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2	characterize Hexavalent chromium loss from soil to surface runoff
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Abstract

We investigate our ability to assess transfer of hexavalent chromium, Cr(VI), from 32 33 the soil to surface runoff by considering the effect of coupling diverse adsorption models with a two-layer solute transfer model. Our analyses are grounded on a set of 34 two experiments associated with soils characterized by diverse particle size 35 distributions. Our study is motivated by the observation that Cr(VI) is receiving much 36 attention for the assessment of environmental risks due to its high solubility, mobility, 37 and toxicological significance. Adsorption of Cr(VI) is considered to be at equilibrium 38 39 in the mixing layer under our experimental conditions. Four adsorption models, i.e., the Langmuir, Freundlich, Temkin, and the linear model, constitute our set of alternative 40 (competing) mathematical formulations. Experimental results reveal that the soil 41 42 samples characterized by the finest grain sizes is associated with the highest release of Cr(VI) to runoff. We compare the relative abilities of the four models to interpret 43 experimental results through Maximum Likelihood model calibration and four model 44 identification criteria (i.e., the information criteria AIC and AIC_C, and the Bayesian 45 criteria BIC and KIC). Our study results enable us to rank the tested models on the basis 46 of a set of posterior weights assigned to each of them. A classical variance-based global 47 sensitivity analysis is then performed to assess the relative importance of the uncertain 48 parameters associated with each of the models considered, within sub-regions of the 49 parameter space. In this context, the modeling strategy resulting from coupling the 50 51 Langmuir isotherm with a two-layer solute transfer model is then evaluated as the most skillful for the overall interpretation of both sets of experiments. Our results document 52

53	that (a) the depth of the mixing layer is the most influential factor for all models tested,
54	with the exception of the Freundlich isotherm, and (b) that the total sensitivity of the
55	adsorption parameters varies in time, with a trend to increase as time progresses for all
56	of the models. These results suggest that adsorption has a significant effect on the
57	uncertainty associated with the release of Cr(VI) from the soil to the surface runoff
58	component.

59 Keywords: Cr(VI) loss; surface runoff; model ranking; global sensitivity analysis

INTRODUCTION

According to the soil assessment results of China nationwide Multi-Purpose 61 Regional Geochemical Survey (MPRGS) project, considerable portions of lands used 62 for agriculture purpose in China show pollution signatures by metals. Chemical transfer 63 from the soil to surface runoff is a key process that contributes to non-point source 64 pollution. Metals (e.g., Cd, Pb and Cr) originating from a contaminated soil can then 65 contaminate surface water and groundwater, with a negative impact on human health 66 and various compartments of the ecosystem (Krishna and Govil, 2008). Among these 67 pollutants, Cr(VI) has received much attention because of its high solubility, mobility, 68 and toxicological significance in the environment. 69 Numerous studies focus on efficient methods (a) to reduce the amounts of Cr(VI) in 70 the environment, (b) to investigate the fate of Cr(VI) in soil and groundwater, and (c) 71 72 to monitor space-time distributions of Cr(VI) in surface runoff. He et al. (2004)

monitored metal (i.e., Cd, Co, Cr, Cu, Fe, Ni, Pb, Zn, Mn, and Mo) concentrations in
surface runoff at 11 sites in Florida and documented a positive correlation between
concentrations of the analyzed metals in runoff and soil. Ghosh et al. (2012) found that
Cr(VI) could be successfully adsorbed onto the fine-grained soil used as a liner material
in a landfill. Núñez-Delgado et al. (2015) found that both pine sawdust and oak wood
ash could be used (as biosorbents) to reduce the concentration of Cr(VI) released from
soil to water.

Migration of Cr(VI) through a soil system is typically described by advection, dispersion, adsorption and reaction processes. Advection is controlled by the mean flow rate of water, and dispersion is characterized by molecular diffusion and mechanical

83	dispersion. Langmuir (1918), Freundlich (1907), and linear isotherm models are often
84	used to simulate Cr(VI) adsorption experimental results under equilibrium assumption
85	(e.g., Ghosh et al., 2012; Núñez-Delgado et al., 2015; Sangiumsak and Punrattanasin,
86	2014; Mendonca et al., 2013; Fifi et al., 2013; Li et al., 2014). Gupta and Bahu (2009)
87	successfully simulated breakthrough curves of Cr(VI) by combining the Langmuir
88	model with a mathematical transport model. Chakraborty et al. (2015) embedded the
89	Langmuir and linear isotherms, respectively, in a one-dimensional advection-
90	dispersion-reaction-equation to estimate Cr(VI) transport parameters. Such isotherms
91	have also been integrated in well-known numerical codes (e.g., Hydrus-1D (Šimůnek
92	et al., 2009) and MT3DMS (Zheng et al., 1998)) to simulate reactive solute migration.
93	Transport of Cr(VI) is affected by the redox reaction of Cr, associated with Cr(III)
94	and/or Cr(VI) in the environment. Based on thermodynamics, Cr(III) oxidation should
95	be a spontaneous process. However, it needs to be catalyzed to take place in a natural
96	system, due to its very slow kinetics (Apte et al., 2005). In this context, one can note
97	that while chromium is highly mobile with flow in the hexavalent form, its migration
98	would be strongly retarded in the trivalent form because of the typically strong
99	adsorption capacity of soil to Cr(III). It is therefore worth noting that detectable
100	amounts of chromium found in natural waters are usually in the hazardous hexavalent
101	form, manganese oxides essentially being the materials that can oxidize Cr(III) to Cr(VI)
102	in a natural system (Fendorf and Zasoski, 1992). Organic materials, such as sulfides,
103	and ferrous iron, can reduce Cr(VI) to Cr(III) (Fendorf, 1995). We refer to Fendorf
104	(1995) for a comprehensive review on this aspect.

Transfer of a chemical from soil to surface runoff is a complex process. Most of the 105 experimental studies on this process are conducted at the laboratory scale (sand boxes). 106 107 Various chemical transport models have been developed to explain experimental results. Modeling methods based on the diffusion (Wallach and van Genuchten, 1990) and the 108 mixing-layer theory (Donigian et al., 1977) were the two approaches generally used to 109 simulate the process. The diffusion theory assumes that chemical exchange between 110 runoff and soil is controlled by an accelerated diffusion process. The theory has been 111 used in a variety of studies (Ahuja et al., 1981, 1983; Sharpley et al., 1980), even though 112 113 the function to describe the accelerated diffusion process is somewhat arbitrarily chosen (Gao et al., 2004). Note that using an accelerated diffusion coefficient in the simulations 114 may aptly capture a set of experimental data, while providing no explanation to the 115 116 physical mechanism of the diffusion process. Because the physical mechanism in the accelerated diffusion theory remains unclear, the diffusion theory has limited ability for 117 prediction. 118

The mixing-layer theory was first proposed by Donigian et al. (1977). It is based on the assumption that surface runoff water mixes entirely and instantaneously with soil water in a thin layer on the soil surface with no contribution from the soil below such a mixing layer. However, Zhang et al. (1997) found that the diffusion flux from the soil underlying the mixing layer could not be neglected under poor drainage conditions. Therefore, Zhang et al. (1999) used the convection-diffusion equation to simulate the diffusion flux from the underlying soil.

