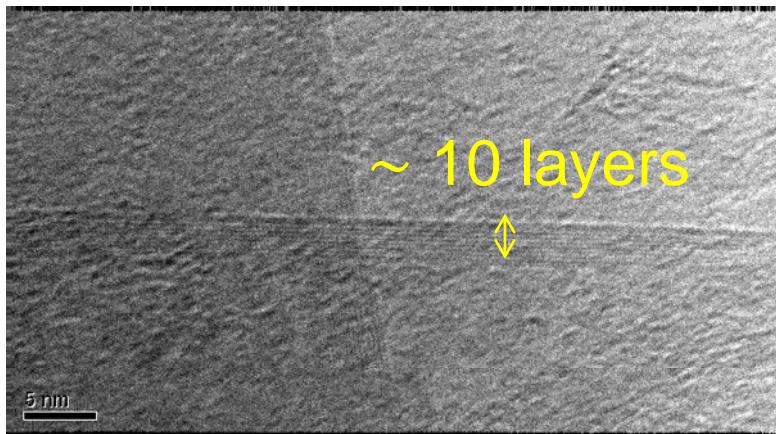
M. Galimberti, V. Barbera, R. Sebastiani, A. Citterio, G. Leonardi, A.M. Valerio. [US10160652B2](#)

M. Galimberti, V. Barbera, R. Sebastiani, A. Truscello, A.M. Valerio. [EP3180379B1](#)

M. Galimberti, V. Barbera, [EP3538511A1](#)

M. Galimberti, V. Barbera, [EP3538481A1](#)

Production of few layers graphene



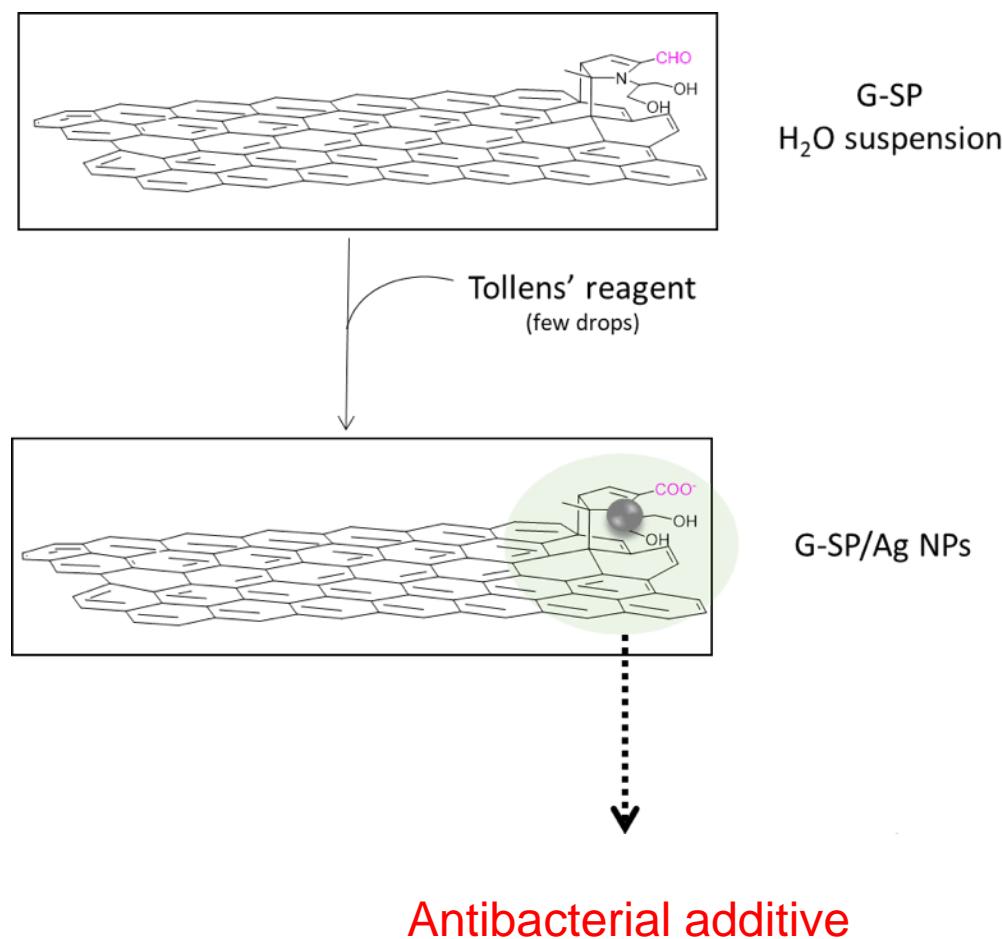