126 A series of mechanistic models have been developed in recent years. Soil erosion

was characterized by the Rose model (Rose et al., 1994; Hairsine and Rose, 1991), and 127 solute transfer between the runoff and the exchange layers was determined by the water 128 129 transfer rate. Gao et al. (2004) assumed that the solute transfer from soil to surface runoff was affected by both drop-liquid and drop-liquid-solid interactions, and the 130 process was conceptualized as the model of the three layers/compartments, i.e., a runoff, 131 an exchange, and the underlying soil compartment. Tong et al. (2010) proposed a new 132 model to integrate the runoff and the mixing layers into a unique mixing region. In the 133 model, solute concentration in the runoff was calculated as αC_w , where α ($0 < \alpha \le 1$) 134 is an incomplete mixing coefficient and C_{w} is the aqueous-phase solute concentration 135 in the mixing layer $[M L^{-3}]$. The net chemical flux from the mixing layer to the 136 underlying soil is calculated as $i\gamma C_w$, and i and γ ($0 < \gamma \le 1$) are, respectively, the 137 infiltration rate [L T⁻¹] and another incomplete mixing coefficient. All of these models 138 assume that the soil surface is (nearly) horizontal, the thickness of the mixing layer is 139 regarded as stable or constant, and lateral and return flows in the soil are negligible. 140 141 Dong and Wang (2013) considered an inclined soil surface and relied on the conceptual model of Gao et al. (2004). They assumed solute concentrations in the runoff and the 142 mixing layer were to be identically affected by raindrop splash. In addition, the model 143 allows the mixing layer to have a variable depth, and takes into account the effects of 144 raindrop splash, lateral flow, and return flow. Notably, solute concentrations in the 145 mixing and/or exchange layers are considered as uniform. 146

The main considered factors affecting Cr(VI) transfer from soil to surface runoff in
the above referenced models include rain intensity (Gao et al., 2004), ponding water

depth (Gao et al., 2004), soil characteristics (Tong et al., 2010), soil slope (Dong and
Wang 2013), and drainage conditions (Walker et al., 2007; Zhang et al., 1999).
Adsorption of a chemically active solute is another important factor that affects solute
loss. Gao et al. (2004) integrated the linear adsorption model with a solute loss model
to evaluate phosphorus concentrations in runoff. In this broad context, it is still lack of
a detailed study on the influence of the adsorption model choice on Cr(VI) loss
simulation from soil to surface runoff.

An objective of our study is to evaluate the impact of a model selected from a set of 156 157 differing adsorption models (i.e., the Langmuir, Freundlich, Temkin and linear models) on the Cr(VI) loss simulation. We do so by relying on a set of two experiments and 158 grounding our study on rigorous model identification criteria (Neuman, 2003; Ye et al., 159 160 2004, 2008; Neuman et al., 2012; Bianchi-Janetti et al., 2012). These enable us to (a) compare the ability of each of the tested models to interpret the target experimental 161 results and (b) rank the models through the evaluation of probabilistic weights assigned 162 163 to each of them. We then provide model-averaged estimates (and associated uncertainty bounds) of Cr(VI) concentrations in runoff by leveraging on the diverse interpretive 164 skills of all models analyzed. Since each of the models is associated with a set of 165 typically unknown/uncertain parameters, we also perform a classical variance-based 166 global sensitivity analysis to assess the relative contribution of the uncertain parameters 167 associated with each model to the variability of Cr(VI) released from the soil to surface 168 runoff. 169

170 The rest of the study is structured as follows. We first provide descriptions for the

171	experiments performed and for the modeling approaches. We then present and discuss
172	our results, in terms of model ranking and multi-model analysis as well as global
173	sensitivity analysis. We finally make our conclusions.
174	
175	MATERIALS AND METHODS
176	Soil Chemical characteristics
177	Representative silt soil samples obtained from the region surrounding the city of
178	Wuhan, China, were dried, ground and passed through 2 mm and 1 mm sieves, to form
179	the two subsamples, respectively termed soil sample 1 and 2, employed in our
180	experimental investigations. Table 1 lists the main chemical characteristics of soil 2.
181	The latter is seen to be characterized by a very low initial Cr(VI) concentration, the
182	overall picture suggesting that the soil is oxic. As such, we do not consider Cr(VI)
183	reduction in our experimental investigations.
184	Experiments
185	We leverage on the experimental set-up used by Tong et al. (2010), to which we refer
186	for additional details. In summary, a steel sandbox (with length, width and depth
187	respectively of 100, 30, and 40 cm) with rustproof paint and equipped with two drainage
188	holes at the bottom was used (see Fig. 1). A 5-cm-thick layer of gravel is packed at the
189	bottom of the sand box to allow for water drainage. The elevations of the drainage
190	outlets can be modified to achieve diverse drainage conditions. The gravel layer is
191	covered with a nylon screen to prevent loss of soil particles, the soil subject to the
192	experiments being packed above the screen. A rectangular hatch that opens into a V-
193	shaped trench and located 30 cm above the bottom of the box is used to collect surface 9

runoff water. The height difference between the soil surface and the runoff hatch
corresponds to the depth of ponded water. A rainfall simulator (formed by 8 hypodermic
needles) is placed at an elevation of 120 cm above the soil layer.

Both soil samples were purposely set at the same Cr(VI) concentration of 300 mg kg⁻ 197 ¹ in both the liquid and soild phases, taking into account differences of their initial 198 volumetric water contents (respectively equal to 0.28 and 0.30 for soil 1 and 2) and the 199 preset saturated volumetric water contents (0.49 for both soils), the initial Cr(VI) 200 concentrations (equal to 0 mg kg^{-1} for both soils), and the soil bulk density (1.35 g cm⁻¹ 201 ³). The soils were then packed gradually to a depth of 18.5 cm. The depth of the runoff 202 layer was 1 cm for both experiments. The soil surfaces were covered with plastic films 203 and allowed to incubate for 12 hours. During this time, the system was kept at a constant 204 205 room temperature of 25°C. Thus, evaporation from the soil surfaces were considered as negligible. Before the beginning of the rainfall simulation, the outlets of the drainage 206 holes were set to a height of 0 cm for both experiments, i.e., both experiments were 207 conducted under free drainage conditions. 208

The simulated rainfall was set to an intensity of 0.092 cm min⁻¹ and 0.100 cm min⁻¹, respectively for experiments 1 (soil 1) and 2 (soil 2). We denote t_p as the time at which water began ponding on the surface of the soil, t_r and t_s respectively as the times at which runoff first occurred and attained stationarity. Table 2 lists the key parameters characterizing the experimental conditions. Collection of runoff samples was initially performed at 4-10 minutes intervals, progressively longer sampling intervals being used as time elapsed. Dissolved Cr(VI) in the runoff samples were measured by the atomic flame method using an atomic flame spectrophotometer. Minute quantities of sediments

eventually found in the collected water samples were neglected.

218 *Two-layer mathematical model*

We rely on the two-layer model for its computational efficiency and because it is grounded on assumptions that are consistent with our experimental conditions. We briefly describe the main characteristics of the two-layer model used in this study, additional details being found in Tong et al. (2010). As shown in Fig. 2, the conceptual model comprised two vertical layers, i.e., the entire mixing zone, that includes the soil mixing layer and the runoff layer, and the underlying soil.

225 Without considering adsorption, the dissolved chemical mass per unit area, M_w [M 226 L⁻²], is given by:

227
$$M_{w} = C_{w} \left[\alpha \left(h_{w} - h_{mix} \theta_{s} \right) + h_{mix} \theta_{s} \right]$$
(1)

Here, C_w is the aqueous-phase solute concentration in the mixing layer [M L⁻³]; α (0< $\alpha \le 1$) is an incomplete mixing coefficient [-]; h_w is the net water depth across the entire mixing zone [L], h_{mix} is the mixing layer thickness [L], and θ_s is saturated volumetric water content in the soil system [L³ L⁻³].

In the presence of adsorption, Eq. (1) becomes:

233
$$M_{w} = \alpha \left(h_{w} - h_{mix} \theta_{s} \right) C_{w} + h_{mix} \left[\theta_{s} C_{w} + \rho_{b} S \right]$$
(2)

where *S* [M M⁻¹] and ρ_b [M L⁻³] respectively are the solute concentration adsorbed onto the soil and the soil bulk density.

The dynamic behavior of the system can then be characterized as:

237
$$\frac{d\left[M_{w}(t)\right]}{dt} = -\gamma i C_{w}(t) - \alpha q C_{w}(t)$$
(3)

where γ (0< $\gamma \le 1$) is an incomplete mixing coefficient [-] (different from α); *q* [L T⁻¹] and *t* [T] respectively denoting the specific discharge rate of the overland flow and time.

241 Adsorption isotherm models

We consider three nonlinear models (i.e., the Langmuir, Freundlich, and Temkin equations) and a linear model to interpret solute adsorption. Given the experimental setting, adsorption is assumed to take place under isothermal conditions.

The Langmuir isothermal nonlinear equation (Langmuir et al., 1918) is:

246
$$S = S_{\max} \frac{K_{eq} C_w}{1 + K_{eq} C_w}$$
(4)

where S_{max} [M M⁻¹] and K_{eq} [L³ M⁻¹] are (typically unknown and uncertain) model parameters, respectively representing the theoretical maximum adsorption capacity of the soil and the equilibrium adsorption coefficient.

Freundlich et al. (1906) proposed the following empirical nonlinear equation:

$$S = K_f C_w^{\frac{1}{n}}$$
(5)

model parameters being the adsorption coefficient K_f [L^{3/n} M^{-1/n}] and the exponent n

[-]. When
$$n = 1$$
, Eq. (5) reduces to the linear model:
 $S = K_d C_w$ (6)
where K_d [L³ M⁻¹] is the adsorption coefficient.
The Temkin isotherm model was first proposed by Temkin and Pyzhev (1940) and

The Temkin isotherm model was first proposed by Temkin and Pyzhev (1940) and can be expressed in the form:

258
$$S = \frac{RT_e}{f} \ln \left(KC_w \right)$$
(7)

where *R* [J mol⁻¹ K⁻¹] is the universal gas constant, T_e [K] is absolute temperature, *f* [J mol⁻¹] is the Temkin isotherm constant, and *K* [L³ M⁻¹] is the Temkin isotherm equilibrium binding constant.

262 Numerical modeling

Embedding each of the adsorption isotherm models (4)-(7) into equations (2)-(3) 263 leads to multiple mathematical models whose relative skill to interpret the observed 264 system behavior is evaluated here. Hereinafter, we denote (a) Langmuir, (b) Freundlich, 265 (c) linear adsorption, or (d) Temkin coupled models as the formulations respectively 266 arising by coupling (2)-(3) with (4), (5), (6), or (7). The numerical solution of these 267 solute transport models relies on direct observation of water flow characteristics from 268 the experimental campaign. The observed flow dynamics are then used as input to each 269 transport model and are characterized by segmenting the overall flow process onto the 270 five sequential temporal windows described in the following, each corresponding to a 271 well-defined hydrological manifestation. 272

273 <u>*Period* 0</u>. It comprises observation times ranging from the beginning of the simulated 274 rainfall to the saturation of the mixing layer. During this period, i.e., before the start of 275 water ponding, the water infiltration rate i_{up} [L T⁻¹] coincides with rainfall intensity, 276 p, i.e., $i_{up} = p$ (corresponding to i = 0, and q = 0). Infiltration from the mixing layer 277 to the underlying soil is assumed to be negligible because the mixing layer is very thin 278 and the time t_{sa} required to attain saturation of the mixing layer can be approximated 279 as:

280
$$t_{sa} = \frac{h_{mix} \left(\theta_s - \theta_0\right)}{i_{up}}$$
(8)

where θ_0 is the initial volumetric water content in the soil system [M³ M⁻³].

282 <u>*Period* 1</u>. During this period, ranging from saturation of the mixing layer 283 (corresponding to time t_{sa} given by Eq. (8)) to the onset of water ponding, water has 284 not yet ponded on the soil surface, and the infiltration rate of soil equals the rainfall 285 intensity, i.e.:

286
$$h_w = h_{mix}\theta_s$$
; $i_{up} = i = p$; $q = 0$ (9)

Hereinafter, we illustrate our derivations using the Langmuir isotherm as a test bed, the corresponding derivations associated with the other isotherm models being directly inferable from this. Considering Eqs. (2) and (4) leads to:

290
$$M_{w} = \alpha \left(h_{w} - h_{mix} \theta_{s} \right) C_{w} + h_{mix} \left[\theta_{s} C_{w} + \rho_{s} \frac{C_{w}}{BC_{w} + C} \right]$$
(10)

where $B = 1/S_{\text{max}}$, and $C = 1/(K_{eq}S_{\text{max}})$. The initial concentration of the solute in the mixing layer, $C_w(t_{sa})$, is obtained from Eq. (10). Replacing Eq. (10) into Eq. (3) yields:

$$\frac{\left(h_{mix}\theta_{s} + \frac{h_{mix}\rho_{s}}{C}\right)\ln\frac{C_{w}(t)}{C_{w}(t_{sa})} - \frac{h_{mix}\rho_{s}}{C}\ln\frac{BC_{w}(t) + C}{BC_{w}(t_{sa}) + C} - \frac{h_{mix}\rho_{s}}{BC_{w}(t) + C} + \frac{h_{mix}\rho_{s}}{BC_{w}(t_{sa}) + C} = \gamma p(t_{sa} - t)$$

$$(11)$$

We solve Eq. (11) via a fourth-order Runge-Kutta method with a uniform time step of0.01 seconds.

296 <u>Period 2</u>. During this window, spanning from the onset of water ponding to the 297 beginning of runoff, respectively observed at experimental times t_p and t_r , the 298 infiltration rate is assumed to decrease linearly in time according to 299 $i(t) = p - a_0(t - t_p)$, the value of $a_0 > 0$ characterizing the rate of increase of the 300 ponding depth and being evaluated via $i(t_r) = p - a_0(t_r - t_p)$. As no surface runoff is 301 observed, q = 0 during this period and the rate of ponding depth increase equals to 302 p - i(t). The temporal variation of the net water depth of the mixing zone can then be 303 obtained as:

304
$$h_w = h_p + h_{mix}\theta_s = \frac{1}{2}a_0(t - t_p)^2 + h_{mix}\theta_s$$
 (12)

where we recall that h_p is the depth of the ponding layer. Concentration $C_w(t_p)$ of the solute in the mixing layer is calculated through Eq. (11) evaluated at time t_p . Combining Eqs. (12), (10), and (3), leads to the following equation describing $C_w(t)$ across this time period:

$$309 \quad \frac{\frac{1}{2}a_{0}\alpha(t-t_{p})^{2}\frac{dC_{w}(t)}{dt} + a_{0}\alpha C_{w}(t)(t-t_{p}) + h_{mix}\theta_{s}\frac{dC_{w}(t)}{dt} + \frac{h_{mix}\rho_{s}C}{\left[BC_{w}(t)+C\right]^{2}}\frac{dC_{w}(t)}{dt} = \gamma \left[a_{0}(t-t_{p})-p\right]C_{w}(t)$$
(13)

Similar to Eq. (11), we solve Eq. (13) via a fourth-order Runge-Kutta method with auniform time step of 0.01 seconds.

312 <u>*Period* 3</u>. During this interval, ranging from the onset of runoff to its stabilization (at 313 time t_s), the temporal decrease of infiltration is approximated via:

314
$$i(t) = i(t_r) - b(t - t_r);$$
 $q = p - i$ (14)

where *b* is a parameter characterizing the infiltration decrease rate and is evaluated via $i(t_s) = i(t_r) - b(t_s - t_r)$. The initial solute concentration in the runoff fluid, $\alpha C_w(t_r)$, can be obtained from the solution of Eq. (13), evaluated at time $t = t_r$. Substituting Eqs. (14) and (10) into Eq. (3) yields:

$$\begin{bmatrix} \alpha h_p + h_{mix} \theta_s \end{bmatrix} \ln \frac{C_w(t)}{C_w(t_r)} + \frac{h_{mix} \rho_s}{C} \ln \frac{C_w(t)}{C_w(t_r)} - \frac{h_{mix} \rho_s}{C} \ln \frac{BC_w(t) + C}{BC_w(t_r) + C} - \frac{h_{mix} \rho_s}{BC_w(t) + C} + \frac{h_{mix} \rho_s}{BC_w(t_r) + C} = \left\{ (\alpha - \gamma) \left[i(t_r) + bxt_r \right] - \alpha p \right\} t - t_r - \frac{b}{2} (\alpha - \gamma) \left(t^2 - t_r^2 \right) \right\}$$
(15)