Electrically conductive coating layers

HSAG-K stearate

HSAG-SP

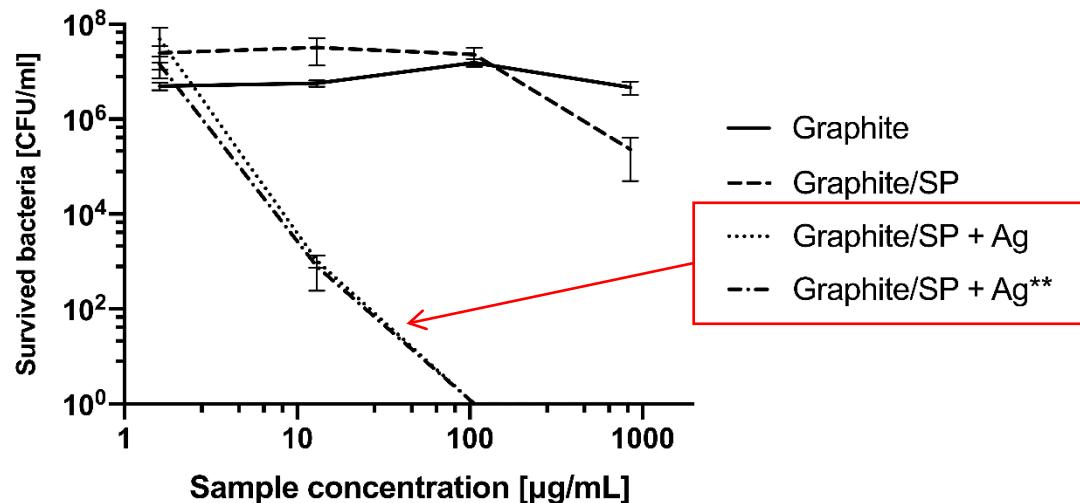
HSAG

Antibacterial additives



Antibacterial activity

Survived bacteria [CFU/ml] for Graphite samples



Test: ASTM Standard Guide E2315 – 16

Bacteria: *E. coli* JM109 – DSM3423

Material state: Dispersions of powders

Inoculum concentration: $\sim 10^6$ CFU/mL

Culture medium: LB / PBS 1:100 (v/v)