322 <u>*Period* 4</u>. The infiltration rate is stationary during this last period and can be obtained 323 by evaluating Eq. (14) at time $t = t_s$. The initial concentration of the runoff fluid, 324 $\alpha C_w(t_s)$, is obtained by evaluating Eq. (15) at time $t = t_s$. The resulting format of Eq. 325 (3), i.e.:

$$326 \qquad \frac{\left[\alpha h_{p}+h_{mix}\theta_{s}\right]\ln\frac{C_{w}(t)}{C_{w}(t_{r})}+\frac{h_{mix}\rho_{s}}{C}\ln\frac{C_{w}(t)}{C_{w}(t_{r})}-\frac{h_{mix}\rho_{s}}{C}\ln\frac{BC_{w}(t)+C}{BC_{w}(t_{r})+C}-\frac{h_{mix}\rho_{s}}{BC_{w}(t)+C}+\frac{h_{mix}\rho_{s}}{BC_{w}(t_{r})+C}=\left[i(t_{s})(\alpha-\gamma)-\alpha p\right](t-t_{s})$$
(16)

is then solved by a fourth order Runge-Kutta method, as described above.

For brevity, the solute transport models obtained by considering the Langmuir, Freundlich, Temkin and linear adsorption models are respectively denoted as L, F, T and H models.

331 Maximum Likelihood Model calibration

We consider the vector $\mathbf{Y}^* = [Y_{(1)}^*, Y_{(2)}^*, \dots, Y_{(N_C)}^*]$ collecting N_C measurements of dissolved Cr(VI) concentration observed in surface the runoff at sampling time i = 1, 2, ..., N_C . The general relationship expressing Cr(VI) concentration in runoff through a mathematical model f(.) associated with the vector $\mathbf{X} = [x_1, x_2, \dots, x_{N_p}]$ of N_p unknown parameters is here represented as $\mathbf{Y} = f(\mathbf{X})$, vector $\mathbf{Y} = [Y_{(1)}, Y_{(2)}, \dots, Y_{(N_C)}]$ including N_C simulated Cr(VI) concentration values in the runoff at time i = 1, 2, ..., 338 N_c . Experimental measurement errors are considered to be independent (e.g., Carrera 339 and Neuman, 1986; Bianchi-Janetti et al., 2012), the corresponding error covariance 340 matrix, \mathbf{B}_c , being diagonal, σ_i^2 ($i = 1, 2, ..., N_c$) representing observation error 341 variance. The Maximum Likelihood (ML) estimate $\hat{\mathbf{X}}$ of the vector of the N_p 342 uncertain model parameters can be obtained by minimizing with respect to \mathbf{X} the 343 negative log likelihood criterion (e.g., Carrera and Neuman 1986):

344 NLL =
$$\sum_{i=1}^{N_c} \frac{Y_{(i)}^* - \hat{Y}_{(i)}}{\sigma_i^2} + \ln |\mathbf{B}_c| + N_c \ln(2\pi)$$
 (17)

where $\hat{Y}_{(i)}$ is the output provided by a given interpretive model at the *i*th observation time. We note that minimizing Eq. (17) corresponds to minimization of the least square criterion (Carrera and Neuman, 1986; Bianchi-Janetti et al., 2012 and references therein):

349
$$J = \sum_{i=1}^{N_c} \frac{Y_{(i)}^* - \hat{Y}_{(i)}}{\sigma_i^2}$$
(18)

Here, minimization of (18) is obtained upon relying on the iterative Levenberg-Marquardt algorithm as embedded in the well documented computational framework PEST (Doherty, 2002). We consider minimizing Eq. (18) with uniform (and generally unknown) measurement error variance, i.e., $\sigma_i^2 = \sigma^2$ for $i = 1, 2, ..., N_c$ (see, e.g., Bianchi-Janetti et al., 2012).

355 Model Identification Criteria

To evaluate the performance of the four alternative models considered (i.e., L, F, T and H), we rely on the four criteria:

$$358 \qquad \text{AIC} = \text{NLL} + 2N_p \tag{19}$$

359 AIC_c = NLL + 2N_p +
$$\frac{2N_p(N_p+1)}{N_c - N_p - 2}$$
 (20)

$$360 \quad BIC = NLL + N_P \ln(N_C) \tag{21}$$

361
$$\operatorname{KIC} = \operatorname{NLL} + N_{P} \ln\left(\frac{N_{C}}{2\pi}\right) - \ln\left|\mathbf{Q}\right|$$
 (22)

where Q represents the Cramer-Rao lower-bound approximation for the covariance 362 matrix of the parameter estimates, i.e., the inverse expected Fisher information matrix, 363 which renders a quantitative appraisal of the quality of parameter estimates and of the 364 information content carried by data about model parameters (see, e.g., Ye et al., 2008 365 366 for details). Here, Eq. (19) is proposed by Akaike (1974), Eq. (20) by Hurvich and Tsai (1989), Eq. (21) by Schwartz (1978) and Eq. (22) by Kashyap (1982). It is noted that 367 the lowest value of a given model identification criterion indicates the most favored 368 369 model (according to the criterion itself) at the expense of the other ones.

Maximum Likelihood Bayesian Modeling averaging 370

The discrimination criteria (19)-(22) can also be considered to assign posterior 371 372 probability weights quantifying uncertainty associated with each of the tested isotherm models. The posterior probability linked to model M_k ($k = 1, 2, ..., N_M, N_M$, which is 373 equal to 4 in our study, being the number of interpreting models assessed) is evaluated 374 as (Ye et al., 2008): 375

376
$$P(M_{k} | \mathbf{Y}^{*}) = \frac{\exp\left(-\frac{1}{2}(IC_{k} - IC_{\min})\right)P(M_{k})}{\sum_{k}^{N_{M}} \exp\left(-\frac{1}{2}(IC_{k} - IC_{\min})\right)P(M_{k})}$$
(23)

where IC_k is either AIC (19), AIC_c (20), BIC (21), or KIC (22), $IC_{min} = \min\{IC_k\}$ 377 being its minimum value calculated across the set of the four models examined; and 378 18

379 $P(M_k)$ is the prior probability associated with the k^{th} alternative model. Because no 380 prior information is available, we set $P(M_k) = 1/N_M$.

Averaging across the moments provided by all alternative models renders the following (Bayesian-averaged) expressions for the leading moments (Draper, 1995; Hoeting et al., 1999):

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$$E\left(\mathbf{Y} \mid \mathbf{Y}^*\right) = \sum_{k=1}^{N_M} E\left(\mathbf{Y} \mid \mathbf{Y}^*, M_k\right) P\left(M_k \mid \mathbf{Y}^*\right)$$
(24)

$$V\left(\mathbf{Y} \mid \mathbf{Y}^{*}\right) = \sum_{k=1}^{N_{M}} Var\left(\mathbf{Y} \mid \mathbf{Y}^{*}, \boldsymbol{M}_{k}\right) P\left(\boldsymbol{M}_{k} \mid \mathbf{Y}^{*}\right) + \sum_{k=1}^{N_{M}} \left(E\left(\mathbf{Y} \mid \mathbf{Y}^{*}, \boldsymbol{M}_{k}\right) - E\left(\mathbf{Y} \mid \mathbf{Y}^{*}\right)\right)^{2} P\left(\boldsymbol{M}_{k} \mid \mathbf{Y}^{*}\right)$$
(25)

Here, $E(\mathbf{Y} | \mathbf{Y}^*)$ and $V(\mathbf{Y} | \mathbf{Y}^*)$ respectively are model-averaged estimate and variance of \mathbf{Y} conditional on the set of Cr(VI) observations collected in \mathbf{Y}^* ; and $E(\mathbf{Y} | \mathbf{Y}^*, M_k)$ and $V(\mathbf{Y} | \mathbf{Y}^*, M_k)$ respectively are the mean and variance of \mathbf{Y} conditional on \mathbf{Y}^* and model M_k .

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RESULTS AND DISCUSSION

Here, we start by illustrating the available Cr(VI) observations and discuss the results of ML-based calibration of the four models analyzed. We then quantify posterior model weights according to the selection criteria considered and use these to (*a*) rank the models in terms of their relative skill to interpret the available data and (*b*) compute model-averaged estimates and corresponding uncertainty bounds. We resort to a classical variance-based global sensitivity analysis (GSA) to quantify the relative contribution of the uncertain parameters characterizing each of the models tested to the

variability of Cr(VI) concentration in the runoff water. We then discuss the implication
of adsorption function by relying on the most skillful amongst the models tested to
characterize Cr(VI) loss from soil to surface runoff.

402 *Model ranking*

Observed Cr(VI) concentrations in the runoff for experiments 1 and 2 are respectively depicted in Fig. 3a and 3b. These figures depict the corresponding concentration values obtained after ML model calibration for the four models analyzed. Table 3 lists ML parameter estimates together with the associated standard deviation (SD).

Simulated concentrations are generally in good agreement with experimental 408 evidences, a remarkable exception being model T in experiment 1. The high values of 409 the estimated standard deviations listed in Table 3 can be partially due to linearity 410 411 assumptions upon which the employed parameter optimization strategy is implemented and/or a trade-off between the information content associated with data and the number 412 and nature of model parameters considered (Doherty 2002). The latter observation is 413 consistent with the observed high values of the entries of the cross-correlation matrix 414 associated with parameter estimates. These are listed in Table 4 and suggest that the 415 available data are not conducive to unique estimates of model parameters. 416

Model calibration results indicate that the soil used in experiment 2 has a higher adsorption capacity than the soil in experiment 1. This result is consistent with the experimental setting, which comprises a finer soil texture in experiment 2. One can also note that experiment 2 is associated by a higher depth of the mixing layer than experiment 1. This finding is consistent with our experiment setting, according to which

soil 1 can form a much stronger shield (Heilig et al., 2001) against rain drop erosion on 422 surface soil 1 than soil 2, because of its sedimentological composition. This partially 423 424 supports the higher Cr(VI) concentrations obtained for experiment 2, which are consistent with the observation that the depth of the mixing layer directly influences 425 the total mass of solute that can be transferred from soil to surface runoff. One should 426 also notice that a higher infiltration rate occurs in experiment 1 than in experiment 2, 427 thus indirectly suggesting that less Cr(VI) mass is lost in the mixing layer through 428 surface runoff in the former set-up than in the latter. 429

430 Calculated values for each of the model identification criteria considered are listed in Table 5. We can see that AIC, AIC_C and BIC values are close, because of their similar 431 structures. For experiment 1, the lowest AIC, AIC_c, BIC and KIC values are -140.79, -432 433 137.46, -134.90, and -156.22, respectively, their corresponding counterparts for experiment 2 being -36.04, -30.79, -29.23, and -61.81, respectively. Posterior model 434 weights of the alternative models analyzed are listed in Table 6. For experiment 1, the 435 436 highest AIC-, AIC_C- and BIC-based posterior model weights are 47.09%, 66.54% and 61.60%, respectively, all of them being associated with the H model. Otherwise, the 437 highest KIC-based posterior model weight (i.e., 53.97%) is linked to the L model. For 438 experiment 2, the highest AIC-, AIC_C- and BIC-based posterior model weights are 439 90.39%, 90.30% and 90.34%, respectively, all of them being connected to the L model. 440 The highest KIC-based posterior model weight is 77.87% and is linked to the T model. 441 442 One can then conclude that the H and/or L model are identified as the most skillful ones for experiment 1, respectively according to the BIC- and/or KIC-based posterior model 443

444 weights, the L and/or T models being favored to the interpretation of experiment 2.

According to Ye et al. (2008), AIC and AIC_C are based on the premise that the true 445 model is comprised in the set of alternative models tested, a constraint which is not 446 shared by BIC or KIC. Additionally, it can be noted that KIC imbues a balancing effect 447 between expected value of information and model parsimony. At the same time, it has 448 been observed that KIC tends to favor models which can lead to potentially 449 controversial results, due to the effect of **Q** (e.g., Tsai and Li, 2008; Li and Tsai, 2009). 450 On the basis of this discussion, and for the purpose of our application, we base our 451 452 model selections on KIC. By further noting that the T model displays an unsatisfactory pattern in the interpretation of experiment 1 (see Fig. 3a), we conclude that the L model 453 can be considered as the most skillful amongst the alternatives considered for the 454 455 purpose of the overall interpretation of both experimental datasets.

We now illustrate the results of the MLBMA multi-model analysis. The latter is 456 performed through a numerical Monte Carlo (MC) framework structured according to 457 the following steps: (a) N = 100,000 MC samples of each uncertain model parameter 458 x_i ($i = 1, 2, \dots, N_p$) are randomly generated for each candidate model M_k 459 $(k = 1, 2, \dots, N_M)$ through the Latin Hypercube sampling technique considering each 460 x_i as independent and identically distributed (i.i.d.) random variables, uniformly 461 distributed within the support $[\hat{x}_i - 30\% \hat{x}_i, \hat{x}_i + 30\% \hat{x}]$, with mean value equal to \hat{x}_i 462 (i.e., the ML-based parameter estimate); (b) calculating N MC realizations of dissolved 463 Cr(VI) concentrations in the surface runoff (collected in vector **Y**) via each candidate 464 model M_k ; (c) computing (ensemble) mean $E(\mathbf{Y} | \mathbf{Y}^*, M_k)$ and variance 465

466 $V(\mathbf{Y} | \mathbf{Y}^*, M_k)$; (*d*) and evaluating model-averaged estimates $E(\mathbf{Y} | \mathbf{Y}^*)$ and 467 $Var(\mathbf{Y} | \mathbf{Y}^*)$ through (25) and (26).

Fig. 4 depicts scatterplots of model-averaged estimates $E(\mathbf{Y} | \mathbf{Y}^*)$ and 468 corresponding uncertainty bounds (of width equal to $\pm \sqrt{V(\mathbf{Y} | \mathbf{Y}^*)}$) versus Y 469 measurements for experiment 1 on the basis of IC = AIC (Fig. 4a), AIC_C (Fig. 4b), BIC 470 (Fig. 4c), and KIC (Fig. 4d). For completeness, each subplot also depicts 471 $E(\mathbf{Y} | \mathbf{Y}^*, \tilde{M}_k)$ and $E(\mathbf{Y} | \mathbf{Y}^*, \tilde{M}_k) \pm \sqrt{V(\mathbf{Y} | \mathbf{Y}^*, \tilde{M}_k)}$, as computed by the most 472 skillful model, \tilde{M}_k , identified by the corresponding IC. Maximum Likelihood 473 estimates of Y obtained through model \tilde{M}_k (with corresponding model parameter 474 set $\hat{\mathbf{X}}$) are also depicted. Values of the highest posterior model weights, $P(\hat{M}_k | \mathbf{Y}^*)$, 475 are also included in each subplot. Corresponding depictions for experiment 2 are shown 476 477 in Fig. 5.