Culture conditions: 24h, 37°C, 5% CO₂, 90% R.H. Dynamic
conditions

**The sample has been functionalized
with an almost double quantity of Tollens' reagent.

Bionanocomposites



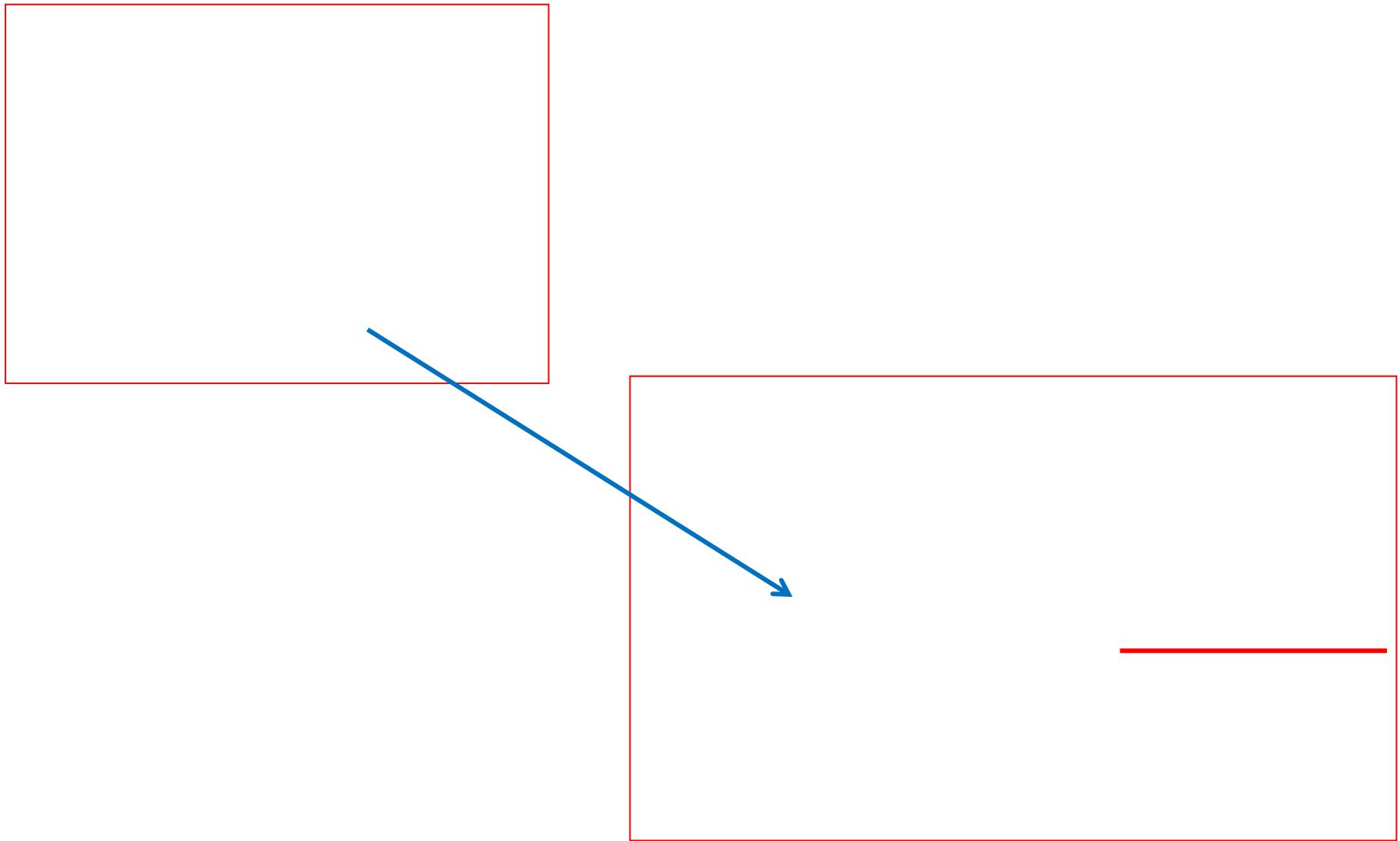
Carbon papers

Aerogels

V. Barbera, S. Guerra, L. Brambilla, M. Maggio, A. Serafini, L. Conzatti, A. Vitale, M. Galimberti, *Biomacromolecules*, 2017, 18 (12), 3978–399

S. Guerra, V. Barbera, A. Vitale, R. Bongiovanni, A. Serafini L. Conzatti, L. Brambilla, M. Galimberti, *Materials* 2020, 13, 39