These results suggest that: (i) $E(\mathbf{Y} | \mathbf{Y}^*)$ and $E(\mathbf{Y} | \mathbf{Y}^*) \pm \sqrt{V(\mathbf{Y} | \mathbf{Y}^*)}$ are 478 respectively very close to $E(\mathbf{Y} | \mathbf{Y}^*, \tilde{M}_k)$ and $E(\mathbf{Y} | \mathbf{Y}^*, \tilde{M}_k) \pm \sqrt{V(\mathbf{Y} | \mathbf{Y}^*, \tilde{M}_k)}$ 479 when $P(\hat{M}_k | \mathbf{Y}^*) \ge 90\%$ (see Fig. 5); (*ii*) $E(\mathbf{Y} | \mathbf{Y}^*)$ can provide better estimates 480 than $E(\mathbf{Y} | \mathbf{Y}^*, \tilde{M}_k)$ when the individual models of the set considered yield very 481 different results (see Fig. 4); (*iii*) the KIC-based $Var(\mathbf{Y} | \mathbf{Y}^*, \tilde{M}_k)$ is always smaller 482 than its counterparts based on the other IC considered for a given experiment (see Figs. 483 4 and 5). These results are consistent with observations by Ranaee et al. (2016), who 484 noted that model-averaged estimates were virtually coinciding with those associated 485 with the most skillful model in their study when the latter was characterized by 486 $P(\hat{M}_k | \mathbf{Y}^*) \ge 95\%$. They are also consistent with the results of Winter and Nychka 487

(2010), who note that a model average can only be more skillful than the model
identified as best solely when the individual models in the collection produce very
different forecasts.

Figs. 6 and 7 complement the results illustrated above by depicting the Monte Carlo 491 based probability density functions (pdfs) of Cr(VI) concentration in the water runoff 492 evaluated through the candidate models, respectively for experiments 1 and 2. Results 493 are illustrated for early (T_1) , median (T_2) , and late (T_3) simulation times, respectively 494 corresponding to sampling times when observations $Y_{(i)}^*$ (*i* = 1, 5, and 23) are collected. 495 496 One can see that the densities calculated by the L model generally peak at a value closely corresponding to the measured Cr(VI) concentration, an exception being given 497 by the late times results of Fig. 6c, where the *pdfs* associated with the H and L models 498 499 resemble a Delta function. The pdfs associated with F and H models generally show higher skewness and heavier tails, which partially indicate increased probability of 500 extreme values, in comparison with L and T models. 501

502 Variance-based global sensitivity analysis

We provide further insights on the way that model uncertain parameters can contribute to the variability of model responses through a sensitivity analysis. In this context, local sensitivity analysis approaches (*i*) quantify the sensitivity of a model output to small perturbations of parameter values and (*ii*) provide an appraisal of the behavior of the modeled system in regions of the parameter space close to the perturbed value of the parameter (e.g., Razavi and Gupta, 2015 and references therein). GSA techniques enable us to evaluate sensitivities of model outputs across the overall support defining the space of variability of model parameters (e.g., Formaggia et al., 2013; Ciriello et al., 2013, 2015; Riva et al., 2015; Razavi and Gupta, 2015, Dell'Oca et al., 2017 and references therein). The latter is based on the evaluation of the Sobol' indices, which are based on a classical decomposition of variance and provide information on the relative contribution of each uncertain model parameter to the variance of a target model output. Here, we apply the variance-based GSA approach (Saltelli et al., 2008; 2010).

Evaluation of the variance-based sensitivity indices is performed by considering uncertain model parameters as independently random variables, within the space of variability described above. The total sensitivity indices for the i^{th} model parameter (or factor) corresponding to the j^{th} Cr(VI) measurement in the runoff is evaluated as (Saltelli et al., 2008, 2010)

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$$S_{T(ij)} = 1 - \frac{V(E(Y_{(j)} | x_1, x_2, \dots, x_{i-1}, x_{i+1}, \dots, x_{N_p}))}{V(Y_{(j)})}$$

$$= \frac{E(V(Y_{(j)} | x_1, x_2, \dots, x_{i-1}, x_{i+1}, \dots, x_{N_p}))}{V(Y_{(i)})}$$
(25)

Here, $V(Y_{(j)})$ is the variance of model response at the time corresponding to the j^{th} 523 Cr(VI) observation in the runoff; $V(E(Y_{(i)} | x_1, x_2, \dots, x_{i-1}, x_{i+1}, \dots, x_{N_n}))$ represents the 524 variance of model responses expectation conditioned to all factors, excluding factor x_i ; 525 $E(V(Y_{(i)} | x_1, x_2, \dots, x_{i-1}, x_{i+1}, \dots, x_{N_p})) = 1 - V(E(Y_{(i)} | x_1, x_2, \dots, x_{i-1}, x_{i+1}, \dots, x_{N_p}))$ is 526 the expected variance conditioned on all factors, excluding factor x_i . In our study, 527 $E(V(Y_{(j)} | x_1, x_2, \dots, x_{i-1}, x_{i+1}, \dots, x_{N_p}))$ is evaluated through Eq. (19) in Saltelli et al. 528 (2010). We note that Sobol' total sensitivity indices are informative of the relative 529 importance of each model input to the variance of model output and are not amenable 530

to interpretations leading to ranking of the relative interpretive skill of the collection ofmodels considered.

We illustrate our results of GSA for the setting of experiment 2. Figs. 8-11 depict the temporal pattern displayed by $S_{T(ij)}$ as a function of the size, *N*, of the collection of random parameter values for the models tested. These results suggest that N = 5000samples yields sufficiently stable results.

Fig. 12 depicts the temporal variation of the contribution (quantified by 537 $S_{T(ij)} / \sum_{i=1}^{N_p} S_{T(ij)} \times 100 \%$) of each input factor to model output for all candidate models. 538 These results indicate that model output Y is most sensitive to h_{mix} for all alternative 539 models, and are consistent with the observation that the depth of the mixing layer 540 directly determines the mass of Cr(VI) that could be transported from soil to surface 541 runoff. An exception to this pattern is given by model F whose model output is very 542 sensitive to parameter *n*, which is mainly due to the importance of *n* to drive the power 543 law behavior in Eq. (5). Results in Fig. 12 suggest that the sum of the contributions 544 associated with the adsorption parameters consistently increases as simulation time 545 progresses for all models. Considering model L as an example, the contributions of 546 $K_{\rm eq}~$ and $~S_{\rm max}~$ are respectively 0.47% and 1.38% at the beginning simulation, and 547 increase up to 6.61% and 24.02% at the end of simulation. This corresponds to an 548 increase of the combined total contributions of K_{eq} and S_{max} to σ_Y^2 from 1.85% to 549 30.63% during the simulation period and denote the significant effect of the adsorption 550 551 process on Cr(VI) loss from soil to the surface runoff, especially at late times.

Fig. 12 also suggests that the parameters with the lowest (in an average sense)

contribution to σ_Y^2 are K_{eq} (for model L), α (for models F and H), and K_t (for model T). We denote here as $C_{T(K_{eq})} = 5.41\%$, $C_{T(\alpha_{F})} = 2.33\%$, $C_{T(\alpha_{H})} = 10.41\%$, and $C_{T(K_t)} = 0.00\%$, the average of the temporal contributions evaluated for each of these parameters over the set of 23 observation times in experiment 2. It is interesting to note that the ranking $C_{T(K_t)} < C_{T(\alpha_{F})} < C_{T(\alpha_{H})}$ is somehow consistent with the ranking of posterior model weights based on KIC (i.e., 77.87\%, 17.76\%, 4.37\% and 0.00\%, respectively for models T, F, L and H; see Table 6).

As a complement to these results, Fig. 13 depicts the temporal variation of the 560 quantity $\frac{1}{N} \sum_{i=1}^{N_p} \left| S_{T(ij)} - S_{I(ij)} \right|$, representing the mean of absolute difference between the 561 total $(S_{T(ij)})$ and first-order $(S_{I(ij)})$ sensitivity indices. As a remark, one can see that low 562 values of this quantity typically correspond to high KIC-based posterior model weights, 563 with the exception of model T in experiment 1. This finding might be considered as an 564 indication that low values of $\frac{1}{N} \sum_{i=1}^{N_p} |S_{T(ij)} - S_{I(ij)}|$ can be consistent with the low 565 566 expected Fisher information. Additional theoretical developments are needed to fully explore possible implications of these results and will be the subject of future 567 investigations. 568

569 Effect of adsorption on Cr(VI) loss from soil to surface runoff

570 While global sensitivity analyses of the kind we illustrate provide global measures 571 quantifying the contribution of uncertain input parameters to the variance of a model 572 output *pdf* across the entire investigated parameter space, they do not yield a 573 straightforward assessment of (*a*) the actual values attained by model outputs within the parameter space or (*b*) the direction of variation of model results as a function of parameter variation. These features can be readily visualized through a scatterplot analysis. We illustrate the results of the latter by considering the L model, which has been ranked as best in our prior analyses.

Figs. 14 and 15 depict the data clouds associated with the scatterplots of Cr(VI) concentrations in the water runoff resulting from the L model at sampling times respectively corresponding to the collection of the first and last concentration measurement (i.e., Y_1^* and $Y_{N_c}^*$) in experiment 2. Linear regression curves are also depicted for completeness.

These results generally indicate that *Y* is positively correlated to α and negatively correlated to γ . They are consistent with the definition of $Y = \alpha C_w$ and with the observation that a strong incomplete mixing (associated with high values of the mixing coefficient γ) results in decreased values of Cr(VI) in the water runoff (i.e., an increased transfer of Cr(VI) from the mixing layer to the underlying soil). The positive correlation of *Y* with the depth of the mixing layer h_{mix} (Figs. 14e, and 15e) stems from the physical effects of the mixing layer.

The correlation between Y and the adsorption parameter S_{max} changes from negative to positive with elapsing time (compare Figs. 14d and 15d). At the beginning of the simulation period, high values of S_{max} tend to increase the total Cr(VI) mass adsorbed onto the solid phase with a decrease of the mass released to runoff. As time progresses, the dissolved Cr(VI) concentration in the mixing layer decreases, promoting desorption from the solid and subsequent transfer to runoff. These results are also consistent with the tailing observed for the breakthrough curve of Cr(VI) for experiment1.

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CONCLUSIONS

The results from two experiments are used to investigate Cr(VI) losses from soils 600 with diverse grain sizes to water runoff. Four solute transport models (denoted as L, F, 601 602 T and H), coupling a two-layer solute transfer model, respectively, with Langmuir, Freundlich, Temkin, and linear adsorption isotherms, are assessed to simulate Cr(VI) 603 transfer from soil to surface runoff. Each of the analyzed models is calibrated against 604 605 experimental data through Maximum Likelihood (ML) parameter estimation. Four model identification criteria (i.e., AIC, AIC_C, BIC and KIC) are evaluated, and posterior 606 probabilistic weights are then calculated to identify the most skillful model to interpret 607 the available data. The classical variance-based global sensitivity and scatterplot 608 sensitivity analyses are then performed in the context of both experimental settings. Our 609 study leads to the following key conclusions. 610

(1) Cr(VI) concentrations in the runoff are much higher in experiment 2 than in
experiment 1. The most significant difference between the two experimental
scenarios is the soil grain size (i.e., soils were sieved through 2 mm and 1 mm sieves
for experiments 1 and 2, respectively). This result can be explained by two
mechanisms: (*a*) larger grain sizes would increase infiltration rates; and (*b*) a mixing
layer depth tends to be decreased in a soil with larger grain size, resulting in a

617 decreased Cr(VI) loss through runoff.

(2) For experiment 1, the largest AIC, AIC_C and BIC-based posterior model weights
are associated with model H, respectively with values of 47.09%, 66.54% and
61.60%. Model L is ranked highest by KIC, with a posterior weight of 53.97%. For
experiment 2, the highest AIC, AIC_C and BIC-based posterior model weights are
90.39%, 90.30% and 90.34%, respectively, and are linked to model L; Model T is
favored by KIC, with a posterior weight equal to 77.87%. Model L is then evaluated
as the most skillful for the overall interpretation of both experimental results.

625 (3) Variance-based global sensitivity results suggest that the thickness of the mixing layer, h_{mix} , is the most sensitive parameter for all models, an exception being the 626 Freundlich model, where *n* is the uncertain parameter with the highest contribution 627 628 to the model output variance. The total sensitivity of the adsorption parameters tends to increase with simulation time. For example, in the case of model L the sum 629 of the K_{eq} and S_{max} contributions (see Fig. 12b) to model output variance 630 increases from 1.85% to 30.63% across the temporal window spanned in 631 experiment 2. This result suggests that adsorption has a significant effect on the 632 uncertainty of prediction for Cr(VI) loss from soil to runoff. 633

(4) The scatterplot analysis results from model L suggest that the incomplete mixing coefficient (α) and the depth of mixing layer (h_{mix}) are positively correlated with Cr(VI) concentration in the runoff, and the incomplete mixing coefficient (γ) and Cr(VI) concentration are negatively correlated. The correlation between Cr(VI) concentration in the runoff and the maximum adsorption capacity (S_{max}) changes from negative to positive during the simulation process. These results are obtained from the finding that (*a*) large values of S_{max} tend to increase the total Cr(VI) mass adsorbed onto the solid phase at the beginning of the simulation period; and (*b*) the dissolved Cr(VI) concentration in the mixing layer decreases as time progresses, promoting Cr(VI) transfer from soil to runoff.

It is remarked that the mathematical formulation for the adsorption process plays a 644 significant role on prediction of solute loss from soil to surface water runoff, and the 645 equilibrium adsorption assumption is not accurate in some field scenarios. Non-646 647 equilibrium adsorption models should be further explored in future studies when they are used in the two-layer model to describe chemical transport from soil to water runoff. 648 Additional elements of future study should also include the use of global sensitivity 649 650 techniques that allow exploring the relative importance of each uncertain model parameter through the evaluation of (statistical) Moment-based Metrics of the kind 651 proposed by Dell'Oca et al. (2017). The latter are not confined to a description of the 652 653 feedback between uncertain model inputs and outputs via variance-based metrics (such as the Sobol' indices) and aim at providing a comprehensive picture, quantifying the 654 impact of model parameter uncertainties on the statistical moments driving the main 655 features of the probability density function of model outputs. 656

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Tables

2

3 Table 1. Main chemical characteristics of soil sample 2 used in the experiments.

Cr(VI)	Fe ₂ O ₃	Al ₂ O ₃	MnO	Eh	pH
$\leq 0.015 \text{ (mg kg}^{-1}\text{)}$	0.0377 (g kg ⁻¹)	0.0094 (g kg ⁻¹)	0.0051 (g kg ⁻¹)	497.31 (mV)	7.6

4

5 Table 2. Key parameters and quantities characterizing the experimental conditions.

Experiment	Rainfall intensity, <i>p</i> /(cm min ⁻¹)	Stable infiltration rate, $i_s /(\text{cm min}^{-1})$	The time start to produce ponding water, t_p /(min)	The time runoff occur, <i>t_r</i> /(min)	The time of infiltration rate reached stable, <i>ts</i> /(min)
1	0.092	0.028	6.5	27	40
2	0.100	0.024	5.75	28.7	39

mod	lel	L					F				
param	eter	α	γ	K_{eq}	Smax	hmix	α	γ	K_{f}	п	hmix
Experiment	estimate	0.28	0.78	0.02	69.00	0.37	0.40	0.69	6.61	1.31	0.13
1	SD	34.34	12.89	0.44	2753.87	5.61	162.21	96.57	1162.71	5.19	2.84
Experiment	estimate	0.15	0.21	0.51	111.00	0.46	0.22	1.0	17.50	2.89	0.88
2	SD	10.87	2.16	12.81	1722.21	11.74	76.13	18.52	590.91	50.92	28.85
mod	el			Т					Н		
mod	lel eter	α	γ	T K	f	hmix	α	γ	H Kd	hmis	ι
mod param Experiment	lel eter estimate	α 0.19	γ 0.70	T <i>K</i> 1.13	f 291.64	h _{mix} 0.35	α 0.13	γ 0.66	H <i>K</i> _d 0.45	hmi: 0.38	3
mod param Experiment 1	lel eter estimate SD	α 0.19 3749	γ 0.70 988.1	T <i>K</i> 1.13 295223.64	<i>f</i> 291.64 1542530.39	h _{mix} 0.35 1863.25	α 0.13 359.57	γ 0.66 69.39	H <i>K</i> _d 0.45 424.10	h _{mi} 0.33	α 3 1
mod param Experiment 1 Experiment	lel eter estimate SD estimate	α 0.19 3749 0.28	γ 0.70 988.1 0.92	T <i>K</i> 1.13 295223.64 0.004	<i>f</i> 291.64 1542530.39 124.41	h _{mix} 0.35 1863.25 0.67	α 0.13 359.57 0.035	γ 0.66 69.39 0.52	H <i>K</i> _d 0.45 424.10 0.82	h _{mi} 0.33 1.8 0.70	