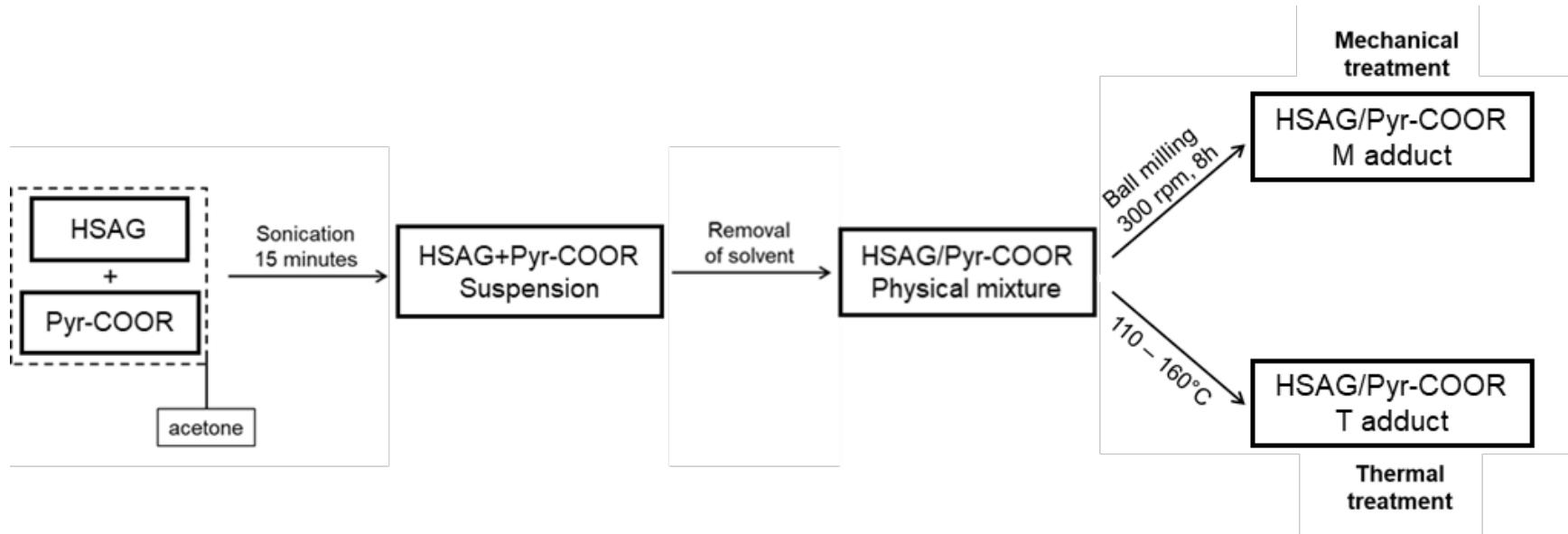
Bionanocomposites



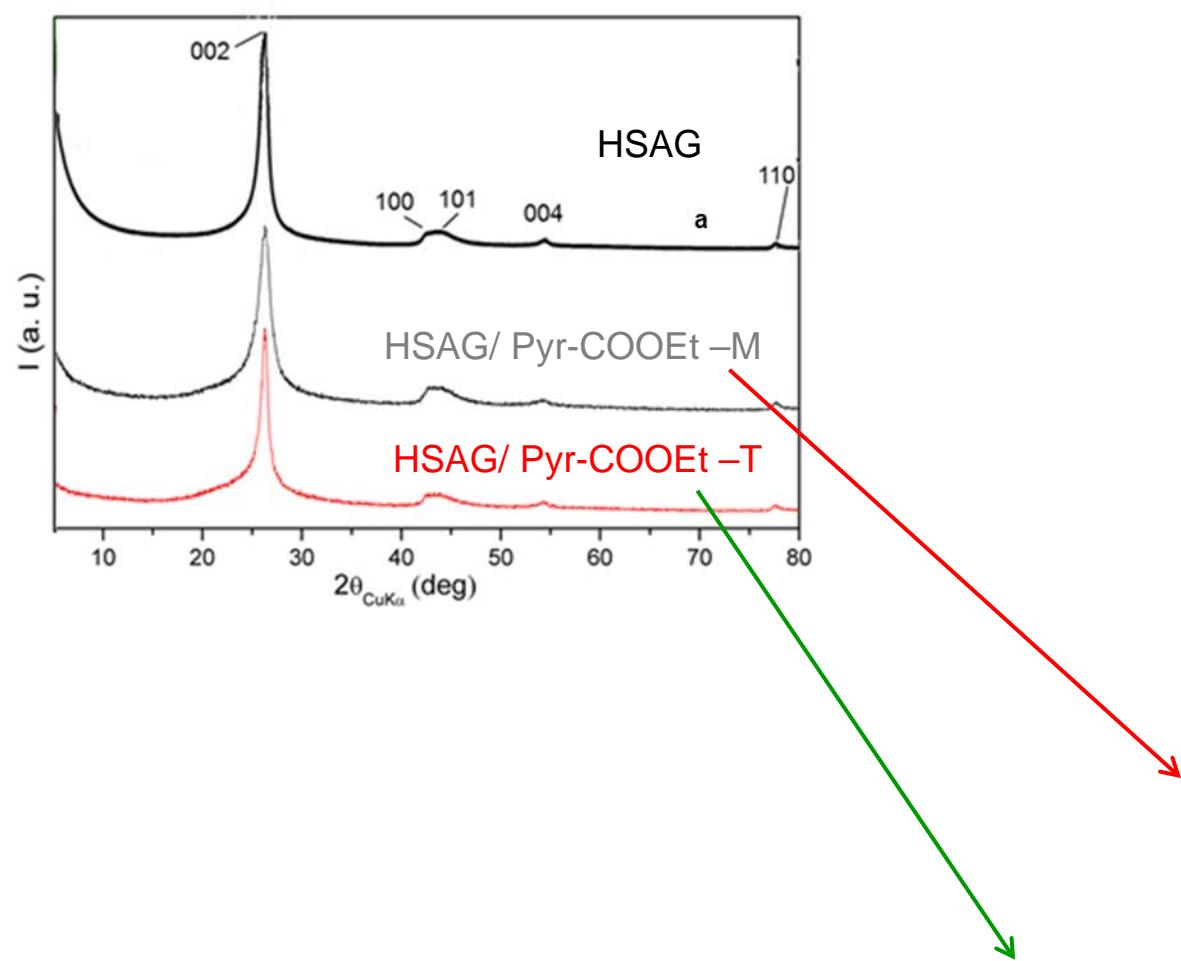
V. Barbera, S. Guerra, L. Brambilla, M. Maggio, A. Serafini, L. Conzatti, A. Vitale, M. Galimberti, *Biomacromolecules*, 2017, 18 (12), 3978–399