8 Table 3. Maximum Likelihood model parameter estimates and associated standard deviation (SD).

	L model							F model			
	α	γ	K_{eq}	$S_{\rm max}$	h_{mix}		α	γ	K	п	h_{mix}
α	1	0.99	0.61	0.77	0.65	α	1	1	0.99	-0.99	0.98
γ	0.99	1	0.62	0.76	0.64	γ	0.99	1	0.99	-0.99	0.97
K _{eq}	-0.99	-0.99	1	-0.02	-0.19	K	-0.99	-0.98	1	-0.99	0.97
S _{max}	0.98	0.98	-0.97	1	0.98	п	-0.95	-0.97	0.95	1	-0.95
h _{mix}	0.99	0.99	-0.99	0.99	1	h_{mix}	0.97	0.95	-0.97	-0.87	1

11 Table 4. Correlation coefficients between model parameters

T model

H model

	α	γ	K_{t}	f	h_{mix}		α	γ	K_d	h_{mix}
α	1	1	0.99	1	1	α	1	1	1	-0.97
γ	0.99	1	0.99	1	1	γ	0.99	1	1	-0.97
K_t	0.91	0.94	1	1	1	K_{d}	0.99	1	1	-0.97
f	0.98	0.99	0.96	1	1	h_{mix}	-0.43	-0.44	-0.44	1
h _{mix}	0.99	0.99	0.95	0.99	1					

Note: the upper and lower triangular regions respectively represent correlation coefficients for experiment 1and 2.

Model		Ι			F				
Index	AIC	AICC	BIC	KIC	AIC	AICC	BIC	KIC	
Experiment 1	-139.29	-134.34	-132.22	-156.22	-139.94	-135.00	-132.87	-155.78	
Experiment 2	-36.04	-30.79	-29.23	-56.05	-30.38	-25.13	-23.57	-58.85	
Model]	Г		Н				
Index	AIC	AICC	BIC	KIC	AIC	AICC	BIC	KIC	
Index Experiment 1	AIC -80.22	AICC -75.27	BIC -73.15	KIC -135.52	AIC -140.79	AICC -137.46	BIC -134.90	KIC -150.26	

15 Table 5. Results of model identification criteria for both experiments.

18 Table 6. Posterior model weights (%) for the set of alterative models tested.

Model		1	_		F				
Index	AIC	AICc	BIC	KIC	AIC	AICc	BIC	KIC	
PMW 1	22.18	14.03	16.10	53.97	30.7	19.43	22.30	43.29	
PMW 2	90.39	90.30	90.34	4.37	5.32	5.32	5.32	17.76	
Model		r	Γ			ł	ł		
Model Index	AIC	AICc	Г BIC	KIC	AIC	H AIC _C	H BIC	KIC	
Model Index PMW 1	AIC 0.00	AIC _C 0.00	Г ВІС 0.00	KIC 0.00	AIC 47.09	H AIC _C 66.54	H BIC 61.60	KIC 2.74	

19 Note: PMW 1 (or 2) denotes posterior model weights for experiment 1 (or 2)

Figures 1 entrance of rain water to simulator rain simulator ponding water 8[#] hypodermic needles 30 cm fix-flux pump to control rain intensity 40 cm rain water bucket (distilled water) runoff sample soil gravel drainage holes 2 3 Figure 1. Sketch of the experimental set-up 4 р αC_w the runoff layer the entire soil surface mixing zone C_w h_{mix} the mixing layer γC_{u} the underlying soil 5

Figure 2. Conceptual depiction of the two-layer model. Notations are: the rainfall intensity p; the specfic discharege rate of overland flow q; the depth of the ponding layer h_p ; the depth of the mixing layer h_{mix} ; infiltration rate i; the solute concentration in the mixing layer C_w ; the solute concentration in the ponding layer /the runoff layer αC_w ; the solute concentration in the underlying soil γC_w .



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Figure 3. Temporal variation of measured Cr(VI) concentrations in the runoff together with their simulated counterparts based on optimized parameters (L, F, T and H models) for experiments (a) 1 and (b) 2.



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Figure 4. Scatterplots of estimates of Cr(VI) concentration in the surface runoff (collected in vector **Y**) versus data (collected in vector **Y**^{*}) of experiment 1. When

19 used, posterior model weights are evaluated through model identification criterion (a)

20 AIC, (b) AIC_c, (c) BIC, and (d) KIC; \tilde{M}_k represents the most skillful model, as



21 identified by the corresponding criterion.

Figure 5. Scatterplots of estimates of Cr(VI) concentration in the surface runoff (collected in vector **Y**) versus data (collected in vector **Y**^{*}) of experiment 2. When used, posterior model weights are evaluated through model identification criterion (a) AIC, (b) AIC_C, (c) BIC, and (d) KIC; \tilde{M}_k represents the most skillful model, as identified by the corresponding criterion.

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- 29



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Figure 6. Sample probability density functions (*pdfs*) of Cr(IV) in runoff water (*Y*) based on the Monte Carlo simulations performed for each model at (a) early (T_1), (b) median (T_2), and (c) late (T_3) simulation times, respectively corresponding to sampling times where observations $Y_{(i)}^*$ (*i* = 1, 5, and 23), are collected in experiment 1. Vertical lines correspond to measured values $Y_{(i)}^*$.



Figure 7. Sample probability density functions (*pdfs*) of Cr(IV) in runoff water (*Y*) based on the Monte Carlo simulations performed for each model at (a) early (T_1), (b) median (T_2), and (c) late (T_3) simulation times, respectively corresponding to sampling times where observations $Y_{(i)}^*$ (*i* = 1, 5, and 23), are collected in experiment 2. Vertical lines correspond to measured values $Y_{(i)}^*$.





- 48 i.e., N = (a) 500, (b) 1000, (c) 2000, (d) 5000, (e) 10,000, and (f) 100,000.
- 49





51 Figure 9. Total sensitivity indices of F model versus time and for diverse sample size,

52 i.e., N = (a) 500, (b) 1000, (c) 2000, (d) 5000, (e) 10,000, and (f) 100,000.



54 Figure 10. Total sensitivity indices of T model versus time and for diverse sample size,

55 i.e., N = (a) 500, (b) 1000, (c) 2000, (d) 5000, (e) 10,000, and (f) 100,000.









Figure 12. Contributions of each input factor to variation of Cr(VI) in surface runoff
versus time for experiment 2 according to model (a) L, (b) F, (c) T, and (d) H.

63



Figure 13. Temporal variation of the mean of the absolute difference between the total $(S_{T(ij)})$ and first-order $(S_{I(ij)})$ sensitivity indices for experiment (a) 1 and (b) 2; the order of model ranking according to the KIC-based posterior model weights are indicated.



Figure 14. Scatterplots depicting the dependence of model output *Y* and parameters (a) 74 α , (b) γ , (c) K_{eq} , (d) S_{max} , and (e) h_{mix} of the Langmuir coupled tow-layer (L) 75 for experiment 2 at observation time corresponding to the first sampling time. Linear 76 regression curves (solid lines) are included.



Figure 15. Scatterplots depicting the dependence of model output Y and parameters (a) α , (b) γ , (c) K_{eq} , (d) S_{max} , and (e) h_{mix} of the Langmuir coupled tow-layer (L) for experiment 2 at observation time corresponding to the last sampling time. Linear regression curves (solid lines) are included.