S. Guerra, V. Barbera, A. Vitale, R. Bongiovanni, A. Serafini L. Conzatti, L. Brambilla, M. Galimberti, *Materials* 2020, 13, 39

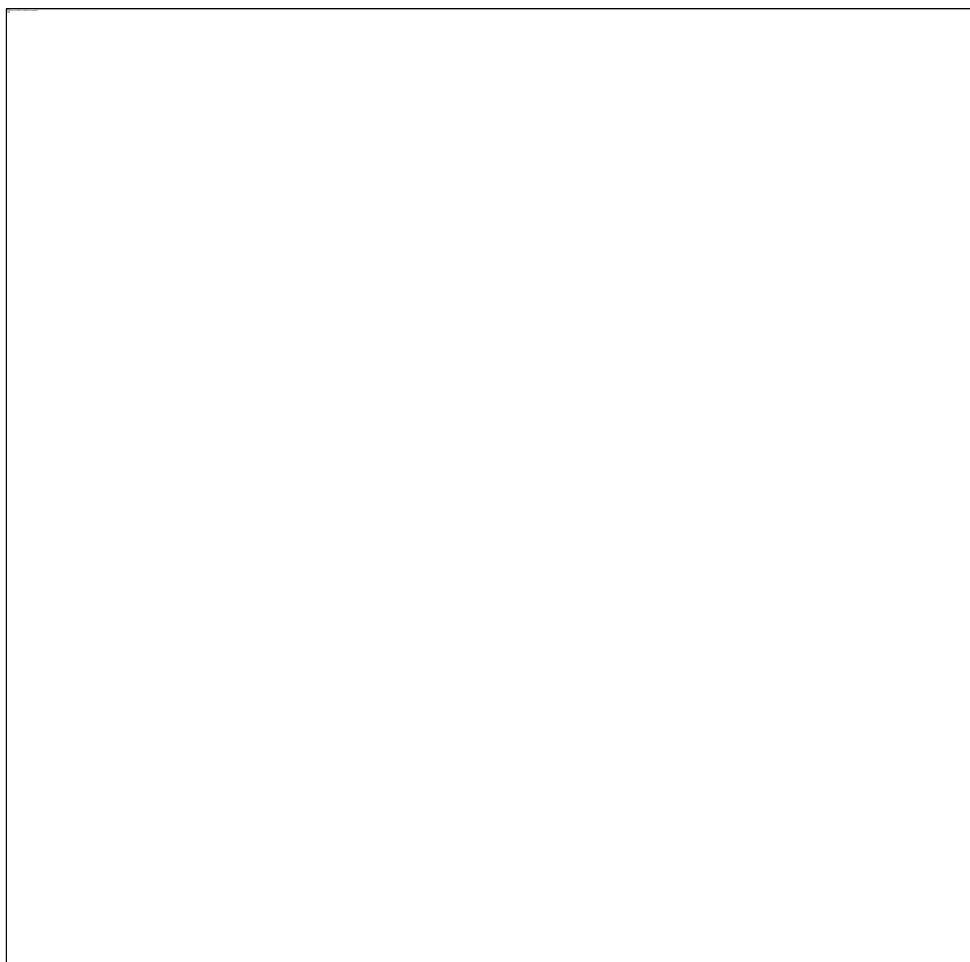
Functionalization of HSAG with a Pyrone derivative



Adduct of HSAG with Pyr-COOEt



EDGE-GO - Functionalization of HSAG with a Pyrone derivative



Preparation of few layers graphene



Conclusions



NanoCarbon Up Technology: 10 families of patents

Rubber
nanocomposites



Concrete with high
flexural strength



Carbon paper



Aerogels



No HSAG HSAG HSAG-SP

Polyurethanes



Polyol dispersions



Innovative Materials Group



instagram: @ismaterials.polimi

Politecnico di Milano, Department of Chemistry, Materials and Chemical Engineering "G. Natta"